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Report Title: Scale Modeling in Fire Reconstruction

Award Number: 2008-DN-BX-K178

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Abstract

Scale modeling can allow fire investigators to replicate specific fire dynamics at a dramatically reduced cost. A gas burner, liquid pool, wood crib, and polyurethane foam block are used to represent the wide range of fuels that investigators encounter. These fuels are classified into two groups: the burner and liquid pool that reach a semi-immediate steady state (static fires) and the crib and foam that have a fire spread and growth period (dynamic fires). This research examines the proposed scaling method for the static fires. The enclosure consists of a large corridor that provides an interesting challenge due to the presence of partitions at the ceiling. The design fires and the model enclosure are designed based on Froude scaling derived from conservation equations. The eight various sized fires demonstrate acceptable scaling results in the prediction of flame height and temperature at various elevations in the enclosure.

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Executive Summary

This project was undertaken to examine the use of scaling modeling in fire. In particular it was intended to bring the subject to the fire investigator. In that regard cooperation with the ATF Fire Laboratory and presentations at two annual meeting of the International Association of Arson Investigators (IAAI) were undertaken. These activities helped to transfer the technology, in particular, the IAAI presentations were 3-4 hour training sessions.

Scale modeling is an approach to design and analysis that has pervaded many aspects of engineering. It is well know that without scale modeling studies in a wind tunnel, the Wright brothers would not have had a successful flight. The key parameter in the wind tunnel is the Reynolds number that is the ratio of momentum to viscous forces that govern the airfoil. Scale modeling is based on such dimensionless parameters as the Reynolds number (Re) that says if you preserve the Re between your model and full-scale, and you scale the airfoil geometrically, then the velocity in the scale model wind tunnel must be higher according to maintaining Re constant. In the case of aircraft and other viscous and pressure drag phenomena, such as "streamline" shapes for vehicles, only the Re is relevant. However, if one considers the design of ships, the wave action produces another drag effect and an additional dimensionless parameter becomes relevant. That is the Froude number (Fr) that is the ratio of the momentum to gravity force. In this case, the Fr scaling conflicts with the Re with two different results for velocity in the scale model. In this case, "partial scaling" is adopted, that favors the Fr over the Re in scaling velocity. Tow tanks for boats follow this scaling. In general, the scaling of complex phenomena requires the adopting of partial scaling rules. These rules can be justified by the particular level of physics or chemistry that one is interested. It becomes the art of applying science. Many fields use scale modeling from aircraft design, to noise abatement, concert acoustics, wind loading of buildings and bridges, and even the tidal flow dynamics in the Netherlands. Its value must be established in each case through demonstrative examples comparing the results between the model and the full-scale.

The approach of this study is to review the methods of fire scaling, and to apply the scaling hypothesis to specific examples in which good full-scale data exist. In this report fire scaling in air will be discussed. The flows associated with fire can also be modeled by using the density differences between saltwater and fresh water. This analog approach of using water to represent buoyant flows in fire is quite attractive. It will be presented in a companion report to this NIJ project. Here we will henceforth focus on fire scaling in air.

Scale modeling in fire has been used sporadically since the beginning of the study of fire. It is fair to say that it has been looked upon as dubious, yet fascinating. P. H. Thomas was an early advocate of the UK Fire Research Station, now the British Research Establishment (BRE) in the 1960's. He demonstrated both the use of geometric scale models in smoke venting, as well as the use of dimensionless groups to correlate a range of fire phenomena. G. Heskestad of FM Global research demonstrated similar techniques. In particular, he developed a simple formula to predict the burning rate of wood cribs at various scales, and related compartment fire data at several scales. W. Parker of the National Institute of Standards and Technology (NIST) pushed the envelope in using scale modeling to predict the fire growth in rooms with combustible linings. K. Saito (University of Kentucky) took on the modeling of large conflagrations such as Dresden and Hamburg fires following bombings in WWII. Despite all of this work over the past 50 years, there is not a widespread appreciation for the use of scale modeling in fire.

Fire scale modeling cannot be perfect, as it cannot satisfy all of the variables that govern its behavior. For simple issues of smoke movement in buildings it does a fairly complete job. As in the scale modeling of ship dynamics, it neglects the Re but insures that the model scale is large enough to achieve adequate turbulent flow behavior. It has been generally found that models for smoke movement of about 0.3 m tall in room height are sufficient. For complex fire issues of growth involving real materials in spread and ignition, fire scale modeling can only capture some features. It cannot be perfectly quantitative, nor predictable over time. We shall examine these two extreme cases of smoke movement to fire growth in this study. From the details of the modeling, the trade-offs, and the distortions, a full appreciation should be gained for the value of scale modeling. Moreover, we shall examine the use of scale modeling in compartment fires. It should be realized that it could be used to study other fire phenomena such as forest fires, conflagrations, and external structural fire effects.

Typically in the scaling of fire phenomena, the physical dimensions are completely scaled down by a fixed factor. For example, the external dimensions of all objects, and the internal dimension of a room scaled down by a factor of 4, would be called "quarter-scaling". At time there are exceptions to this, as is the case Parker used for modeling lining fires. He reasoned that the flow rate of air through the door should follow the Fr number used in fire. That Fr number is given a new name Q^* (the Zukoski number, for Ed Zukoski of Califronia Institute of Technology who introduced it). This parameter Q^* has no explicit velocity in it, except for a velocity related to buoyancy, the driving force for flows in fire. That buoyancy velocity can be represented as the square-root of the gravity force per unit mass times the height. Its units can display its relationship to velocity:

Square-root of (g, Newton/kilogram = $m/s^2 x$ Height, m) = m/s, units of velocity.

The parameter Q^* is given as

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$$Q^* = \frac{\dot{Q}}{\rho_o c_p T_o g l^{5/2}}$$

Firepower (kW)

 $\overline{\text{Density}(\text{kg/m}^3) \text{ Specific heat}(\text{kJ/kg-K}) \text{ Temperature}(\text{K}) \text{ gravity}(\text{m/s}^2) \text{ Length}^{5/2}(\text{m}^{5/2})}$ $= \frac{\text{kW}}{\text{kJ/s}} = \frac{\text{kW}}{\text{kW}}, \text{ dimensionless.}$

It can be thought of the as the ratio of firepower (heat release rate by the fire) to the flow rate of energy due to buoyancy. Normally in a scale model, the Q* remains the same in the full-scale and the model. As the fire is done on Earth (g constant) and in air with the same initial density, specific heat and temperature, then the firepower to length to the 5/2-power must be constant. For a quarter-scale geometric model, the firepower in the scale model would be $(1/4)^{5/2}$, or 1/32 of full-scale fire power.

Difficulties immediately emerge as one tries to keep the O^* constant. The firepower in full-scale may not be known to begin with, as the growth of the fire would make it do what it wants. Then there should be considerations in other scaling parameters to insure that the scaled energy always is in sync. This cannot be done with perfection; so distortions arise. In Parker's strategy, he reasoned that for lining fires, the firepower would be proportional to the area or length-squared. He then reasoned that the flow rate of air through a door or window would be proportional to the area of the opening times the square root of the opening height (buoyancy velocity). For his Q*, he required the width of the opening to be proportional to the square root of the height. The rest of the dimensions would be scaled as ¹/₄. Thus we see the complications, imperfections and distortions possible in scale modeling. In other cases, the times for event might also be different between the scale-model and full-scale, or the thickness and type of materials used. In Parker's case, he used the same materials, and reasoned time was invariant. He achieved remarkable results in predicting fire growth in his fire scenario. By ignoring issues of fire spread, radiation heat transfer, heat losses and other factors, the reason for his success is not clearly apparent. In this study we will examine the Parker opening distortion, and an alternative to maintaining geometric similarity of the openings. In addition, some analyses will be put forward to try to explain the likely success or distortion of the model results.

Two scenarios will be examined herein. They are at the extremes of modeling from relatively simple to complex fire phenomena. Also both scenarios were available from the ATF Fire Research Laboratory with detailed full-scale data.

The first scenario was the effect of small or early fires on detection in a tall bay corridor-like space. The main corridor had two stub right-angle branches, and the main ends were fully open. The ceiling had an array of thin baffle-beams that repeated along the main corridor. The beams did not fully extend to the wall, so that flow could go around them as well as over them. Temperature arrays were measured longitudinally and vertically along the main corridor. In addition smoke detectors and smoke light attenuation instruments were set in some of the beam bays. Various fuels were burned in the full-scale tests. University of Maryland students, with the help of the ATF Fire laboratory staff, did the tests. Several steady fires were examined in the scaling including a gas burner and a liquid pool fire. The full details of this study will be reported in Article III of

this report. The full-scale corridor presented a complex flow pattern from the fire on the floor. The smoke and combustion products flowed to the ceiling, then around and over the beams, filling each bay as it moved toward the exits. The fire was small so that the temperature rise at points along the corridor did not rise significantly (of the order of 100 K). The walls and ceiling were gypsum board and the floor concrete.

