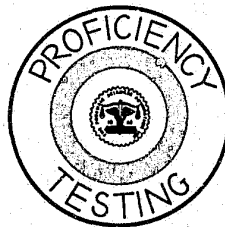


LABORATORY PROFICIENCY TESTING PROGRAM



47525



THE FORENSIC SCIENCES FOUNDATION, INC.

11400 ROCKVILLE PIKE

ROCKVILLE, MARYLAND 20852

(301) 770-2723

LABORATORY PROFICIENCY TESTING PROGRAM

REPORT NO.4

GLASS EXAMINATION

PROJECT ADVISORY COMMITTEE

J.F. Anderson
Spokane, Washington

J.D. Chastain
Austin, Texas

Richard H. Fox
Independence, Missouri

Anthony Longhetti
San Bernardino, Ca.

Charles McInerney
Pittsburgh, Pa.

Andrew H. Principe
Highland Park, Illinois

John Thornton
Berkeley, Ca.

B. Edward Whittaker
Miami, Florida

PROJECT STAFF

K. S. Field

E. Fabricant

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Points of view or opinions stated in this document are those of the
authors and do not necessarily represent the official position or
policies of the U.S. Department of Justice.

FOREWORD

The analysis summarized in this report is the fourth of a series that will be made in conjunction with this proficiency testing research project.

In the course of this testing program participating laboratories will have analyzed and identified ten different samples of physical evidence similar in nature to the types of evidence normally submitted to them for analysis.

The results of Test Number Four are reflected in the charts and graphs which follow.

The citing of any product or method in this report is done solely for reporting purposes and does not constitute an endorsement by the project sponsors.

Comments or suggestions relating to any portion of this report or of the program in general will be appreciated.

July 1975

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BACKGROUND

This laboratory proficiency testing research project, one phase of which is summarized in this report, was initiated in the fall of 1974.

This is a research study of how to prepare and distribute specific samples; how to analyze laboratory results; and how to report those results in a meaningful manner. The research will be conducted in two cycles, each of which will include five samples: a controlled substance; firearms evidence; blood; glass, and paint.

Participation in the program is voluntary. Accordingly, invitations have been extended to 235 laboratories to share in the research. It is recognized that all laboratories do not perform analyses of all possible types of physical evidence. Thus, in the data summaries included in this report, space opposite some Code Numbers (representing specific laboratories) may be blank, or marked "No Data Returned."

A final project report will be prepared at the conclusion of Cycle II.

The Project is under the direct control of the Project Advisory Committee whose members' names are listed on the Title Page. Each is a nationally known criminalistic laboratory authority.

Supporting the Project Advisory Committee in their efforts is the Forensic Sciences Foundation with additional support from the National Bureau of Standards in the areas of sample evaluation and data analysis and interpretation.

SUMMARY

Test Sample #4 consisted of glass samples A and B, packaged in a plastic box. The samples were mailed on April 28, 1975 with instructions to handle the sample in a manner similar to like evidence and submitted for analysis.

Test Sample #4 was sent to 234 laboratories. Three of those laboratories served as referees, reducing the actual number to 231.

In the accompanying data summaries, 124 laboratories responded with completed data sheets, 49 laboratories responded that they did not do glass analysis and no response was received from 61 laboratories. This represents a participation rate of 68%.

No effort was made in this report to highlight areas wherein laboratory improvements might be instigated.

ANNEX A

FIGURE 1



LAB CODE A- _____

☐ CHECK HERE (AND RETURN) IF YOU DO NOT PERFORM GLASS EXAMINATION

DATE RECEIVED IN LAB _____

DATE PROCESSED IN LAB _____

DATA SHEET

PROFICIENCY TESTING PROGRAM

TEST #4 GLASS EXAMINATION

Item A represents a glass sample taken from the scene of a burglary. Item B represents a glass sample taken from the trousers of a suspect.

1. Item A could have common origin with Item B.

☐ YES

☐ NO

☐ Inconclusive

2. What information (quantitative and qualitative) did you develop to arrive at your conclusion in No. 1?

Item A

Item B

- 2 -

3. Method(s) and instrument(s) used:

DATA SHEETS MUST BE RECEIVED AT THE FOUNDATION OFFICE BY MAY 30, 1975.

ANNEX B

National Bureau of Standards Analysis

LABORATORY TESTING PROGRAM

Test No. 4 - Glass Examination

In Test No. 4, 234 laboratories were each sent two pieces of glass referred to as A and B. Participants were asked three questions: (1) Could item A have a common origin with item B? (2) What information was developed to answer question 1? (3) What methods and instruments were used?

Of the 234 laboratories, 124 responded with data, 49 indicated they do not do glass analysis, and 61 did not respond. A tabulation of the codes for laboratories in each of these last two categories is given in Table 1.

Both the supplier of the glass samples and the referee laboratories indicate, in Tables 2 and 3, that A and B do not have a common origin. 118 participating laboratories reported A and B could not have a common origin, 4 were inconclusive. These results are tabulated in Table 4 for each of the 124 responding laboratories.

The supplier notes in Table 2 that the thickness of glass B is more uniform than the thickness of glass A. It may be noted from Tables 2 and 3 that the refractive indexes and densities of the glass B samples are also more uniform.

A quantitative analysis was made of the two most frequently reported methods (see below) for the laboratories which reported data suitable for this purpose. Table 5 shows refractive index and density differences between A and B. The density measurements produced greater variations than did the refractive index measurements. The average refractive index difference for 35 participating laboratories (those that reported to at least four decimal places) agrees well with the refractive index differences reported by the referee laboratories and the sample supplier. However, the density differences in Table 5 do not agree as well. Of the 12 laboratories reporting density data that can be interpreted quantitatively, two reported that A had a greater density than B, while the reverse was reported by the other 10.

On the average, 2.7 methods per laboratory were reported for question 3. The relative frequencies of the methods reported are shown in Table 6. Tables 7 and 8 show the responses to questions 2 and 3 from each laboratory. Table 9 was especially requested by the Project Advisory Committee.

This annex was prepared by the Law Enforcement Standards Laboratory (LESL) of NBS, in conjunction with the NBS Laboratory Evaluation Technology Section (LETS). The anonymous test results reported by the participating forensic laboratories were analyzed and tabulated by Jeffrey Horlick and Charles Leete of LETS, and Robert Mills of LESL. This work was supported by the National Institute of Law Enforcement and Criminal Justice, Department of Justice.

Table 1

THE FOLLOWING LABS INDICATED THEY DO NOT DO GLASS ANALYSIS:

706	775	845	953
711	785	860	983
720	788	862	992
721	791	875	998
735	793	877	
741	803	886	
743	807	913	
744	810	918	
749	812	920	
753	824	924	
755	826	927	
758	828	932	
759	830	935	
761	841	950	
767	844	951	

Total Labs = 49

THE FOLLOWING LABS DID NOT RESPOND:

703	738	825	879	942
707	764	834	887	946
708	770	836	891	964
710	772	850	898	966
715	774	858	900	972
723	780	859	902	973
724	781	861	903	984
728	782	864	904	985
732	795	865	905	988
733	811	867	912	
734	814	869	914	
736	816	870	917	
737	817	871	937	

Total Labs = 61

Table 2

SUPPLIER'S CHARACTERIZATION OF SAMPLESThickness

The thickness of glass B is much more uniform than the thickness of glass A.