In developing a scale model for this scenario, the geometric scale first has to be selected. The fullscale was 4.5 m tall, and a 1/8th geometric scale was selected, given a model height of 56.2 cm. This was tall enough to allow the student to scrunch into the model, lying on her back, to install the thermocouples along the ceiling. In addition, at this height the model should provide turbulent flow where necessary. Primarily temperature and flame height were measures of testing the model result with the full-scale. The scaling was based primarily on Q^* , and in addition the scaling specifically addressed convection and conduction into the boundary materials. To accomplish this last scaling, the materials of construction in the model could not be the same as full-scale. In addition, the time scaling of the model followed flow transport scaling so that the flow would reach a proportional position at an earlier time than full-scale. This scaling is time is proportional to the square root of the length scale. So time in the model occurs faster than full-scale by a factor of $(1/8)^{1/2}$ or about 0.35. The scenario considered checks with a gas burner and liquid pool fire. The same fuel was used in the scale model as full-scale. The energy release rate of the model fuel had to be scaled according to Q^* , or proportional to length to the power 5/2. Radiation aspects were ignored here because of the small fires and relatively low temperatures. However, radiation effects will distort the scaling result to some degree.

If the scale were perfect here, the temperatures would be the same at corresponding positions in the two systems, and time would be different as discussed above. In other words, at the same location of the model and full-scale the temperatures should be the same at full-scale and the lesser model times. The accuracy is shown in Article III, and demonstrates that this type of scaling is quite accurate. This will also be supported in the companion part of this project using salt-water modeling for the same full-scale system. In other words, if the fire is small, not growing, and radiation effects can be neglected; the results are nearly guaranteed to be good. On the other hand, for growing fires the scaling is deficient in accommodating all the phenomena. The only course is to recognize this, state a rationale scaling hypothesis, and see how it might deviate.

The second scenario examines the case where scaling is questionable. This scenario considered fire growth in a bedroom, in which the fire began in a plastic wastebasket filled with newspaper. That fire ignited adjoining bedding, and the side of an upholstered chair. Carpeting was on the floor, and the walls and ceiling were painted gypsum board. Other items of furniture filled the room. The doorway provided the only means of flow to and from the room.

A variety of instrumentation was measured during the full-scale tests. These were done at the ATF Fire Laboratory some years prior to the start of this project. The tests were done to investigate fire patterns following flashover, and were conducted under the auspices of the International Association of Arson Investigators (IAAI). Arrays of temperature were measured throughout the room, total heat flux was measured at the mid-height on walls, and gas composition measurements were conducted at several points; namely, carbon dioxide, carbon monoxide, and oxygen. Video

photography was taken at several locations including the starting fire, the doorway and other directions into the room.

In the scale model tests we considered all of these measurement types, but to the same extent as used in full-scale. The scale modeling tests were also done at the ATF Fire Laboratory. They supported in part a special ATF agent certification assignment to student the fire patterns in small scale.

The inability to address all aspects of fire growth through scaling principles necessitated a bold and simple hypothesis for the scaling. The hypothesis stated that all corresponding material shall be the same between the model and full-scale, and the room geometry and external dimensions of all the furniture should be ¹/₄ in scale. However, the thickness of all the construction and combustible material would be invariant, as possible. Thus, the construction of the furniture required special consideration in the model. For example, the gypsum board and the carpet could be the same. However, the frame of a chair or bed, and the soft coverings had to be invariant as possible to maintain overall geometric scaling. More details are presented in Article IV.

Analysis of the modeling strategy indicated that flame spread and burning rate might occur faster in the model during the early growth. However, as the smoke layer became hotter and darker, its radiation pushed the fire growth faster in the full-scale. When the flame dimensions in the model and full-scale were geometrically similar, scaling indicated that corresponding temperatures and gas concentration should be the same in the model and full-scale. This was validated, along with the growth behavioral differences. Moreover, the scale model nearly duplicated the phenomena of the full-scale, if not perfect over time, but in sequence and form. Its results were even surprising to us; as such work had not been done before in such detail.

In our view the testing of both scenarios offers a full perspective on the accuracy and benefits of scale modeling in fire. The early fire scenario presents a case where scale model should be accurate and appropriate. This scenario is applicable to issues of smoke detection and smoke movement. In contrast, the second scenario presents a real challenge in modeling. While it cannot be established as complete, and is partial scaling, it did offer some significant value. The phenomena were well represented in the model, the timing was off but predictable, and the cost of a model is far less than full-scale tests.

It is fair to say these results had an impact on the AFT fire staff and investigators. The results were presented at a course for the ATF current investigators. Two teaching sessions were performed at recent IAAI national meetings. Also an overview was presented at an international conference of fire protection engineers.

The remainder of this report is comprised of distinct aspects of this project. The first, Article I is based on the overview presented at the Society of Fire Protection Engineering conference on Fire Safety Design Methods. Article III is an MS thesis at the Department of Fire Protection Engineering (FPE), University of Maryland-College Park. It can be found at the website <u>www.fpe.umd.edu</u>. Finally, Article IV is an informal report drafted in collaboration between the University of Maryland (FPE), ATF Fire Laboratory, and the ATF certification program. The

various "Articles" are presented in distinct form, as they were developed independently. It would be too difficult to merge these perfectly. However, they act as chapters in a book.

We think this presentation on scale modeling should serve the reader with a good knowledge base. We present aspects in complete terms, show how they apply from the governing equations, and illustrate how to conduct the scaling. The results speak for themselves, and the user can decide the benefits. The video comparisons rendered by the ATF Fire Laboratory of the bedroom fire scaling are dramatic, and perhaps they could be made available to NIJ.

Article I. Introduction to Scale Modeling

The area of fire investigation and fire safety design is commonly engaged in the use of computer models or formulas. The use of computer models to address complex fire problems is limited by the scale of phenomena that have to be predicted within the grid spacing of the numerical model. Moreover, many phenomena, such as the formation of soot, the unraveling of veneer wood paneling in flame spread, and water droplet breakup -- not to mention turbulent combustion -cannot be represented in adequate form for the computer. So a computer model may lack accuracy and completeness. On the other hand, formulas for specific phenomena are usually grounded in data. The data has generally been taken in the laboratory with some variation in scale, and over a range of relevant parameters. These data are then subject to an analysis using some theory and dimensionless parameters that extend the resulting correlation. Many such correlating formulas have found consensus by their widespread testing and adoption. For a singular phenomenon these formulas are usually accurate to +/- 25 % and can serve as benchmark tests for a computer. For example, the temperatures and velocity in a fire plume, the flame height of the plume, and its entrainment rate all have served as computer modeling tests. Perhaps more should be done before a computer model is used to predict many phenomena, such as the fire plume in a room, its ceiling jet, the burning rate of fuel, and the temperature field in the room.

When all the room dynamic fire phenomena need to be addressed an approach often overlooked is scale modeling. Other fields use scale modeling, most notably the design of aircraft in a wind tunnel. Even the Wright brothers used this technique to their advantage. It might be surprising to some how widespread is the use of scale modeling. It is not without its limitations. Thomas [1] wrote a telling paper on it with the words: "a magic art" in the title. The complex world of fire, especially confined in a room, cannot be brought to perfect similitude as that of an aircraft in subsonic flight that uses only the Reynolds number as its basis. But experience has told me that the use of scale modeling in fire for design and investigation is overlooked. It may not be perfect in preserving all dimensionless groups, but with an understanding of their role the main phenomena can be addressed. This paper is primarily based on my experience using scale modeling, and the omission of other work is not to slight it, as the paper is not meant as a review. Indeed, I would encourage the reader to seek out further examples in the field. Neither is this paper intended as a treatise for scale modeling. In that regard I refer the reader to the list of references, and perhaps my chapter on scale modeling [2]. As a point of reference a table from that chapter is inserted here to display the multitude of phenomena that can be involved in fire. The nomenclature is generally obvious and more definition can be found in the book. The table lists the dimensionless groups found in fire phenomena. As examples of scaling are discussed, some of these groups will be identified. Groups pertaining to structural scaling in fire are not shown in the table, but this aspect will be discussed as well.

Table 1. Dimensionless Variables and Scaling in Fire

Variable/Group

DEPENDENT:

Velocity, uTemperature, TPressure, pConcentration, Y_i Droplet number, nDroplet diameter, D_l

Burning rate per area, \dot{m}_{F}''

INDEPENDENT:

Coordinates, x,y,z

Time, t

PI GROUPS:

 $Re = \frac{\rho_{\infty}\sqrt{gl^{3/2}}}{\mu}$ $\frac{\dot{Q}}{\rho_{\infty}c_{p}T_{\infty}\sqrt{gl^{5/2}}}$ $\Pi_1 \left(\frac{\text{inertia}}{\text{viscous}} \right)$, Re usually ignored $(u \sim l^{-1})$ $\Pi_2\left(\frac{\text{firepower}}{\text{enthalpy rate}}\right), Q^*$ significant in combustion $\Pi_3\left(\frac{\text{radiant emission}}{\text{ideal emission}}\right)$ $\kappa \sim l^{-1}$, when gas is important κl $\Pi_4\left(\frac{\text{radiant loss}}{c}\right), X_r$ $X_r \sim l^{0}$, important for free $X_r = \dot{q}_r / \dot{O}$ burning $\Pi_5\left(\frac{\text{conduction}}{\text{enthalpy}}\right), Q_k^*$ $\frac{(k\rho c)_{w}^{1/2}}{\rho_{w}c_{v}g^{1/4}l^{3/4}}$ $k_w \sim \rho_w \sim l^{-3/4}$, conduction important $\Pi_6\left(\frac{\text{convection}}{\text{anthelmy}}\right), Q_c^*$ $\frac{h_c}{\rho_c c_c \sqrt{gl}}$ $h_c \sim l^{1/2}$, convection important $\Pi_7\left(\frac{\text{radiation}}{\text{enthalpy}}\right), Q_r^*$ $\frac{\sigma T_{\infty}^{3}}{\rho_{\infty}c_{n}\sqrt{gl}}$ $T_{\infty} \sim l^{1/6}$, inconsistent with

others

Table 1. Dimensionless Variables and Scaling in Fire (cont.)

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Dimensionless

 $\hat{u} = u / \sqrt{gl}$

 $\hat{T} = T / T_{\infty}$ $\hat{p} = p / \rho_{\infty} g l$

 $\dot{m}_{F}^{\prime\prime} l \,/\, \mu$

 $r^{\infty \mathcal{S}^l}_{Y_i \, / \, Y_{i,\infty}} \ n \, / \, n_{ref}$

 $\frac{x_i / l}{t / \sqrt{\frac{l}{g}}}$

 D_l/l

Scaling/Comment

 $u \sim l^{1/2}$

 $T \sim l^0$

 $P \sim l$

 $Y_i \sim l^0$

 $n \sim 1^{3/2}$

 $\Pi_{12} \rightarrow D_l \sim l^{1/2}$

 $\dot{m}_{F}'' \sim \frac{h_{c}l}{\mu c_{n}} = \frac{Nu}{\Pr}$

 $x_i \sim l$

 $t \sim l^{1/2}$

Variable/Group **Dimensionless** Scaling/Comment $\left(\frac{\rho c}{k}\right)^{\frac{1}{2}} \left(\frac{g}{l}\right)^{\frac{1}{4}}$ $\Pi_8\left(\frac{\text{thickness}}{\text{thermal length}}\right)$ $\delta_w \sim l^{1/4}$, thickness of boundaries $\frac{\dot{m}_{Fan}}{\rho_m \sqrt{gl^{\frac{5}{2}}}}$ $\Pi_9\left(\frac{\text{fan flow}}{\text{advaction}}\right), \ m^*_{Fan}$ $\dot{m}_{Fan} \sim l^{5/2}$, forced flows $\Pi_{10}\left(\frac{\text{fuel flow}}{\text{advantian}}\right), m_F^*$ $\frac{\dot{m}_{F}}{\rho \sqrt{gl^{\frac{5}{2}}}}$ Fuel mass flux depends on B, Gr, Re, etc. $\Pi_{11}\left(\frac{\text{sensible}}{\text{latent}}\right), \tau_o$ $c_p(T_v - T_\infty)/L$ Burning rate term $\Pi_{12}\left(\frac{\text{available }O_2}{\text{stoichiometric }O_2}\right), r_o$ $Y_{o_2,\infty} / rY_{F,o}$ Burning rate term $\Pi_{13}\left(\frac{\text{evaporation energy}}{\text{sensible energy}}\right)$ $M_{g}h_{fg}/RT_{i}$ "Activation" of vaporization $\Pi_{14}\left(\frac{\text{collision loss}}{\text{initial particles}}\right)$ $\hat{n}_{col} = \dot{n}_{col} / \left(\frac{\dot{V}_{l,o}}{D^3} \right)$ $\dot{n}_{col} \sim l$, collision number rate $\Pi_{15}\left(\frac{\text{spray thrust}}{\text{iet momentum}}\right)$ $F_{o} / \left(\frac{\dot{V}_{l,o}}{D}\right)^{2}$ $F_{\rm o} \sim l^{-3}$, $D_{\rm o}$ nozzle diameter, $D_{\rm o} \sim l$ $\Pi_{16}\left(\frac{\text{evaporation rate}}{\text{droplet mass loss}}\right)$ $\dot{m}_{\circ}'' \sim l^{0}$ $\dot{m}_{a}^{\prime\prime}/\rho_{l}D_{l}\sqrt{gl}$ $\Pi_{17} \left(\frac{\text{weight of droplet}}{\text{drag force}} \right), \hat{D}_{\mu}$ $\hat{D}_{ii} = D_i \operatorname{Re}_i^{\frac{1}{3}}$ $D_l \sim l^{1/2}$ $\Pi_{18}\left(\frac{\text{advection}}{\text{mass transfer}}\right)$ $\Pr^{\frac{2}{3}} \hat{D}_{i}^{\frac{1}{2}} \operatorname{Re}_{i}^{\frac{1}{2}}$ $D_l \sim l^{-1/4}$, inconsistent with Π_{17} $\Pi_{19}\left(\frac{i^{\text{th}}\text{enthalpy}}{chemical energy}\right)$ $v_i \sim l^0$ $y_i c_n T_m / \Delta h_c$

$$\Pi_{20} \left(\frac{\text{droplet momentum}}{\text{surface tension}} \right) \qquad \text{We} = \rho_l u_l^2 D_l / \sigma_3 \qquad D_l \sim l^{-1}, \text{ inconsistent with } \Pi_{17}$$

$$\Pi_{21} \left(\frac{\text{enthalpy}}{\text{combustion energy}} \right) \qquad rc_p T_{\infty} / Ah_k \qquad \text{nearly always constant}$$

$$\Pi_{22} \left(\frac{\text{convection}}{\text{conduction}} \right) \qquad \text{Nu} = h_c l/k \qquad h_c \sim l^{-1}$$

A pioneer in the use of scale modeling in fire is G. Heskestad, who inspired me to use it. His work on compartment fire modeling [3] and on suppression by water droplets [4] should be studied. Early in my experience in fire research I was intrigued with the ability of Parker and Lee to predict flashover in the burning of lining materials in a room by a scale model at ¹/₄ the geometric size [5]. These works inspired me to explore scale modeling in a variety of applications. References [6-19] represent my experience, and they are listed in chronological order. The paper gives an overview of these applications, and the interested reader might wish to seek out the details in each reference. Also of interest might be to explore how scale modeling is used in other fields. The scaling symposia founded by Professor R.I. Emori and carried on by Professor K. Saito [See references 9 and 18] contain a vast array of scaling in many fields of engineering.

MAIN FEATURES OF FIRE SCALE MODELING

Many dimensionless parameters are shown in Table 1. As Thomas said there is a "magic art" to the process. Only a few groups can be preserved in scaling. As in the scaling of ship dynamics, in fire scaling the Reynolds number is not preserved but as full-scale flows are turbulent, the size of the model must be big enough to ensure turbulent flow. This is generally about 0.3 m (1 ft.) in height as a minimum. The key parameter is to preserve Π_2 or Q^* , the Zukoski number. As is often the case in computer modeling, this requires that the firepower (or more commonly the heat release rate) must be known for the full-scale. The ability to perfectly scale fire growth is impossible, as too many groups are required for preservation, and they cannot be controlled. They have a mind of their own. Yet by understanding how they might behave and distort the scaling, a scale model with fire growth can still be revealing and useful. Indeed the ultimate key is to preserve enough groups, first principally Q^* , so that the scale model data yield at least the dependent variables: temperature, velocity, and species concentrations. To get the species right, the same fuel must be used in the model and full-scale. These dependent variables are then related at corresponding dimensionless position and time. The geometry is fully scaled by the scaling factor, length of model to length of full scale. Time is often scaled by the "flow time" as displayed in Table 1, but other characteristic times might have advantage. Often it is common to avoid the flow time and not satisfy that aspect, and use the burning time as a key parameter. For example if burning scales as Q^* then it follows $l^{5/2}$, mass follows l^3 ; then the burn time scale goes as $l^{1/2}$. However, if the thickness of the fuel must obey a constraint as in using wood cribs as a fuel, the footprint of the crib would follow l^2 , but the thickness needs to follow the behavior that burn rate per unit area follows stick thickness to the (-1/2)-power. At times in scaling the firepower is formed in the model by the same fuel, but a liquid pool fire or a gas burner might also simulate it.

The next set of parameters that need consideration to get the heat loss right for the construction materials are groups Π_5 to Π_8 . However, the confluence of radiation, convection and conduction make it not possible to preserve all of these groups. Consequently something has to give. This can be radiation where the application is small fire smoke movement and detection; alternative convection can be sacrificed when the application is large fires.

To go beyond the above constraints in compartment fires, the application of suppression or structural fire behavior demand the addition of new groups. Again all of them will no be preserved and the "magic art" comes into play, but based on the common sense of science.

ADVANTAGES OF FIRE SCALING

Although scaling can never be perfectly complete there are some distinct advantages in using it. First, for a specific phenomenon, such as the average layer temperature of smoke in a room, the key dimensionless groups can be identified and then a correlation can emerge that encompasses many scales. Reference 9 discusses this aspect and the role scaling has had in establishing many formulas used in fire research. Of course this is not uncommon to other fields, in particular correlations for the heat transfer coefficient or friction factor in terms of the Reynolds number. These correlations provide formulas in complex areas where turbulent flows are impossible to model, and in fire combustion is intermingled with turbulence. So scaling gives formulas for use over a wide range of conditions.

Second, scale models that aim to emulate, like studies of geometric models in a wind tunnel, possess inherent flow physics. Turbulence is manifested in the model as it would in full-scale. There is no need for a special sub-grid model in the computer code. In addition, for fire, combustion occurs as it would in nature, soot is formed and species emerge as the flame behaves. Again, there is no need for a special sub-grid model in the fire computer code for these effects.