Color

Both are untinted glasses.

Fluorescence

Glass B has tin dissolved in one of its surfaces, and thus will fluoresce when exposed to UV light. Glass A does not contain tin.

Refractive Index

Measurements made on six samples gave the following results for refractive index, N_D (Sodium Line):

	<u>A</u>	<u>B</u>
	1.5167	1.5186
	1.5167	1.5185
	1.5158	1.5186
	1.5167	1.5185
	1.5168	1.5186
	<u>1.5166</u>	<u>1.5186</u>
Average:	1.51655	1.5186
Standard Deviation:	0.0004	0.00005

Density

Density measurement results for six samples are as follows:

	<u>A</u>	<u>B</u>
	2.4860 g/cm ³	2.4945 g/cm ³
	2.4862	2.4947
	2.4821	2.4949
	2.4876	2.4949
	2.4859	2.4944
	<u>2.4852</u>	<u>2.4952</u>
Average:	2.4855 g/cm ³	2.4948 g/cm ³
Standard Deviation:	0.002	0.0003

Composition

	<u>A</u>	<u>B</u>
SiO ₂	73.37%	73.20%
Na ₂ O	13.16	13.64
CaO	8.26	8.87
MgO	3.61	3.95
Al ₂ O ₃	1.22	0.15
K ₂ O	0.24	0.03
SO ₃	0.18	0.25
Fe ₂ O ₃	<u>0.11</u>	<u>0.08</u>
Total	100.15%	100.17%

Table 3

RESULTS FROM THREE REFEREE LABORATORIES

Lab 1. (NBS was requested to do only refractive index and density measurements)

Refractive Index (3 specimens)

	<u>A</u>	<u>B</u>
1	1.51552	1.51845
2	1.51648	1.51844
3	1.51550	1.51844
Average:	1.51583	1.51844
Standard Deviation:	0.0006	0.000006

Refractive index of each specimen was determined at $\lambda = 0.5893 \mu\text{m}$, the mean of the D lines of sodium at 20°C . These determinations are considered to be accurate to within 2×10^{-5} .

Density

	<u>A</u>	<u>B</u>
1	2.4740 g/cm ³	2.4919 g/cm ³
2	2.4802	2.4916
3	2.4739	2.4919
Average:	2.4761 g/cm ³	2.49185 g/cm ³
Standard Deviation:	0.004	0.0002

Densities were determined using the ASTM C-729 method, which is a sink-float comparator method using a combination solution of iso-propyl salicylate and S-tetra bromethane. The determination of the difference in density between the two glasses is considered to be accurate to 1×10^{-4} , although the absolute accuracy of any one determination may be no greater than 1×10^{-3} .

Lab 2.Refractive Index (N_D Sodium Line)

<u>A</u>	<u>B</u>
1.516	1.518

Refractive index measurements by Becke line method; samples A and B showed a clear difference when measured comparatively.

Table 3 continued

Lab 2 continued:

Density

<u>A</u>	<u>B</u>
2.479 g/cm ³	2.485 g/cm ³

Density determined by pycnometer measurement of liquid mixture in which glass samples floated; samples A and B showed unequivocal difference in density when measured comparatively.

Lab 3.

Fluorescence*

Sample A No fluorescence in U.V. (short wave)

Sample B Fluorescence on one side in U.V. (short wave)

*Would normally stop at this point since A is different from B.

Refractive Index

Sample A	N _C - 1.5129	N _D - 1.5157	N _F - 1.5216
	(Dispersion curve different shape from B)		
Sample B	N _C - 1.5158	N _D - 1.5185	N _F - 1.5247

Refractive index and dispersion curve was done using AOAC method with hot stage, monochromator, and Phase-Star Microscope.

Density

<u>A</u>	<u>B</u>
2.4911 g/cm ³	2.5054 g/cm ³

Density by sink-float method using bromoform-alcohol mixture at 25° ± 0.1°C with plumb bob and density balance.

Table 4

RESPONSES TO QUESTION 1: Item A could have common origin with Item B?

Lab Code	Response	Lab Code	Response	Lab Code	Response
705	NO	805	NO	915	NO
709	NO	806	NO	921	NO
712	NO	809	NO	923	NO
713	NO	813	NO	925	NO
717	NO	815	NO	926	NO
718	NO	818	NO	931	NO
719	NO	820	NO	938	NO
722	NO	821	NO	944	NO
726	NO	822	YES	948	NO
727	NO	823	NO	958	NO
729	NO	827	NO	960	NO
730	NO	829	NO	961	NO
731	NO	831	NO	962	NO
739	NO	832	NO	969	NO
740	Inconclusive	833	NO	970	NO
742	NO	835	NO	974	NO
745	NO	837	NO	975	NO
746	NO	838	NO	978	NO
747	NO	839	NO	979	NO
748	NO	842	NO	980	NO
750	NO	843	NO	986	NO
751	NO	847	NO	987	NO
752	NO	848	NO	989	NO
754	NO	849	NO	994	NO
756	NO	852	NO	995	NO
757	NO	853	NO	999	YES
760	NO	854	NO		
762	NO	855	NO		
763	NO	856	NO		
765	NO	863	NO		
766	YES	866	NO		
768	NO	868	NO		
769	NO	872	NO		
773	NO	873	NO		
777	NO	874	NO		
778	NO	876	NO		
779	NO	880	NO		
783	NO	883	NO		
784	YES	884	NO		
786	NO	885	NO		
787	Inconclusive	888	NO		
789	NO	889	NO		
790	NO	892	NO		
792	NO	894	NO		
794	NO	895	NO		
796	NO	897	NO		
797	NO	899	NO		
798	NO	907	NO		
799	NO	908	NO		

Total labs
Responding: 124

NO = 118

YES = 4

Inconclusive = 2

Table 5

Refractive Index and Density Differences:
B minus A

	<u>Differences in Refractive Index</u>	<u>Differences in Density - g/cm³</u>
Laboratory 1 (NBS) - Ave. of 3 pieces	0.00261	0.01575
Laboratory 2 -	0.002	0.006
Laboratory 3 - RI measured at 3 λ 's	0.0029/0.0028/0.0031	0.01430
Sampler Supplier - Ave. of 6 pieces	0.00205	0.00930
Average of Results from 35 Labs	0.00254	-----
Standard Deviation of these 35 results	0.0007	

Table 6

Relative Frequencies of the Reported Methods

Refractive Index	90
Density	77
Thickness	50
U.V. Light	42
Elemental Analysis	18
Dispersion Curves	14
Color	9
Dispersion Staining	8
X-Ray Fluorescence	8
Physical Edge Match	4

Table 7

RESPONSES TO QUESTION 2: Information Developed.