Third, observing a scale model be eye directly and by using enhancing visualization techniques reveals many aspects for learning, understanding and discovery. Indeed, I believe scale modeling in fire led to the concept of the zone model, or specifically a discrete layer of mixed smoke in a room fire.

Finally, the use of a scale model has the advantage of size. It is less expensive to construct and operate. And it allows ease of adding most instrumentation, and observing overall fire behavior.

EXAMPLES OF SCALE MODELING IN FIRE

Three basic applications of scaling with models will be presented. The first deals principally with the behavior of fire in an enclosure, the second addresses suppression, and the third consider the fully developed fire and the effect on steel structures. In most cases the firepower is known and can easily be modeled, but fire growth effects of thermally enhanced burning and spread and the mitigation by the reduction in oxygen will be considered too.

Enclosure Fire Dynamics

References addressed the behavior of fire emanating from the room into an adjoining corridor. The study was prompted by full-scale experiments to investigate the spread of fire from a room along the floorcovering of a corridor. The dramatic rapid spread through the corridor that could occur was no fully understood. So we had the luxury of ample funding to introduce a scale model into probing understanding. One peculiar phenomenon was fire whirl that could occur on the burning corridor floor as the flow turned to enter the room; another was the blockage of the room flow as buoyancy of the corridor floor fire became dominate. These aspects were not addressed in the scaling, but remain. In the scaling, good agreement was achieved for the temperature and velocity distributions. The scale model used gas burners in place of wood cribs. Visualization of the smoke in the upper layer showed the clear layering effect as seen in Figure 1. But by using smoke traces, the flow within the upper and lower layers was revealed to be more complex: recirculating and both turbulent and laminar, as seen in Figure 2. In addition at the right flow exit, the large eddies display the mixing of the upper layer into the lower layer.



Figure 1. Smoke layer in a corridor form a room fire



Figure 2. Recirculating layer flows

Visualization in saltwater modeling of complex smoke filling of two rooms connected by a single ceiling vent is another example of revealing effects as shown in Figure 3 [8,11]. In saltwater simulation, the Q^* is maintained through the flow rate of dyed saltwater into a fresh water model (upside down). This approach using saltwater will be amplified in a separate standalone report to NIJ.

The early motion of smoke for control in an atrium or in understanding the response or optimum placing of detectors can be other applications of scale modeling. A court case presented itself in the 1990s that involved the issue of a novel smoke control system sanctioned by the west coast building codes at the time. It consisted of a vertical intake of outside air directed upwards into an

atrium that was intended to assist the rise of smoke to the exhaust fans at the atrium roof. Instead it helped to mix and overturn the smoke layer and carried smoke throughout the building. A scale model proved this point [10]. I so dramatically di this that the opposition demanded that



Figure 3. Saltwater modeling

the courts not release the results to the other defendants, but would allow us to use the model after all the cases settled. Alas we could find no support for such research.

Another more recent study involved a complex corridor arrangement that had been used for a forensic investigation at the ATF Fire Laboratory. We were allowed to use it to study the smoke movement and scaling ability of a model. The model is shown in Figure 4 along with a sample of predictions of temperature distribution with the model for liquid pool fires.



Figure 4. Scale model of complex corridor flows

Scaling with Suppression

Several years ago a problem arose to see if water suppression could extinguish or control a large test fire established to qualify suppression systems for ferry ships in Europe. An attempt to pass the test, invented at SP (Maritime Safety Committee Circular MSC 914), failed with sprinklers. Our approach, through Vtec Laboratories, was to scale the test, and then select a variety of nozzle types, configurations, and flow rates to suppress the scaled-fire [12,13]. Once found we would scale up the nozzle configuration and flow rates and test the suppression at full-scale. The suppression scenario is combustible cargo of FM Global polystyrene cups in cardboard boxes on two covered open-bed trailer-trucks subjected to a large heptane pool fire of 3 m². We conducted a successful test, but not without difficulty. The test was done in a semi-outdoor building at near freezing weather, the facility water supply failed at the start of suppression, and the building was

nearly destroyed before the fire fighters could react. Fortunately, the rented facility allowed for another test. Figure 5 shows aspects of the scale model features. In this work, the flow rate, water droplet, pool fire, commodities, and thrust of the spray were scaled. It is not likely that a design nozzle configuration could have been found without scaling.



Figure 5. Suppression scaling showing commodity and "knock-down" in a test

Fully Developed Fire and Structures

Following 9/11, a proposal was made to study the fire and collapse of the WTC towers by scale modeling. It was common for structural engineers to use scale modeling as a tool before about 1960. Indeed, even impact on structures could be modeled, so the effect of the aircraft could be simulated. The advantage to such an approach is that it would provide data to real event where no data exists. Then any mathematical modeling tested against the scaled simulation could be validated. Moreover, the scaled experiments offer insight, repeatability, and parameter variations in tests. This was not done in the official investigation. But it prompted a NSF grant that allowed generic enclosure testing of the effect of fire on insulated loaded steel structures. Three scale compartments were studied using wood cribs to simulated large fully developed fires [15-17]. An example of these tests and results for predicting the deflection of an insulated and loaded steel frame is shown in Figure 6.



Figure 6. Scaling of structures

An aspect of the WTC fire was scaled by a student class at the University of Maryland. The students researched the fire aspects of one floor of the North Tower. They established the vented area caused by the aircraft and the fire movement, the fuel loaded (likely more correct than that of the official investigation), and made many measurements – many of which they designed and built the instrumentation. Even an insulated floor truss and external column were added by M. Wang, a PhD student. The results were in harmony with the information of the fire. Figure 7 shows some

aspects of this scaling project: the layout of the fuel load as wood cribs, the damage, the flames through vented areas on each side, and the damage to model structural elements.



Figure 7. Aspects of scaling a floor of the North WTC Tower fire

Fire Growth of a Bedroom to Flashover and Full Development

This last example is stretching the ability of scaling. We knew it would not work, but we wished to see how far the abilities of scaling could take us. It is yet to be published [20]. The hypothesis for scaling was to construct all room dimensions and overall furniture elements to geometric scaling of ¹/₄. All materials between the full-scale and model were of the same material and same thickness. This meant that in scaling a mattress, the overall object was ¹/₄, but the foam and coverings were of the same thickness in full-scale and model. It created a challenge for L. Reeves an ATF agent who likes to make his own furniture, but he was up to it. Analysis before the tests suggested that the early growth of the fire would be faster in the model, but once the smoke layer got hot (above 300 °C), radiation in the full-scale dominated and made it grow faster. However, surprisingly the phenomena of growth were the same, carbon monoxide levels comparable, and overall the results proved potentially highly useful for both design and investigation. Figure 8 shows some of these results.



Figure 8. Scale modeling of a bedroom fire

CONCLUSIONS

This paper has tried to illustrate my experience with the use of scale modeling. It is a neglected technique that could play a useful role in performance based-design. It is a tool that requires understanding of the phenomena to be scaled, it has advantages over computer modeling, and it can provide a source of insight and a validity check on mathematical forms of modeling.

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Article II. Methods of Scale Modeling in Fire

Nomenclature used in Article II and III

A_s	Area of Thermocouple Bead
A_o	Exposed Surface Area of the Wood Crib
Α	Vertical Shaft Area within the Wood Crib
С _р , С	Specific Heat
\mathcal{C}_W	Specific Heat of Boundary
С	Material Constant for Wood Crib Design
D	Diffusivity
D_c	Diameter of the Wood Crib Sticks
E	Full Scale Values (for Uncertainty)
ΔE	Full Scale Temperature Rise
Fr	Froude Number
g	Gravity
Gr	Grashof Number
Н	Enclosure Height
ΔH_c , Δh_c	Heat of Combustion
h_c	Convective Heat Transfer Coefficient
k	Conductivity of Gas
k_w	Conductivity of Boundary
l	Dimensionless Length Scale
l	Length of the Wood Crib
Μ	Scale Model Values (for Uncertainty)
ΔM	Scale Model Temperature Rise
т	Mass
ṁ	Mass Flow Rate
<i>ṁ</i> "	Mass Flow Rate per Unit Area
\dot{m}_{f}^{*}	Dimensionless Mass Flow Rate of Fuel
Nu	Nusselt Number
n	Number of Sticks in the Wood Crib
Р	Porosity
Pr	Prandtl Number
р	Pressure
<i>p</i> _R	Reference Pressure
\hat{p}	Dimensionless Pressure
<u> </u>	Heat Release Rate
Q^*	Dimensionless Heat Release Rate
$q_{R}^{"}$	Radiative Heat Flux
• •	

Reynold's Number
Spacing Between Sticks
Gas Temperature
Boundary Temperature
Ambient Temperature
Gas Temperature Difference
Time
Reference Time
Dimensionless Time
Sampling Period
Flow Velocity
Reference Flow Velocity
Dimensionless Flow Velocity
Volume
Linear Distance
Boundary Distance
Dimensionless Linear Distance
Dimensionless Boundary Distance
Species Yield
Ambient Species Yield

Greek

μ	Dynamic Viscosity
υ	Kinematic Viscosity
κ	Absorption Coefficient
Π_{i}	Dimensionless Pi Group
σ	Stephan-Boltzmann Constant
3	Emissivity
ε _g	Gas Emissivity
θ	Dimensionless Temperature
ρ	Density
$ ho_{\scriptscriptstyle W}$	Boundary Density
$ ho_{\scriptscriptstyle\infty}$	Ambient Density
$\hat{ ho}$	Dimensionless Density
τ	Dimensionless Time
$\delta_{\rm w}$	Boundary Thickness

1. Introduction

In past decades, scale models have been utilized in numerous fields. From civil to aerospace engineering, scale models demonstrate how a final product may perform. Models have been used to visualize interactions between various parts of a design, to experiment with different design ideas, and to improve the overall product [1]*. They offer engineers an additional way to analyze and understand phenomena. In order to create a representative scale model, equations applying the conservation of mass, momentum, chemical species, and energy are reviewed. Dimensionless parameters are extracted to show the relationship between the actual and the scaled model. These dimensionless groups are the key to building an accurate model. If they do not encompass the physics of the full scale prototype, the model will not accurately predict the desired phenomena. Additionally, not every dimensionless group can be satisfied simultaneously. The main idea of scale modeling is to obtain reasonable accuracy from practical approaches. This is the art of scaling [2]. Errors may arise in scale modeling because it is impossible to match all dimensionless groups. By minimizing these errors through the art of partial scaling, the scale model becomes a more accurate way to predict behavior in the full scale scenario.

Scale modeling has been applied in transportation, power, structural, material, and environmental areas. For example, wind tunnel tests are used to examine the dynamic effects of wind-induced vibrations on long-span bridges and high-rise buildings [3]. It has also been utilized in fire research. In this research, scale modeling has been used to characterize static fires in a complex geometry. Fires have the ability to cause severe damage to manmade structures, especially when contained in an enclosure. There is a growing need to further understand enclosure fire dynamics, particularly in complex geometries. Full scale research is costly when trying to simulate fires in large buildings or with extensive fuel loads. By using scale modeling, fire can be studied without the significant time or monetary cost limitations. Additionally, scale modeling can be applied where the building or enclosure no longer remains. Forensic fire investigation can benefit from scale modeling by allowing fire investigators to replicate and study the fire dynamics within a particular enclosure. Using various scaling techniques, the fire can be characterized and the effects on the surrounding enclosure can be observed.

One of the earliest examples of fire scaling in research can be traced to Rosin [4] in 1939, who used dimensionless groups to physically model domestic fireplaces. Other notable scientists have discussed the merits of scale modeling: Spalding [5] indicated the benefit of partial scaling in combustion, Williams [6] noted more than 28 independent dimensionless groups apply in combustion, and Thomas [7] emphasized that scaling in fire is an art of selecting the proper groups to characterize the fire. Enclosure fires are of particular interest to this study since the overall goal is to apply physical scaling methods for fire investigations. In particular, scale modeling can be used to reproduce burning rates, temperatures, heat fluxes, and gaseous species.

Modeling requires an in depth understanding of fire physics. Scaling a particular fire begins with assessing the governing conservation equations and

* See References for Article II and III at end of III

selecting the appropriate dimensionless groups. One major obstacle is the extent of existing information about the fire. If the fire size, or heat release rate of the fire, is known, then certain dimensionless groups can be used with ease. The engineer will scale the fire based on this heat release rate. However, many fires exhibit a growth period where the exact heat release rate as a function of time is unknown. The burning of the fuel can change with oxygen concentration, radiation effects, the area of the fire, and the specific fuel properties; all of these factors complicate how a fire is scaled [8]. An investigator would need to understand the fire dynamics involved

with their case to select the dimensionless groups that would characterize that specific fire. For example, a static fire such as a liquid fuel spill can be modeled based on the heat release rate where radiation is neglected. However, radiation plays a key role in dynamic fires where flame spread and re-radiation from the enclosure occurs, such as the flame spread across a mattress. Therefore, the dimensionless groups considered in each of these cases would vary.

This work will serve as a foundation for fire investigators to use scale modeling as a research tool in their cases and litigation efforts. First, static fires where the heat release rate is known will be examined and scaled. In this portion of the research, full scale experiments with various fuels were conducted. The full scale experiments were conducted in a wide, complex geometry at the Bureau of Alcohol, Tobacco, Firearms, and Explosives in Beltsville, MD. Partitions extended 0.6 meters down from the ceiling throughout the enclosure to create a total of twenty bays. This configuration limited the spread of hot gases and smoke since the ceiling jet was contained in the bays. The general scaling methodology presented addresses the key roles of convection in the full scale experiments. This research concentrates on the scaling of a gas burner and a liquid pool fire. These simple, static fires are used as a benchmark for developing a scaling theory that can be used for fire investigations. They also serve as controlled fuels that may be similar to fuels found at a fire scene. Full scale data has also been collected for a wood crib and blocks of polyurethane foam. In the full scale High-Bay test series at ATF, temperature, velocity, optical density, and smoke detection were measured.

Early convection played a role in this research. After the full scale experiments, a scale model was constructed based on the selected dimensionless groups. The fires were replicated in the small scale. After the static fire scale theory has been successfully applied, dynamic fires including flame spread will be explored in the same complex geometry. Finally, the scaling theory will be applied to various configurations, including an enclosure that eventually reaches flashover. Additional full scale data is available for complex geometries, such as a two-story building and a room reaching flashover. A two-story configuration would be used to develop conduction and convection models and a post-flashover room can develop radiation scaling for various scenarios. These examples may be used in future research efforts.

This research presents a review of scaling theory used in fire research. Scaling rules for design fires and enclosure material boundaries are derived and presented. Full scale testing of a gas burner, heptane pool fire, pine wood crib, and polyurethane foam is described. The scaling theory is applied to the full scale scenario and a ¹/₈ scale compartment is constructed. The gas burner and pool fires tests are run in the scale model. The accuracy of the scale model compared to the full scale data is discussed.

2. Background

The following discusses various fuels used in scale modeling and their relevance to fire investigators. Successful past examples of scaling theory are outlined and later related to the theory presented in this paper. In addition, suggestions for the fire investigator with respect to scaling are provided.

Importance of Fuel PackagesPrevious Fuels Used in Modeling

The key to scale modeling is to understand and represent the fire dynamics. This can become very complex depending on the type of fire. The size, flame height, heat flux, soot production, and growth rate of the fire are all aspects that must be considered. The interaction between the fire and the surroundings is also important. A deep concrete slab could act as a heat sink whereas a sooty wall may increase re-radiation. In general, scale modeling must try to match the heat release rate of a fire. This is done depending on the fuel, but it generally leads to similar fire behavior. The duration of the fire trial is important between the full scale and the model. Some modeling efforts keep time the same between the large and small scales while others shorten the time for the scaled model. All of these parameters are selected according to the scaling methodology and the researcher with some approximations. With so many variables to account for, scale modeling becomes an art of accurately representing the full scale fire.

Scale modeling uses various fuels to represent the specific load as accurately as possible. For example, an office building with boxes and papers may be represented using wood cribs, but this is a challenge. A gas burner could also be used, but controlling the mass flow rate of the gas becomes very important. Fires can be modeled using a wide variety of scenarios. The key to selecting the fuel is to choose fuels that are representative of the fire size and flame temperatures in the actual fire. This way, the scale model has the highest probability of reproducing the same temperatures, heat fluxes, burn patterns, etc. as the full scale [6]. Gas burners, liquid fuel, and wood cribs have all been used in successful examples of scale modeling. It is important to realize each fuel is applied to the scaling theory in a different way, but the overall concepts are the same. In the method of scaling used for this research, the dimensionless heat release rate is matched between the full scale and the model scale. A gas burner can be scaled geometrically; the size of the burner acts linearly with the scale of the model. The flow rate of the gaseous fuel is adjusted based on a dimensionless heat release rate that is calculated using the full scale data. The heat release rate of a pool of flammable liquid is dependent on both the diameter of the pool and the amount of fuel (depth). The diameter is scaled based on the dimensionless mass loss rate of the fuel and the amount of fuel is calculated using the diameter and the dimensionless heat release rate. Wood cribs are also scaled to a dimensionless heat release rate, but the primary factors that change are the spacing of the sticks, the size of the sticks, and the number of sticks. When scaling wood cribs, it is important to consider the porosity of the wood crib. If the sticks are too far apart, they will burn individually. However if they are too close together, the crib will not have proper ventilation to achieve pyrolysis of the sticks. Gross and Robertson [9] experimented with scaling wood cribs. They attempted to match the Froude number based on their recognition that the fire plume flow was governed by buoyancy.

$$Fr = \frac{u}{\sqrt{g\ell}}$$
(2.1)

The compartment was geometrically scaled, but the boundary materials remained the same between the full scale and model. This became a source of error later in the research when their results from various scales did not compare well [9]. This is a prime example of the necessity to select appropriate dimensionless groups. Proper scaling of wood cribs will be explored in future experiments. Polyurethane foam has not been widely used in scale modeling. It is important to capture the behavior of the foam and the resulting fire dynamics through the dimensionless groups.

Relevance of Fuels to Fire Investigators

Fire investigators must understand the relationship between various scaling fuels and the fuels found in their specific cases. This can be very difficult, especially at a scene where the majority of the fuel has been consumed by the fire or destroyed by fire fighting measures. Instead of trying to scale the numerous types of fuels that can be found in enclosures, it is up to the investigator to determine which reliable scaled fuels can be used to represent the fuel load. For

example, an armchair can be represented using a wood crib designed to reach a specific heat release rate. If the heat release rate of armchair is known, a wood crib can be designed based on this parameter. An investigator needs to decide which aspect of the fire is the priority in scale modeling. While a wood crib may produce a similar fire size, the products of combustion would surely be different between the foam armchair and the wood.