	Sample A	Sample B																
705	Elemental composition of "A" differs from that of sample "B". Refractive Index of "A" differs from sample "B". Dispersion of light from sample "A" differs from sample "B".																	
709	Item A fluoresces pink.	Item B fluoresces white.																
712	No fluorescence on either side. Thickness is .222 - .223 inches. Refractive index is $n_D^{25} = 1.5156 \pm .0004$	Fluoresces on one side - float glass. Thickness is .223 - .224 inches. Refractive index is $n_D^{25} = 1.5185 \pm .0004$																
713	Thickness is 5.67mm. Refractive index of A less than 1.516.	Thickness is 5.70mm. Refractive index of B greater than 1.516.																
717	Thickness is .2282 inches. Relative Density less than B. Refractive index is approx. 1.5186. Spectrograph indicated the presense of tin in B and none detected in A.	Thickness is .2252 inches. Relative Density greater than A. Refractive index is approx. 1.5206.																
718	Colorless under ultraviolet light. Thickness is .226 inches. B is approx. .005 grams/cc more dense than A. Refractive index of B is approx. .002 units greater than A.	Yellow fluorescence under UV light. Thickness is .225 inches.																
719	A has different optical and physical qualities and different chemical contents from B.																	
722	Visual - qualitative; Refractive index - qualitative; Density gradient-qualitative.																	
726	Refractive index is 1.516. Density is less than B. Fluorescence is non-detectable.	Refractive index is 1.518. Density is greater than A. Fluorescence - light blue on one side.																
727	Becke line readings were: 656: 66.5 - 71.5 Average 589: 73.3 - 75.4 Average 486: 86.1 - 88.7 Average Glass chips were at different levels in the sink float method.	Becke line readings were: 656: 61.6 - 67.0 Average 589: 67.5 - 70.7 Average 486: 80.6 - 83.9 Average																
729	There is difference in fluorescence between the two samples.																	
730	Optical density @ 500nm = .160 U.V. cutout = 318 nm. Emission Spec shows no tin. Comparison sample to B shows lower Mg, Al, Ca, Zn, higher Cu.	Optical density @ 500nm = .158 U.V. cutout = 325 nm. Emission Spec shows tin.																
731	Density gradient shows .0008 gm/cc difference between A and B.																	
739	Refractive index of A (approx. 1.514) less than B (approx. 1.516). Density of A less than B on comparison basis.																	
740	Refractive index = 1.5160 Elemental - Ca, Fe, Zn, SrZn Count ratio - not concentration <table><tr><td>Ca/Fe</td><td>Ca/Zn</td><td>Ca/Sr</td><td>Ca/Zn</td></tr><tr><td>4.7</td><td>53.0</td><td>6.5</td><td>25.0</td></tr></table>	Ca/Fe	Ca/Zn	Ca/Sr	Ca/Zn	4.7	53.0	6.5	25.0	Refractive index = 1.5191 Elemental - Ca, Fe, Zn, SrZn Count ratio - not concentration <table><tr><td>Ca/Fe</td><td>Ca/Zn</td><td>Ca/Sr</td><td>Ca/Zn</td></tr><tr><td>6.26</td><td>53.0</td><td>4.6</td><td>13.5</td></tr></table>	Ca/Fe	Ca/Zn	Ca/Sr	Ca/Zn	6.26	53.0	4.6	13.5
Ca/Fe	Ca/Zn	Ca/Sr	Ca/Zn															
4.7	53.0	6.5	25.0															
Ca/Fe	Ca/Zn	Ca/Sr	Ca/Zn															
6.26	53.0	4.6	13.5															

Sample ASample B

- 742 Thickness is .222 of an inch.
No fluorescence to UV light.
Refractive index and dispersion:
 $N_D^{25} = 1.515028$, $N_F^{25} = 1.520215$, $N_C^{25} = 1.512824$
 $V_x = 69.68$
- 745 A is less dense than B.
A has lower (estimated .002-003)
refractive index than B.
Refractive index is approx. 1.517
Thickness is .225 inch.
- 746 Refractive index is 1.516 (comparative). Refractive index is 1.518 (comparative).
A is easily separated from B by flotation; A is less dense than B.
- 747 Thickness is .223 inches.
Refractive index is 1.515665
Dispersion is 58.29
- 748 Color: Blue-green tint.
Fluorescence: None.
Thickness: .2213 inch to .2216 inch.
Density: 2.484 g/ml
- 750 No fluorescence to UV light.
A is less dense than B and a lower refractive index than B on a comparison basis.
- 751 Thickness is .224 inch.
Short wave UV: Dark purple absorbance.
Refractive index (direct): 1.5163 - 1.5164
Refractive index and dispersion (Emmons Double Variation Method) gives:
 $N_D^{25} = 1.5165$
Both A and are isotropic, conchoidal fractures - apparent glass.
- 752 Thickness is 6.19 mm.
A is less dense than B and has a lower refractive index than B.
- 754 Glass is slightly darker green in color. Glass is slightly lighter green in color.
No fluorescence under UV light. One surface fluoresces under UV light.
A is less dense than B and refractive index does not match B.
- 756 Thickness = .568 cm.
 $N_D = 1.5162$, $N_C = 1.5137$, $N_F = 1.5226$
Dispersion curve for A tapers in 486-500 nm region while B is straight line.
- 757 No fluorescence under short wave UV light. One side fluoresces under UV light.
- 760 Orange fluorescent color under UV light. Yellow fluorescent color under UV light.
Thickness is 5.61 mm (average). Thickness is 5.71 mm (average).
A is less dense than B and has a lower refractive index than B (comparative basis).

Sample ASample B

- 762 Thickness - 5.665 mm.
A has lower density than B and lower refractive index than B at $25^{\circ}\text{C} \pm .1^{\circ}\text{C}$.
- 763 Refractive index is 1.512 - 1.516.
Relative density of A is less than B.
- 765 Thickness is 5.69 mm.
Refractive index @ $26\frac{1}{2}^{\circ}\text{C}$ = 1.5153.
Density @ 25°C = 2.4621 and less than B.
- 766 Refractive index of both A and B is 1.560.
Density Gradient - A floats in tube at same level as B.
- 768 A does not fluoresce under UV light.
- 769 Thickness = .2237 in.
Density = 2.4856 @ 27°C
Spec. Grav. = 2.4945 @ 27°C
Refractive index = 1.5157 @ 20.5°C
Normal color of both A and B is greenish-blue.
Ultraviolet fluorescence of A is gold and B is bright yellow (@ 254 nm)
- 773 Neither side fluoresces under short wave UV light.
- 777 Color of A and B are similar (Blue-green).
Thickness = .224 inches.
 $n_D = 1.515$
Dispersion of A and B are similar.
Density similar to B (about $2.48 \pm .02$)
- 778 A and B do not have same refractive index (comparative basis).
A contains K.
- 779 Refractive index and density on both A and B.
- 783 Thickness: .227 - .228 inch.
Refractive index: 1.5161 - 1.5163
- 784 Thickness: 5.77 mm.
Refractive index on both A and B greater than 1.516 and less than 1.518.
Elemental composition (both A and B) - Si, Ca, Fe, Sr, and Zr.
Density of A identical to B.
- 786 UV fluorescence - none.
Column density determination.
Tin not detected.
- 787 Match by "jigsaw" method gave negative results.
Gradient density method was used with inconclusive results.
- 789 Density and refractive index of A less than B.
- 790 Thickness: .2218 inch average.
A is less dense than B.
Temperature match to Silicone oil:
 $n_C - 61.1^{\circ}$, $n_D - 67^{\circ}$, $n_F - 79.8^{\circ}$
- Thickness - 5.710 mm.
Refractive index is 1.516 - 1.520.
Thickness is 5.72 mm.
Refractive index @ $26\frac{1}{2}^{\circ}\text{C}$ = 1.5158
B fluoresces yellow on one side with UV light.
Thickness = .2249 in.
Density = 2.4979 @ 27°C
Spec. Grav. = 2.5068 @ 27°C
Refractive index = 1.5185 @ 20.5°C
One side is fluorescent under short wave UV light.
Thickness = .225 inches.
 $n_D = 1.518$
Density similar to A (about $2.51 \pm .02$)
B has 2 times Zr than A.
- Thickness: .224 - .225 inch.
Refractive index: 1.5180 - 1.5182
Thickness: 5.71 mm.
Thickness: .2248 inch average.
 $n_C - 66^{\circ}-67.5^{\circ}$, $n_D - 75^{\circ}$, $n_F - 85.6^{\circ}-87^{\circ}$