The scope of this project includes scaling gas burners, flammable liquid pools, wood cribs, and blocks of polyurethane foam (PU foam). A gas burner provides a standard to check the scaling theory, a liquid pool can represent an accelerant used in arson, a wood crib is similar to furniture, and PU foam promotes spread and growth comparable to a mattress or sofa. Given the tools in this project and future research, a fire investigator can apply scaling theory to recreate some fire scenarios.

Successful Examples of Scale Modeling

These examples of scale modeling describe how various researchers have applied the dimensionless groups. It is interesting to note that, while some slight differences exist; most studies have applied dimensionless groups similar to the general scaling theory. The type of fire and environmental conditions do play a role in the selection of the dimensionless groups, which generally address position, velocity, time, heat generated (or heat release rate), the boundary behavior, the fuel behavior, and radiation.

In most laboratory experiments, the heat release rate of the full scale experiment is known. In a forensic setting, the materials in the fire must be researched and modeled in order to accurately model the burning rate.

Corridor Fire Conditions

Quintiere, McCaffery, and Kashiwagi [10] scaled a corridor subjected to a room fire. The experiments consisted of a 1/7th geometric scale with a gas burner as the fuel source. Over the course of the 30 minute experiment, the heat release rate of the burner was manually adjusted to maintain the same fire behavior as in the full scale experiment. The duration of the experiment was the same in the full and model scale experiments. Manually adjusting the heat release rate ensured the fire events occurred at the same time in both size enclosures. This method is appropriate in this case because early convection is not imperative to the overall study. Overall, the gas temperature and the velocity scaled well (see Figure 2.1).



II-12

Figure 2.1: Quintiere, McCaffery, and Kashiwagi Gas Temperatures [10].

The gas velocity measured in the scale model was slightly higher than the velocity measured in the full scale experiments. When time is scaled as a burn time, time lags due to flow are not accounted for in the scale model. This had minimal impact in this experiment; however it could play a large role in tests where automatic detection and suppression are factors. The velocities depicted in Figure 2.2 are considered to agree reasonably well for a scale model.





SCALE



Figure 2.3: Quintiere, McCaffery, and Kashiwagi Surface Temperatures [10].

The fact that the fire was larger over the same period of time in the full scale experiment meant that more flame and smoke radiation affected the surrounding enclosure.

Initial Ceiling Jet

Heskestad [11] conducted a study of initial convective flow generated by a fire. The main goal of this research was to apply scaling to fire detection. Only turbulent flows were considered.

The experiments observed steady fires, where fire growth to steady-state was almost instantaneous, and quasi-steady fires, where the heat release rate varies either slowly with time or as a function of time to the nth power. Liquid pools were used to model the steady fires. Wood cribs were used to model the quasi-steady fires. The tests were conducted in two different enclosures. One enclosure had 0.31 meter partitions at the ceiling. The other configuration had a flat ceiling. Similar to Quintiere, McCaffery, and Kashiwagi [10], time remained the same between the full scale and model experiments. Heskestad developed a scaling theory where the velocity and temperature were scaled based on the heat release rate. This method is valuable since his equations can be applied to various scale sizes, along with the full scale experiment. The data collapses using these equations and direct comparisons can be made. Figure 2.4 employs Heskestad's dimensionless groups. Several measurements under diverse conditions are well correlated.



Figure 2.4: Heskestad's Dimensionless Groups Applied to Various Fires [11].

In this application, Heskestad does not include boundary scaling or radiation. In some cases, radiation and boundary scaling are an important part of the modeling process to ensure similar heat transfer in the model when compared to the full scale experiment. Other parameters that scaled well include the velocity of the gas and the concentration of various products of combustion. The ceiling contours did affect the detection response in the scale models for steady fires. It was observed that the initial smoke front arrival time in the ceiling bays were insensitive to fire size for the power-law fires.

Wood Crib in an Enclosure

Heskestad [12], and later, Croce and Xin [13], scaled wood crib fires in enclosures. In these studies, peak averages were used, so transient data was not presented. However, the burn time (defined by t_R) of the wood crib was included in the dimensionless groups pertaining to boundary scaling. The time in these experiments was the same between the full and model scales. One influential scaling factor is the porosity of the crib, which had a direct impact on the heat release rate produced. Block developed the concept of porosity using the theory of burning of densely packed wood cribs [14]. Porosity is a measure of stick spacing and placement, with an optimal value of about 0.06. The porosity can be defined as:

$$P = \left(sD_c\right)^{1/2} \frac{A_o}{A} \tag{2.2}$$

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Where s is the stick spacing, D_c is the diameter of the stick, A_o is the exposed surface area of the crib, and A is the vertical shafts area within the crib. Figure 2.5 displays the crib porosity and burning rate results from Croce's experiments.





If the sticks were too close together, the fire would not sustain on the crib as a whole due to underventilation. If the sticks were too far apart, the sticks would burn individually instead of as an entire crib. The free burning rate of wood cribs was found to be related to the porosity factor [2]:

$$\frac{m}{A_o D_c^{1/2}} = f(P)$$
(2.3)

Since ventilation factors changed the burning behavior, only scenarios with large ventilation factors were studied. The dimensionless groups used by Heskestad and Croce are outlined below. Radiation is ignored. The burn time follows $t_R \sim b^{3/2}$, where b is the thickness of the sticks in the wood crib. The general scaling laws of the research can be described by:

$$\left[\frac{T-T_{\infty}}{T_{\infty}}, Y_{i}, \frac{\dot{m}_{crib}}{\dot{m}_{free}}\right] = \left[\frac{x}{\ell}, \frac{t}{t_{burn}}, P, \frac{k_{w}t_{burn}}{\delta^{2}\rho_{w}c_{w}}, \frac{T_{\infty}k_{w}}{\dot{q}_{wall}^{*}\delta_{w}}\right]$$
(2.4)

where the temperature, species, and mass loss ratio are dependent on the position, time, crib porosity, and thermal boundary scaling. The equations used by Heskestad and Croce to scale the thermal properties of the boundaries are:

$$\frac{\ell^{15/8}}{\delta_w \rho_w c_w} = \text{constant}$$
(2.5)
$$\frac{k_w}{\delta_w \ell^{1/2}} = \text{constant}$$
(2.6)

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Heskestad and Croce's enclosure had a vent opening, which also affected the burning of the crib through ventilation. For ¹/₄ scale, ¹/₂ scale, and full scale, the dimensionless groups yielded generally well correlated results for both gas temperatures (Figure 2.6) and wall temperatures (Figure 2.7).



Figure 2.6: Heskestad and Croce's Gas Temperature Increase for Various Scales [13].



Figure 2.7: Heskestad and Croce's Wall Temperature Increase for Various Scales [13]. Good agreement between the model and full scale experiments were also recorded for concentration of oxygen, carbon monoxide, and carbon dioxide. The inaccurate scaling of the wall thickness affected some of the results.

Wood Cribs in an Enclosure II

Perricone [15] studied wood cribs in an enclosure and the response of structural elements in a fire. He conducted $\frac{1}{8}$, $\frac{1}{4}$, and $\frac{3}{8}$ scale model experiments where additional scaling of insulated loaded structural frames provided an estimate of thickness and thermal conductivity of structural fire proofing. Perricone reduced the duration of the scaled experiments by a factor of t $\sim \ell^{1/2}$ in order to accurately capture flow behavior in the experiments. This treatment of time is referred to as flow time scaling. The boundaries used by Perricone were scaled in terms of thickness, conduction, and convection. The following equations were used to scale the boundaries.

$$\Pi_{w,\delta} = \frac{\delta g^{1/4}}{\left(\frac{k}{\rho c}\right)_{w}^{1/2} \ell^{1/4}}$$

$$\Pi_{w,c} = \frac{h_{c} \ell^{2} T_{\infty}}{\rho_{\infty} c_{p} T_{\infty} \sqrt{g} \ell^{5/2}}$$

$$\Pi_{w,k} = \frac{\left(\frac{k\rho c}{t_{R}}\right)^{1/2} \ell^{2} T_{\infty}}{\rho_{\infty} c_{p} T_{\infty} \sqrt{g} \ell^{5/2}}$$

$$(2.8)$$

(2.9)

Perricone assigned $h_c \sim \ell^{1/5}$. This assumes turbulent behavior since the convective heat transfer coefficient is found through the relationship of the Nusselt number to the turbulent Reynold's number. Radiation was considered through scaling the emissivity of the gas compared to black body radiation. This played a significant role since Perricone was not only trying to replicate the fire dynamics, but also the response of the structural elements around the fire. In general, Perricone's scaling is the direct link between the characteristic time and the characteristic length scale [15]. As seen in Figure 2.8, the gas temperatures were matched very well between various scales. His success supports the use of flow time as an independent variable for the modeling of various fuels in enclosures.



Figure 2.8: Perricone's Gas Temperature Increase for Various Scales [15]. Perricone also achieved good results for the mass loss of the crib, the incident radiant heat flux to the boundaries, and the temperature of the structural elements.

Conclusions from Past Experiments

Numerous scaling methods have been successful in the past. The type of fire dictates which scaling methods will yield the most accurate results. The selection of the various dimensionless groups presented above should be considered based on the full scale situation. For example, pool fires provide a relatively constant heat flux to the enclosure, but their size affects the optimal scaling method. For smaller diameter fires, the burning rate of buoyancy-controlled turbulent pool fires is governed by natural convection rather than by radiation [16]. Therefore, these studies can be compared to the general scaling methodology for convection driven fires. However, large diameter pool fires are controlled by radiation and therefore require additional scaling of the emissivity terms. It has been shown that a burn time scaling is successful for enclosure fires where a small time lag will not affect results and radiation plays a role in the fire development. Flow time scaling is ideal when the primary mode of heat transfer is convection and the flow of gases throughout the enclosure can be considered turbulent. In the following sections, a flow time scaling method is applied to various fuels to develop an optimal scaling method of static fires for fire investigators.

Scale Modeling for the Investigator

Physical scale modeling is a science applied to a scaled structure to predict full scale parameters. It has been used to predict temperature rise, heat release rate, detection behavior, and suppression response, among others, for a large scale fire. Scaled models can utilize fire and saltwater experimental results to calculate the full scale values. In this discussion, physical fire scale modeling is described for various scenarios to display how to use basic modeling principles in a related forensic setting. Note that this discussion is limited to non-spreading enclosure fires. Additional information for modeling fire spread within an enclosure is currently being researched.

Using scale modeling to represent full scale fire behavior requires a general understanding of the variables matched between the full scale experiment and the model test. Dimensionless groups are utilized in this process. They are used to obtain variables for the full scale based on the small scale model and vice versa. It is impossible to keep all of these groups consistent between the full scale and small scale experiments. Instead, a few key groups are chosen depending upon the fire scenario. These dimensionless groups are used as a correlation between the full and model scale experiments. Values from the full scale or the model scale must be known to apply these relationships. For example, the thickness of the boundaries is scaled using the thermal properties and the characteristic length scale of the full scale and the model:

$$\frac{\delta_{\text{model}}}{\delta_{\text{full}}} = \frac{\left(\frac{k}{\rho c}\right)_{\text{model}}^{1/2} \ell_{\text{model}}^{1/4}}{\left(\frac{k}{\rho c}\right)_{\text{full}}^{1/2} \ell_{\text{full}}^{1/4}}$$
(2.10)

The conductivity, k, and the density, ρ , are material properties that vary similarly with the scale factor. The specific heat, c, remains the same since most materials have a specific heat of about 1. Therefore, using a $\frac{1}{8}$ scale, the thickness of the boundaries in the model related to the boundaries in the full scale experiment becomes the ratio of ℓ_{mod} to ℓ_{full} , or $\ell^{\frac{1}{4}}$ where ℓ is $\frac{1}{8}$:

$$\delta_{\text{model}} = 0.594 \delta_{\text{full}} \tag{2.11}$$

The principles in this example can be applied to the dimensionless Π groups when scaling as long as the assumptions are stated. The type of fire can limit the use of this general application of dimensionless groups.

Convection Driven

Convection driven fires are predominant for localized burning of a room pre-flashover [16]. The fire may be limited to a burner or piece of furniture. This includes enclosure fires leading up to flashover. Convection plays a large role in the flame height and energy release of the fire. These models are simpler to scale than radiation driven fires. However, the fuel source must be analyzed precisely to simulate full scale behavior. This can be done using various fuels in order to obtain the most accurate results. The general description of scaling can be applied to convection driven fires. This research describes convection driven fires fueled by natural gas burners and heptane liquid pools burning in an enclosure.

Radiation Driven

Radiation driven fires describe global burning of a room. This includes pre-flashover and post-flashover fires. The radiative components of fire spread include radiation from the fire to the surrounding environment, the emissivity of various components of the fire (enclosure walls, smoke layer, etc.), and re-radiation to the fire. The biggest challenge in modeling radiation driven fires is the changing the emissivity of various components, such as the smoke layer and the enclosure boundaries [8]. The emissivity, ε , is a function of the absorption coefficient, κ , of the flame, as seen in the following relationship [8]:

$$\varepsilon \approx 1 - e^{-\kappa \ell}$$

(2.12)

This relationship can be applied for the gas or flame. Convection driven scaling methods require the temperature of the fuel surface and the gas temperature to remain the same as the scale changes. However, the emissivity of the flame and smoke change with scale, making the fuel and gas temperature inconsistent with one another. This transition means radiation must be taken into

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account and the emissivity requires its own dimensionless relationship. Without knowing how to scale emissivity, the radiation effects of the fire to the surrounding environment are unknown. Even a general approach to scaling emissivity does not account for the view factor between the fuel and the surrounding boundaries. A view factor represents the proportion of radiation that leaves one surface and strikes another [17]. Realistically, the view factor between a fire and the surrounding environment is not close to one. The emissivity correlation discussed above assumes that the radiation and re-radiation between the environment and the fire is equal. In other words, the fire and the environment act as two parallel infinite plates. The fact that this does not occur in actual fires makes it even more difficult to scale emissivity in a forensic fire setting. The view factors must be represented as an additional dimensionless group when modeling radiation [18]. Therefore, fires that are driven by flame spread and radiation cannot be scaled using the general scaling relationships.

3. Scaling Methodology

The methodology for developing the scaling relationships and relevant dimensionless groups follows an analysis of the full governing equations and boundary conditions. This approach follows the basic equations as presented in Quintiere [16]. The principal features and assumptions of the scaling procedure are stated below:

- The Reynolds number (*Re*) is maintained large enough to assure turbulent flow, and terms associated with 1/Re in the equations are ignored in the body of the flow field. The requirement to insure turbulent flow in natural convection is $Re > 10^5$ [17].
- Construction of solid combustible materials will be scaled to preserve appropriate heat transfer behavior.
- Fire power and flow rates will be scaled according to gravity and time.
- The inertia and buoyant forces dictated by the Froude number are dominant and the viscous forces dictated by the Reynold's number are considered negligible. This is called Froude modeling.

Scale modeling is developed from the full governing conservation and state equations with particular attention to the initial and boundary conditions. By approximately making these equations dimensionless, the scaling relationships emerge. In general, scaling seeks to reproduce the flow field velocities, temperatures, and species in a fire at corresponding positions and times. Somehow the dynamics of the fire release of energy and species must be reproduced. The approach here is to consider the governing equations in their simplest form. The dimensionless variables are scaled in terms of fixed reference volumes and geometric scale length, ℓ . This scale factor is used in the dimensionless equations as a ratio between the full and model scale. For instance, a ¹/₈ scale model has a geometric scale length of 1 for the full scale and 8 for the model scale. The length scale to compare the full scale to the model scale is therefore ¹/₈. In the following sections, each partial differential equation is made dimensionless using this scale length to reveal the dimensionless groups, labeled as Π groups. A select number of Π groups are used in physical scale modeling depending on the scenario being modeled. For this research, the scenario includes static gas burner and liquid pool fires where the heat release rate of the fire is known. The dependent and independent dimensionless variables are listed below.

Dependent Dimensionless Variables

Temperature: $\theta = \frac{T - T_{\infty}}{T_{\infty}}$ Density: $\hat{\rho} = \frac{\rho - \rho_{\infty}}{\rho_{\infty}}$ Pressure: $\hat{p} = p / p_R$ Velocity: $\hat{u} = u / u_R$ Mass Fraction of Species *i*: Y_i

These dependent variables are the measured outputs in the scale model. They will vary based on the independent variables of time and position. The relationship between the scale model values and the full scale values of these variables depends on assumptions made when deriving the dimensionless groups. In most scale modeling experiments, the temperature is the same between scales. In other words, the temperature rise in the scale model should be comparable to the temperature rise in the full scale. The temperature distribution should also be similar, where variations in temperature occur at the same position and time as in the full scale model.

Independent Dimensionless Variables

Coordinates: $\hat{x} = \frac{x}{\ell}, \hat{y} = \frac{y}{\ell}, \hat{z} = \frac{z}{\ell}$ Time: $\hat{t} = t/t_R$

The reference time, t_R , can vary based on the time scaling used by the researcher. Flow time or burn time scaling is applied using this parameter.

Initial Conditions

Along with a consideration of the governing equations, the following initial conditions are indicated. They are the values of the ambient surroundings, e.g. the air. These conditions are applied in the balance equations in order to obtain the dimensionless groups. Previous scaling methods may have assumed other initial conditions, which would slightly change the dimensionless groups used in physical scaling. Assumptions made later include the treatment of the reference velocity and reference time.

Temperature: $T = T_{\infty}$ Density: $\rho = \rho_{\infty}$ Velocity: $u = u_{\infty}$ Pressure: $p = p_{\infty}$ Mass Fraction of Species *i*: $Y_i = Y_{i,\infty}$

The conservation equations are considered in one space dimension without a loss in generalization. Source terms are represented in a global sense.