Sample ASample B

- 792 Refractive index of B is higher than A at three wavelengths (488, 586.6, 656.3).
- 794 No fluorescence under short or long wave UV light.
Thickness: 5.23 mm
Refractive index: 1.5162
- 796 Specific gravity = 2.4884
Thickness = .225 inch.
- 797 Refractive index = 1.5174
(Emmons Double Variation Method)
Refractive index = 1.5165(direct)
- 798 UV light (short and long wave) was used on both A and B.
Refractive index and Gradient density were determined on both A and B.
- 799 Both surfaces of A absorb short wave UV.
Refractive index - both surfaces 1.51560
Density of A is 2.48
- 805 Thickness: .223 inches
Does not contain tin.
A is less dense than B.
Dispersion curve of A differs from B primarily in the yellow-red region.
- 806 A and B have same thickness - 5.7 mm
A and B have same edge color.
A and B differ in specific gravity - Sink Float method.
Refractive index report as temp. of Silicon oil:
Avg. (going up) 74.2°C
Avg. (down) 73.7°C
- 809 Visual comparison of color with B.
Thickness measurements.
Examination under UV light.
Density gradient comparison with B.
Emission spectrographic analysis.
- 813 Refractive index and density of a less than B.
Dispersion staining color in 1.516 oil was blue (ambient temp.).
X-ray fluorescence elemental ratios:
 $\frac{\text{Calcium}}{\text{Iron}} = 3.74$ $\frac{\text{Calcium}}{\text{Strontium}} = 36.85$
 $\frac{\text{Calcium}}{\text{Zirconium}} = 15.76$
- 815 Thickness: .225 inch
Refractive index: 1.516(not absolute)
Color of both A and B is light green.
- 818 B is significantly denser than A.
- Fluorescence on one side under short wave UV light. None under long wave.
Thickness: 5.21 mm
Refractive index: 1.5189
- Specific gravity = 2.4994
Thickness = .224 inch.
- Refractive index = 1.5193
Refractive index = 1.5184(direct)
- One surface of B fluoresces yellow/green with short wave UV.
Refractive index - side which fluoresces is 1.5189, other side is 1.5172
Density of B is 2.50
- Thickness: .224 inches
Tin present.
- Avg. (going up) 68.8°C
Avg. (down) 68.4°C
- Dispersion staining color in 1.516 oil was blue violet & red orange.
 $\frac{\text{Calcium}}{\text{Iron}} = 4.82$ $\frac{\text{Calcium}}{\text{Strontium}} = 18.18$
 $\frac{\text{Calcium}}{\text{Zirconium}} = 9.11$
- Thickness: .225 inch
Refractive index: 1.519(not absolute)

Sample ASample B

- 820 Does not fluoresce under short wave UV light. $N_D^{25} - 1.5166$
- 821 Thickness of both A and B is .223 inches.
Color of both A and B is clear.
Density: 2.480
Refractive index: 589nm 1.519
488nm 1.524
- 822 Thickness and optical qualities of A and B are similar.
- 823 Visual examination showed both A and B had a slightly green tint.
Polarized microscopic examination indicated that both A and B were possibly glass.
Refractive index analysis using Becke line technique indicated difference between A and B.
- 827 Density: 2.56 approx.
Thickness: .22 in approx.
Fluoresces orange in short wave range.
Elements present: Na, Si, Mg, Ca
 $N_D = 1.5150$, $N_F - N_C = .0082$ approx.
- 829 Refractive index (qualitative) on A and B.
- 831 Found difference with comparative density using brominated solvent and alcohol mixture.
- 832 Color and appearance of A and B similar.
Density gradient of A in mid-2.50 region.
- 833 Index of refraction: $1.5166 \pm .0002$
Ultraviolet cutoff: 337mu(50% absorption)
Thickness: $.2224" \pm .0002$
Density gradient: 245.2/310mm
Fluorescence: None visible short or long wave.
Fluorescence spectrum: Excitation - 250
Emission - 505mu
- 835 Specific gravity different in both samples.
Refractive index different in both samples.
Thickness of both samples averaged the same.
UV examination (short wave) - minute differences noted.
Spectrographic analyses - spectra revealed no significant differences.
- 837 UV light revealed no fluorescence.
Dispersion characteristics: match at 588nm
(oil 1.522 at 38°C)
- 838 Refractive index(sodium light): 1.5164
- 839 Thickness: .221"
Refractive index: Match at 625 to 503mu.
(1.520 oil at 34°C)
- 842 Negative on Flotation, Density gradient, and refractive index.
- Flouresce under UV light.
 $N_D^{25} - 1.5185$
- Density: 2.483
Refractive index: 589nm 1.521
488nm 1.528
- Density: 2.50 approx.
Thickness: .22 in approx.
Fluoresces green in short wave range.
Elements present: Na, Si, Mg, Ca
 $N_D = 1.5180$, $N_F - N_C = .0083$ approx.
- Density gradient of B in lower 2.50 region.
- Index of refraction: $1.5196 \pm .0002$
Ultraviolet cutoff: 333mu(50% absorption)
Thickness: $.2247" \pm .0002$
Density gradient: 250/310mm
Fluorescence: Short wave imparts yellow green fluorescence on one face.
Fluorescence spectrum: Excitation - 250
Emission - 370 and 503mu
Emission intensity approx. 10x greater than than A
- B fluoresced on one side to UV light.
Dispersion characteristics: match at 528nm
(oil 1.522 at 38°C)
- Refractive index(sodium light): 1.5185
- Thickness: .224"
Refractive index: Match at 574 to 509 mu.
(1.520 oil at 34°C)

Sample ASample B

- 843 Density: 2.480
Shear wave and longitudinal velocities vary to such an extent between A and B that they could not have a commonality of source.
- 847 Thickness: .2237 - .2238"
Refractive index (sodium line): 1.516
Dispersion (V) of A higher than B.
Qualitative difference in density.
- 848 Refractive index: match with silicone oil at 74.2°C.
- 849 Color, thickness, and refractive index of A and B.
- 852 Density quantitative differences in A and B.
Thickness, Refractive index, and Dispersion qualitative differences in A and B.
- 853 Thickness: .227"
Approx. Density: 2.400 gm/cm³
Refractive index: 1.5166
- 854 UV(short wave): no fluorescence
Relative density: A differs from B.
Refractive index: 1.513
X-ray fluorescence:
- $K_{\alpha}^{Ca}/K_{\alpha}^{Sr}$: .87 \pm .03
 $K_{\alpha}^{Ca}/K_{\alpha}^{Fe}$: 2.32 \pm .04
- 855 Thickness, color and tint, comparative density, and comparative refractive index.
- 856 Thickness: .2225 to .2230"
Appears blue under UV lamp.
A does not have same density as B.
A has lower refractive index than B (A less than 1.514 and B equal or greater 1.514).
- 863 Refractive index: 1.5157
Thickness: 5.61mm
- 866 Refractive index: 1.5145, 1.5147
Density: 2.4523, 2.4739
- 868 A and B are similar in color and width, B has a higher refractive index than A.
- 872 Density, Refractive index, Spectrographic composition.
- 873 Refractive index: 1.5156
Specific gravity of A different from B.
- 874 Has weak or absence fluorescence.
A has a lower density than B.
- 876 Diameter: 5.59mm
Refractive index: 1.516 at 589.6 mu
Density of A is less than B.
- 880 Visual observation showed A to have an apparent deeper green tint. Short wave UV showed greater fluorescence on A. Two samples differed by more than .001" in thickness. X-ray spectrometer revealed a difference in chemical composition. Rough refractive index determination showed a significant difference.
- Thickness: .224"
Approx. Density: 2.532 gm/cm³
Refractive index: 1.5185
- UV(short wave): yellow fluorescence
Refractive index: 1.515
- .81 \pm .03
3.04 \pm .04 (K_α^{Ca} -- preset at 20,000 counts)
- Thickness: approx. .2246"
Appears yellow under UV lamp.
- Refractive index: 1.5190
Thickness: 5.72mm
- Refractive index: 1.5176, 1.5177
Density: 2.4763, 2.49
- Refractive index: 1.5190
- Has strong yellowish-orange fluorescence.
- Diameter: 5.70mm
Refractive index: 1.519 at 589.6 mu

Sample A

883 Sample is not float glass.
Mean thickness: 5.66 mm
Refractive index: 1.51571

884 Thickness: .2235"
A is less dense than B.

885 Iron and Zirconium content of A different from B (X-ray fluorescence).
Infared scans on a differential basis showed differences in the composition of the two samples.

888 Thickness: 5.517 mm
Color difference
Dispersion curve is different between A and B.
Refractive index: $N_C = 1.5131$, $N_D = 1.5155$
 $N_F = 1.5223$

889 UV examination: no fluorescence
Refractive Index: 1.5158

892 A is less dense than B.
No tin by elemental analysis.
Refractive index of A less than B.

894 Density of A is less than B.
 $N_D^{45.2} = 1.5163$ (match temp. $N_D^{25} = 1.524$ is 45.2°C for sample A while $N_D^{48.0} = 1.5193$
(match temp. for Cargille liquid $N_D^{25} = 1.528$ is 48.0°C)

895 Density
897 Refractive index: 1.516
Thickness: .227 inch
Absorbs short wave UV light.
Density of A is less than B.

899 Thickness: $.2236 \pm .0003$
Refractive index: 1.5162
No UV fluorescence

907 A and B compared on basis of:
1. Density - differ (density gradient)
2. Refractive index - differ (interpolated from 3)
3. Dispersion - differ (McCrone dispersion staining)
4. Metallic composition - similar (x-ray fluorescence)

908 Refractive index: 1.512 - 1.516
Comparative density of A is less than B.

915 Thickness: .223 - .224"
Fluorescence under UV: no
 $n_D = 1.5161_8$
V: 55.82₇

Sample B

Sample is float glass.
Mean thickness: 5.70
Refractive index: 1.51866

Thickness: .2245"

Thickness: 5.5205 mm
Color difference
Refractive index: $N_C = 1.5161$, $N_D = 1.5186$
 $N_F = 1.5249$

UV examination: + fluorescence
Refractive index: 1.5181

Presence of tin by elemental analysis.

Density
Refractive index: 1.518
Thickness: .224 inch
One side fluoresces yellow under UV light.

Thickness: $.2253 \pm .0004$
Refractive index: 1.5174
Strong UV fluorescence one side only

Refractive index: 1.516 - 1.520

Thickness: .225" - .226"
Fluorescence under UV: yes
 $n_D = 1.5187_8$
V: 55.72₉

Sample ASample B

- 921 Refractive index: 1.5155
Refractive index: 1.5185
B has a greater density than A.
- 923 R.I., density, and gross physical measurements were made on both A and B.
- 925 Thickness: .223 inches
n_C = 1.5132, n_D = 1.5159, n_F = 1.5228
Density = less than 2.499 gm/cc³
Thickness: .225 inches
n_C = 1.5159, n_D = 1.5186, n_F = 1.5254
Density = 2.499 gm/cc³
- 926 Thickness: .2237"
A is less dense than B.
Thickness: .2248"
- 931 Refractive index: 411 nm 1.508 ± .001
593 nm 1.516 ± .001
Refractive index: 411 nm 1.511 ± .001
593 nm 1.519 ± .001
- 938 Density: 2.486 g/cc
Refractive index: 1.516 (white light)
Fracture initiation marks were different on A and B indicating that glass samples were broken differently.
Slight microscopic color difference between A and B.
Density: 2.496 g/cc
Refractive index: 1.519 (white light)
- 944 Density of A and B are different.
- 948 A has lesser density than B.
Contains Na, Si, Ca
A and B have similar Refractive index.
Contains Na, Si, Ca and trace of Mg
- 958 A and B were measured for glass color, glass thickness, refractive index, density.
- 960 Density gradient comparison made for A and B, refractive index, x-ray fluorescence, and UV light with B giving white fluorescence under short wave UV light.
- 961 B strongly fluoresced on one surface under UV illumination, while A exhibited a much weaker fluorescence under the same conditions.
A is less dense than B.
Refractive index measurements were not clear-cut.
- 962 N_C - 1.5139, N_D - 1.5164, N_F - 1.5228
N_C - 1.5155, N_D - 1.5182, N_F - 1.5249
- 969 No fluorescence under short wave UV light. Fluoresces under short wave UV light.
- 970 Physical, UV, Refractive index, etc.
- 974 Thickness: .2227"
Fluorescence: none
Density of A is less than B.
Refractive index: C - 1.5131, D - 1.5159
F - 1.5227
Thickness: .2245"
Fluorescence: pale yellow on one side
Refractive index: C - 1.5160, D - 1.5187
F - 1.5258
- 975 Refractive index: between 1.514 - 1.516
A is less dense than B.
A is darker blue green in color than B, similar thickness.
Refractive index: between 1.516 - 1.518

Sample A

978 $N_c = 1.5154$, $N_d = 1.5183$, $N_f = 1.5255$.

$N_f - N_c = .0101$

Density: 2.4767 @ 27.5°C

979 A is much less dense than B.

980 Measurements were made on A and B for refractive index, density, fluorescence, and elemental composition.

986 Refractive index: 1.5156

Thickness: 5.63 mm

Density of A is less than B.

987 Gradient density - different densities.

989 Composition, density, and refractive index of A and B.

994 A is less dense than B (significantly).

995 Thickness: .223 inches

Refractive index(refractometer) 1.5149

Refractive index(hot stage) 1.5150

A is less dense than B.

999 Color (tint)

Thickness

Density gradient

Refractive index: 1.522 at rm temp

Sample B

$N_c = 1.5182$, $N_d = 1.5212$, $N_f = 1.5282$

$N_f - N_c = .0100$

Density: 2.4956 @ 27.5°C

Refractive index: 1.5183

Thickness: 5.71 mm

Thickness: .223 inches

Refractive index(refractometer) 1.5180

Refractive index(hot stage) 1.5181

Color (tint)

Thickness

Density gradient

Refractive index: 1.522 at rm temp

Table 8

RESPONSES TO QUESTION 3: Methods and instruments used.

- | | |
|-----|--|
| 705 | <ul style="list-style-type: none"> 1) Refractive index (Microscopic) three monochromatic wavelengths - 700, 590, 490. 2) Dispersion staining. 3) X-ray fluorescence. |
| 709 | <ul style="list-style-type: none"> 1) UV short wave light. |
| 712 | <ul style="list-style-type: none"> 1) UV short wave light. 2) Thickness with calipers. 3) Refractive index using a monochromator, hot stage, stereo zoom microscope and Cargille liquids 1.520, 1.518, and 1.516. Measurements made by temperature variation at the D line. |
| 713 | <ul style="list-style-type: none"> 1) Thickness by micrometer. 2) Refractive index using polarizing microscope and Cargille liquids. |
| 717 | <ul style="list-style-type: none"> 1) Thickness by calipers. 2) Relative density with mixture of Methylene Iodine and Bromobenzene. 3) Refractive index using Cargille liquids, hot stage and Becke line phenomena. 4) Spectrographic analyses. |
| 718 | <ul style="list-style-type: none"> 1) Density gradient columns with mixture of bromobenzene and bromoform. 2) Refractive index using Cargille liquids, Becke line by Sodium lamp, hot stage, and microscope. |
| 719 | <ul style="list-style-type: none"> 1) Microscopy 2) X-ray fluorescence. |
| 722 | <ul style="list-style-type: none"> 1) Refractive index using microscope, Sodium light, Cargille liquids. 2) Density gradient using tubes and liquids. 3) UV short wave light. |
| 726 | <ul style="list-style-type: none"> 1) Refractive index using Cargille liquids. 2) Density by comparative flotation method. 3) UV short wave light. 4) Emission spectrograph. |
| 727 | <ul style="list-style-type: none"> 1) Refractive index using a microscope, monochromator, hot stage. 2) Sink float density method using mixture of bromoform and MeOH. |
| 729 | <ul style="list-style-type: none"> 1) UV short wave light. |
| 730 | <ul style="list-style-type: none"> 1) UV - Vis. continuous scan (800nm to 310nm). 2) Emission spectrograph. |
| 731 | <ul style="list-style-type: none"> 1) Visual examination - color & thickness. 2) Physical edge match. 3) Density gradient. |
| 739 | <ul style="list-style-type: none"> 1) Refractive index using microscopes (ordinary & polarizing), Cargille oils. 2) Density gradient tubes with mixture of bromobenzene and bromoform. 3) Thickness. |

- 740 1) Refractive index by refractometer.
2) Dispersive X-ray.
- 742 1) Thickness by thickness gauge.
2) UV short wave light.
3) Refractive index by Emmons double variation method using polarizing microscope, hot stage, monochromator, and silicone oil.
4) Emission spectrograph.
- 745 1) Density by tube flotation with mixture of bromoform and bromobenzene.
2) Refractive index using Cargille liquids, Becke line, and interference filters (489, 585, and 667 nm).
3) Thickness with vernier calipers.
- 746 1) Refractive index using microscope, Cargille liquids.
2) Density comparison using mixture of bromoform and bromobenzene.
- 747 1) Thickness by gauge.
2) Refractive index using refractometer.
3) Dispersion by Emmons Double Variation Method using contrast microscope, hot stage and monochromator.
- 748 1) UV short wave light.
2) Thickness by micrometer calipers.
3) Density by macro method using water and analytical balance.
- 750 1) UV lamp.
2) Sink-float comparison with mixture of monobromo benzene and bromoform.
3) Refractive index using monochromator, mixture of xylene and monobromo benzene.
- 751 1) Birefringence and micro appearance by polarizing microscope.
2) Thickness by direct reading gauge.
3) UV short and long wave lights.
4) Refractive index using refractometer.
5) Dispersion and refractive index by Emmons double variation method using microscope, hot stage, and monochromator and Cargille liquids.
- 752 1) Sink-float density method with solution of mercuric potassium iodide.
2) Refractive index using microscope, silicone fluid, hot stage and optical interference filter.
- 754 1) Visual examination for color.
2) UV short wave light.
3) Gradient density with commercial oils of 2.80 - 2.20
4) Refractive index using microscope, monochromator, hot stage, and silicone oil.
- 756 1) Thickness by micrometer.
2) Refractive index and dispersion by Emmons double variation method using microscope, hot stage, monochromator, and Cargille liquids.
- 757 1) UV light.

- 760 1) UV light.
2) Thickness using micrometer.
3) Refractive index using Cargille liquids, white light source and Becke line.
4) Sink-float density comparison using mixture of bromoform and ethanol.
- 762 1) Thickness by micrometer.
2) Density with mixture of liquids.
3) Refractive index using monochromator, microscope, controlled temperature water bath and artifited index of refraction liquids set.
- 763 1) Refractive index using Cargille liquids.
2) Float-sink comparison using $\text{CBr}_4/\text{CHCl}_3$ mixture.
- 765 1) Thickness by micrometer.
2) UV scan by spectrophotometer.
3) Refractive index by refractometer.
4) Density by benzene and bromoform mixture.
- 766 1) Refractive index by refractometer.
2) Density gradient tube comparison.
3) Melting point apparatus.
- 768 1) UV short wave light.
- 769 1) Thickness by caliper and micrometer.
2) Specific gravity and density using analytical balance and water displacement method.
3) Refractive index by refractometer.
4) Physical jigsaw match.
5) Density using bromoform and bromobenzene mixture.
6) UV short wave light.
7) Spectrofluorometric analysis using a spectrofluorometer.
8) Emission spectrograph for spectrographic analysis.
- 773 1) UV short wave light.
2) Density determination using bromoform and EtOH.
- 777 1) Thickness by micrometer.
2) Refractive index by refractometer.
3) Analytical balance.
- 778 1) Refractive index using hot stage and silicone oil.
2) X-Ray fluorescence.
- 779 1) Refractive index using Cargille liquids and hot stage.
2) Density by comparison using methylene iodide and bromobenzene.
- 783 1) Thickness by dial micrometer.
2) Refractive index by refractometer.
- 784 1) Density gradient using bromoform and bromobenzene.
2) Refractive index using Cargille liquids.
3) Elemental composition by X-Ray fluorescence.
4) Thickness using micrometer.

- 786 1) UV fluorescence using Chromato-Vue instrument.
2) Elemental analysis using emission spectrograph.
3) Density using gradient columns.
- 787 1) Physical jigsaw match.
2) Gradient density method.
- 789 1) Refractive index using Cargille liquids, microscope, and 440mu filter.
2) Sink-float density method using KHgI_3 .
- 790 1) Thickness using micrometer.
2) Density comparison using bromoform and bromobenzene.
3) Refractive index using contrast microscope, hot stage, silicone oils, spectrometer filters.
- 792 1) Refractive index using polarizing microscope with a phase contrast condenser and monochromatic filter.
- 794 1) UV light.
2) Thickness using micrometer.
3) Refractive index using refractometer.
4) Verification of glass samples using polarizing microscope.
- 796 1) Thickness using micrometer.
2) Specific gravity using analytical balance, water and tergitol.
3) Elemental analysis calculated to oxides - SEM + EDXA.
- 797 1) Refractive index using refractometer.
2) Refractive index by Emmons Double Variation Method using monochromator compound microscope, hot stage and Cargille liquids.
- 798 1) UV light.
2) Refractive index using microscope and Cargille liquids.
3) Gradient density using bromoform and bromobenzene mixtures.
- 799 1) UV, IR, and normal illumination.
2) Microscopic observation by stereo microscope.
3) Refractive index using refractometer and hot stage.
4) Thickness using micrometer.
5) Comparative density gradient and absolute density by Pycnometer.
6) Emission spectrograph.
- 805 1) Thickness by micrometer.
2) Flotation density using mixture bromoform-bromobenzene.
3) Dispersion staining.
4) Emission spectrograph.
- 806 1) Thickness using calipers.
2) Edge color comparison against white background.
3) Specific gravity using Mercuric potassium iodide solution.
4) Refractive index using microscope, hot stage, silicone oil, and sodium filter.

- 809 1) Thickness by micrometer.
2) Density gradient using bromoform-bromobenzine mixture.
3) Emission spectrograph.
- 813 1) Density using bromoform-bromobenzine mixture.
2) X-Ray fluorescence.
3) Dispersion staining using McCrone objective with Cargille oils.
- 815 1) Visual color using north light.
2) Thickness using micrometer.
3) Refractive index using Cargille liquids and 593nm interference filter.
4) Dispersion staining.
- 818 1) Sink-float density method.
2) Thermal gradient apparatus.
- 820 1) UV light.
2) Refractive index using refractometer.
- 821 1) Thickness using caliper.
2) Density gradient column.
3) Refractive index using microscope with narrow-band pass filters.
- 822 1) Micrometer.
2) Spectrophotometer.
3) Polarizer.
- 823 1) Refractive index using polarizing microscope, Cargille liquids, and three wavelength filters.
- 827 1) Refractive index and dispersion using refractometer.
2) Emission spectrograph.
3) Density gradient tube with bromoform and bromobenzene mixture.
4) UV lamp and fluorometer.
5) Thickness using caliper.
- 829 1) Refractive index using microscope with R & G filters, Cargille liquids, and hot stage.
2) Oil emersion - double variation method.
- 831 1) Mixture of 1, 1, 2, 2, tetrabromoethane and ethyl alcohol.
- 832 1) Density gradient tube with oils (3.0 to 2.2).
- 833 1) Refractive index using refractometer.
2) UV-visible on recording spectrophotometer.
3) Thickness using micrometer.
4) Density gradient using bromoform-bromobenzene mixtures.
5) UV short wave and long wave lights.
6) Fluorescence spectrum using fluorescence spectrometer.
- 835 1) Specific gravity using bromoform-bromobenzene mixture.
2) Refractive index using microscope and Cargille liquids.
3) Spectrograph.

- 837 1) Thickness using caliper.
2) UV light.
3) Dispersion characteristics using microscope, monochromator, and hot stage.
- 838 1) Refractive index using refractometer.
- 839 1) Thickness.
2) Refractive index using Cargille liquids.
3) X-Ray fluorescence spectrometer.
- 842 1) Refractive index.
2) Density gradient.
3) Floatation.
- 843 1) Density by pycnometer.
2) Time of flight using ultrasonic time of flight measurement instrument.
- 847 1) Refractive index and dispersion using microscope, monochromator, Cargille liquids, and hot stage.
2) Thickness using micrometer.
3) Sink-float density comparison using bromoform and nitrobenzene.
- 848 1) Refractive index using microscope, hot stage, silicone oil, and sodium light.
- 849 1) Visual.
2) Micrometer.
3) Refractometer.
- 852 1) Refractive index and dispersion using microscope, monochromator, Cargille liquids, and hot stage.
2) Thickness using micrometer.
3) Sink-float density comparison using bromoform and nitrobenzene.
- 853 1) Density gradient.
2) Refractometer.
- 854 1) X-Ray fluorescence.
2) UV short wave light.
- 855 1) Thickness using micrometer.
2) Comparative density using bromoform and bromobenzene.
3) Refractive index using liquids and hot stage.
4) Dispersion checked by repeating RI procedure at two other wave lengths.
- 856 1) Thickness using a paper gauge.
2) UV light.
3) Density using sink-float method with bromobenzene and methylene iodide.
4) Refractive index using microscope and Cargille liquids.
- 863 1) Refractometer.
2) Micrometer.
- 866 1) Refractive index using refractometer.
2) Density floatation method using bromoform and bromobenzene mixture.
- 868 1) Refractive index using Cargille liquids and 3 wavelength filters.

- 872 1) Density floatation method using bromoform-bromobenzene mixture.
2) Refractive index using hot stage, silicone oil, and 3 light filters.
3) Emission spectrographic composition.
- 873 1) Refractive index using refractometer.
2) Floatation method for comparative specific gravity.
- 874 1) UV light.
2) Bromoform-bromobenzene mixture.
- 876 1) Diameter using micrometer.
2) Refractive index using Cargille liquids and band pass filters.
3) Density column using bromobenzene-bromoform.
- 880 1) Visual observation.
2) UV Light.
3) X-Ray spectrometer.
4) Rough density gradient.
5) Rough refractive index.
6) Dispersion curves using hot stage and optical wedge.
- 883 1) UV light.
2) Refractive index by silicone oil and Dabbs and Pearson method.
- 884 1) Thickness using calipers.
2) Floatation method for density.
- 885 1) Thickness.
2) UV light.
3) Infra-red spectrophotometric analysis.
4) Differential infra-red spectrophotometric analysis.
5) X-Ray fluorescence.
- 888 1) Thickness using micrometer.
2) Refractive index using polarizing microscope, hot stage, monochromator, and Cargille liquids.
3) Data plotted on dispersion graph paper.
- 889 1) UV light.
2) Refractive index using refractometer.
- 892 1) Density gradient using bromoform-bromobenzene mixture.
2) Elemental analysis using spectrographic analyzer.
3) Refractive index using photomicroscope, hot stage, Cargille liquids, and red filter.
- 894 1) Density gradient using benzene and bromoform.
2) Refractive index using Cargille liquids and hot stage.
- 895 1) Density.
- 897 1) Thickness using micrometer.
2) Refractive index using refractometer.
3) UV light.
4) Density gradient tubes.

- 899 1) Micrometer.
2) Refractive index using refractometer.
3) UV short and long wave lamps.
- 907 1) X-Ray fluorescence.
2) Dispersion staining.
3) Refractive index.
4) Density gradient.
- 908 1) UV light.
2) Refractive index using Cargille liquids.
3) Visible color.
4) Comparative density using two mixtures bromoform-xylene and bromoform-ethanol.
- 915 1) Thickness using micrometer.
2) UV short wave light.
3) Refractive index and dispersion using Emmons Double Variation Method with hot stages, microscope, monochromator, and Cargille liquids.
- 921 1) Refractive index using refractometer.
2) Comparative density using bromoform and bromobenzene.
3) Emission spectrum.
- 923 1) Thickness using micrometer.
2) Density column.
3) Becke line using microscope.
4) Phase contrast microscopy with temperature variation.
5) Dispersion staining objective and hot stage with monochromator.
- 925 1) Refractive index using contrast microscope with microfurnace and optical filters.
2) Density according to Greene and Burd method.
- 926 1) Thickness by micrometer.
2) Flootation method for relative density.
- 931 1) Refractive index using polarizing microscope, phase microscope, microfurnace, and narrow band interference filters.
- 938 1) Refractive index using Cargille liquids and hot stage.
2) Density using bromoform-bromobenzene, and 25ml pycnometer.
- 944 1) Density gradient.
- 948 No listing given.
- 958 1) Color by stereo microscope.
2) Thickness using micrometer.
3) Physical matching of edges.
4) Refractive index using refractometer and microscope with Cargille liquids.
5) Density gradient using bromoform-bromobenzene mixtures.

- 960 1) Refractive index using microscope and Cargille liquids.
2) X-ray spectrum.
3) Density column using methylene iodide and bromoethane.
4) UV light.
- 961 1) UV light.
2) Density by sink-float method using bromoform and mono-bromobenzene.
3) Refractive index using microscope, color filters and mixtures of benzyl alcohol and di-n-butylphthalate.
- 962 1) Emmons double variation method using microscope, hot stage, monochromator, and silicone oil.
- 969 1) UV lamp.
- 970 No listing given.
- 974 1) Density by sink-float method using bromoform and alcohol mixture.
2) UV short wave light.
3) Refractive index using microscope, hot stage, monochromator, silicone oil and dispersion graph paper.
- 975 1) Density comparison using tetrabromoethane and tetrachloride mixture.
2) Refractive index using microscope and Cargille liquids.
- 978 1) Refractive index by refractometer, microscope, monochromator, and hot stage.
2) Density by buoyancy method using balance.
3) UV short wave instrument.
- 979 1) Physical measurements.
2) Density.
3) Optical density.
- 980 1) Refractive index using refractometer.
2) Density gradient method.
3) Elemental composition using emission spectroscopy.
4) UV radiation.
- 986 1) Refractive index using refractometer.
2) Thickness using micrometer calipers.
3) Density gradient tubes.
- 987 1) Density gradient tubes using bromobenzene and bromoform.
- 989 1) Emission spectrometer.
2) Density.
3) Refractive index using Cargille liquids.
- 994 1) Density using bromoform and benzene mixtures.
- 995 1) Refractive index using refractometer.
2) Refractive index using hot stage, silicone oil, and Pearson and Dobbs method.
3) Comparative density using bromobenzene and tetrabromoethane.
- 999 1) Visual for color.
2) Thickness using calipers.
3) Density gradient using tubes with a variety of liquids.
4) Refractive index using Cargille liquids and a microscope.

Table 9

METHODS

Lab Code	Color	Density	Dispersion Curves	Dispersion Staining	Elemental Analysis	Physical Edge Match	Refractive Index	Thickness	U.V. Light	X-ray fluorescence	Other
705				X			X			X	
709									X		
712							X	X	X		
713							X	X			
717		X			X		X	X			
718		X					X				
719										X	Microscopy
722		X					X		X		
726		X			X		X		X		
727		X					X				
729									X		
730					X				X		Visible scan
731	X	X				X		X			
739		X					X	X			
740							X				Dispersive X-ray
742					X		X	X	X		
745		X					X	X			
746		X					X				
747			X				X	X			
748		X						X	X		
750		X					X		X		
751			X				X	X	X		Birefringence
752		X					X				

Lab Code	Color	Density	Dispersion Curves	Dispersion Staining	Elemental Analysis	Physical Edge Match	Refractive Index	Thickness	U.V. Light	X-ray Fluorescence	Other
754	X	X					X		X		
756							X	X			
757									X		
760		X					X	X	X		
762		X					X	X			
763		X					X				
765		X					X	X	X		
766		X					X				
768									X		
769		X			X	X	X	X	X		Specific gravity Spectro- fluorometric anal sis
773		X							X		
777			X				X	X			
778							X			X	
779		X					X				
783							X	X			
784		X			X		X	X			
786		X			X				X		
787		X				X					
789		X					X				
790		X					X	X			
792							X				
794							X	X	X		Verification of glass us- ing polarizing microsc
796					X			X			Specific gravity
797			X				X				
798		X					X		X		
799		X			X		X	X	X		IR and normal illumination
805		X		X	X			X			

Lab Code	Color	Density	Dispersion Curves	Dispersion Staining	Elemental Analysis	Physical Edge Match	Refractive Index	Thickness	U.V. Light	X-ray Fluorescence	Other
806	X						X	X			Specific gravity
809		X			X			X			
813		X		X						X	
815	X			X			X	X			
818							X		X		
820							X		X		
821		X					X	X			
822								X			Spectrophotometer, polarizer
823								X			
827		X			X		X	X	X		
829			X				X				
831		X									
832		X									
833		X					X	X	X		visible scan, fluorescence spectrometer
835					X		X				specific gravity
837			X					X	X		
838							X				
839							X	X		X	
842		X					X				
843		X									ultrasonic time of flight measurement instrument
847		X					X	X			
848							X				
849	X						X	X			
852		X	X				X	X			
853		X					X				
854									X	X	

Lab Code	Color	Density	Dispersion Curves	Dispersion Staining	Elemental Analysis	Physical Edge Match	Refractive Index	Thickness	U.V. Light	X-ray Fluorescence	Other
855		X	X				X	X			
856		X					X	X	X		
863							X	X			
866		X					X				
868							X				
872		X			X		X				
873		X					X				
874		X							X		
876		X					X	X			
880	X	X	X				X		X		X-ray spectrometer
883							X		X		
884		X						X			
885								X	X	X	Infra-red spectro analysis differential infrared anal sis
888			X				X	X			
889							X		X		
892		X			X		X				
894		X					X				
895		X									
897		X					X	X	X		
899							X	X	X		
907		X		X			X			X	
908	X	X					X		X		
915			X				X	X	X		
921		X			X		X				
923		X		X			X	X			phase contrast microscopy
925		X					X				

Lab Code	Color	Density	Dispersion Curves	Dispersion Staining	Elemental Analysis	Physical Edge Match	Refractive Index	Thickness	U.V. Light	X-ray Fluorescence	Other
926		X						X			
931							X				
938		X					X				
944		X									
948											
958	X	X				X	X	X			
960		X					X		X		X-ray spectrum
961		X					X		X		
962			X				X				
969									X		
970											
974		X	X				X		X		
975		X					X				
978		X					X		X		
979		X									physical measurement
980		X			X		X		X		
986		X					X	X			
987		X									
989		X			X		X				
994		X									
995		X	X				X				
999	X	X					X	X			

END