Conservation of Mass

The conservation of mass is the simplest governing equation. It ultimately defines the time scaling [8]. Only one dimension is considered here because it is sufficient to represent this form of scaling.

$$\frac{\partial \rho}{\partial t} + \frac{\partial}{\partial x}(\rho u) = 0 \tag{3.1}$$

Make the equation dimensionless by selected normalizing factors: reference time t_R , reference velocity u_R , and characteristics length scale ℓ . ℓ represents the length, width, and height. Since geometric similarity is maintained, ℓ is proportional to each dimension.

$$\frac{\rho_{\infty}}{t_R}\frac{\partial\hat{\rho}}{\partial\hat{t}} + \frac{u_R\rho_{\infty}}{\ell}\frac{\partial\hat{\rho}\hat{u}}{\partial\hat{x}} = 0$$
(3.2)

Rearranging,

$$\frac{\partial \hat{\rho}}{\partial \hat{t}} + \Pi_1 \frac{\partial \hat{\rho} \hat{u}}{\partial \hat{x}} = 0$$
(3.3)

As all terms are dimensionless, the first dimensionless group emerges. Π_1 relates the velocity, time, and geometric length scale. It is the ratio of time to the fluid flow time. This Π group determines the treatment of the reference time. One way to select a reference time is to force: [8]

$$\Pi_{1} = \frac{u_{R}t_{R}}{\ell} = 1 \quad or$$

$$t_{R} = \frac{\ell}{u_{R}}$$
(3.4)
(3.5)

then t_R has the physical meaning of a flow time. This means it is the time required for a fluid particle to travel ℓ . This selection insures that scaled time matches the flow time. Events at a corresponding scaled position will occur at the same scaled time. However, other plausible reference times could be selected. This is the art and style of scale modeling. Such choices lead to indications of the importance of phenomena and whether such phenomena might be dominant or negligible.

The relationship defined by the flow time is explored further using the reference velocity. This simple derivation plays a key role in physical modeling where the scaled time is reduced and fire events occur earlier in the scale model relative to the full scale experiments. To compare the model results to the full scale experiments, the time is converted back to full scale time and plotted against the original data.

Conservation of Momentum

Conservation of momentum applied to include the effects of gravity yields relationships pertaining to velocity and pressure [8]. The conservation of momentum is:

$$\rho\left(\frac{\partial u}{\partial t} + u\frac{\partial u}{\partial x}\right) = \mu \frac{\partial^2 u}{\partial x^2} - \frac{\partial p}{\partial x} - \rho g$$
(3.6)

The partial differential equation is made dimensionless by utilizing the normalizing parameters mentioned earlier.
$$\frac{\rho_{\infty}u_{R}}{t_{R}}\hat{\rho}\frac{\partial\hat{u}}{\partial\hat{t}} + \frac{\rho_{\infty}u_{R}^{2}}{\ell}\hat{\rho}\hat{u}\frac{\partial\hat{u}}{\partial\hat{x}} = \frac{u_{R}}{\ell^{2}}\mu\frac{\partial^{2}\hat{u}}{\partial\hat{x}^{2}} - \frac{p_{R}}{\ell}\frac{\partial\hat{p}}{\partial\hat{x}} - \rho_{\infty}g\hat{\rho}$$
(3.7)

After dividing each term by $\frac{\rho_{\infty}u_R^2}{\ell}$, the dimensionless Π groups can be extracted. Note that Π_1 is seen again in the conservation of momentum.

$$\frac{\ell}{u_R t_R} \frac{\partial \hat{u}}{\partial \hat{t}} + \hat{\rho} \hat{u} \frac{\partial \hat{u}}{\partial \hat{x}} = \frac{\mu}{\rho_\infty u_R \ell} \frac{\partial^2 \hat{u}}{\partial \hat{x}^2} - \frac{p_R}{\rho_\infty u_R^2} \frac{\partial \hat{p}}{\partial \hat{x}} - \frac{g\ell}{u_R^2} \hat{\rho}$$

$$\frac{1}{\Pi_1} \frac{\partial \hat{u}}{\partial \hat{t}} + \hat{\rho} \hat{u} \frac{\partial \hat{u}}{\partial \hat{x}} = \frac{1}{\Pi_2} \frac{\partial^2 \hat{u}}{\partial \hat{x}^2} - \Pi_3 \frac{\partial \hat{p}}{\partial \hat{x}} - \frac{1}{\Pi_4} \hat{\rho}$$
(3.8)
$$(3.9)$$

Where the dimensionless groups are defined and named as follows:

$$\Pi_{2} = \frac{\rho_{\infty} u_{R} \ell}{\mu}, \text{ Reynolds Number, Re } \sim \left(\frac{\text{momentum}}{\text{viscous force}}\right)$$
(3.10)

$$\Pi_{3} = \frac{p_{R}}{\rho_{\infty} u_{R}^{2}}, \text{ Euler Number, Eu } \sim \left(\frac{\text{pressure force}}{\text{momentum}}\right)$$
(3.11)
Froude Number, Fr $\sim \left(\frac{\text{momentum}}{\text{gravity force}}\right)$ (3.12)

All of these groups cannot normally be matched between the model and full scale. For example, if Π_2 and Π_4 are sought to be preserved between a model and prototype in air under normal gravity, the Π_2 requires that $u_R \sim \ell^{-1}$ and Π_4 requires $u_R \sim \ell^{1/2}$. Both cannot be done. However, Π_2 governs turbulence and a criterion for turbulent flow is $Re > 10^5$ [17]. The turbulence in the scale model should be verified once the characteristic scale length has been selected. The dimensionless group Π_3 is the Euler number [19]. It relates the pressure in the scale model to the kinetic energy per volume. It is used to characterize losses in the flow. A perfect frictionless flow is achieved when the Euler number is equal to 1. Allowing $\Pi_3 = 1$ yields $p_R = \rho_\infty u_R^2$ giving a way to select p_R in terms of the reference velocity. Π_4 is the dimensionless Froude number. A reference velocity can be selected by allowing $\Pi_4=1$. This is done when there is no clear prescribed velocity for reference and the flow is induced by gravity. This Π is proportional to the ratio of momentum to gravity forces. If $\Pi_4=1$, then $u_R = \sqrt{g\ell}$. This resulting relationship is appropriate for buoyancy driven flows. The selected reference velocity also impacts the choice of t_R as the of flow time from Π_1 . Substituting for u_R , the reference time is now:

$$t_{R} = \frac{\ell}{u_{R}}$$

$$t_{R} = \frac{\ell}{\sqrt{g\ell}} = \sqrt{\frac{\ell}{g}}$$

$$(3.13)$$

$$(3.14)$$

This resource was prepared by the author(s) using Federal funds provided by the U.S. Department of Justice. Opinions or points of view expressed are those of the author(s) and do not necessarily reflect the official position or policies of the U.S. Department of Justice Therefore, time is related to scale as $t_{R} \sim \ell^{1/2}$. This means that the scale model experimental time is calculated by reducing the fullscale experimental time:

$$\frac{t_{\text{mod }el}}{t_{full}} = \left(\frac{\ell_{\text{mod }el}}{\ell_{full}}\right)^{1/2}$$
(3.15)

This selection of u_R could change based on the full scale scenario. If there is a wind present with a specific velocity u_{∞} , then u_{∞} could be selected for u_R . The Froude number will then remain as

$$\Pi_4 = \frac{u_R^2}{g\ell} = \frac{u_\infty^2}{g\ell}$$

Conservation of Energy

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Conservation of energy yields the dimensionless groups pertaining to firepower and fuel behavior [8]. The governing equation and its boundary conditions account for conduction, convection, and radiation.

$$\rho c_p \left(\frac{dT}{dt} + u \frac{dT}{dx} \right) = k \frac{\partial^2 T}{\partial x^2} - \frac{\partial \dot{q}_R^{"}}{\partial x} + \frac{\partial p}{\partial t} + \dot{m}^{"} \Delta h_c$$
(3.16)

The source term on the right hand side is dimensionally represented in terms of the overall heat release rate, \dot{Q} . The overall heat release rate is convenient since it is commonly known in some experiments. In some cases the heat release depends on the phenomena of fire growth where the mass loss rate depends on external factors, i.e. oxygen concentration, radiation, fuel, etc. [8]. As the volume can be represented using ℓ^3 since the enclosure dimension is geometrically scaled, the heat generation term can be represented as:

$$\dot{m}^{"}\Delta h_c = \dot{q}_{comb}^{"} \sim \frac{Q}{\ell^3} \qquad (3.17)$$

Additionally, the radiation transfer term is represented in terms of the absorption coefficient of the fluid media, κ :

$$\dot{q}_R^{"} \sim \kappa \sigma T^4 \tag{3.18}$$

The dimensionless conservation of energy equation is easily manipulated by dividing each term by $\frac{\rho_{\infty}T_{\infty}u_R}{\ell}$ to reveal additional dimensionless groups.

$$\rho_{\infty}\hat{\rho}c_{p}\left(\frac{T_{\infty}}{t_{R}}\frac{d\hat{T}}{d\hat{t}} + \frac{u_{R}T_{\infty}}{\ell}\hat{u}\frac{d\hat{T}}{d\hat{x}}\right) = \frac{T_{\infty}}{\ell^{2}}k\frac{\partial^{2}\hat{T}}{\partial\hat{x}^{2}} - \kappa\sigma T_{\infty}^{4}\hat{T}^{4} + \frac{p_{R}}{c_{p}t_{R}}\frac{\partial\hat{p}}{\partial\hat{t}} + \frac{\dot{Q}}{\ell^{3}c_{p}}$$
(3.19)

$$\hat{\rho}\left(\frac{\ell}{u_R t_R}\frac{d\hat{T}}{d\hat{t}} + \hat{u}\frac{d\hat{T}}{d\hat{x}}\right) = \frac{k}{\rho_{\infty}c_p\ell u_R}\frac{\partial^2 \hat{T}}{\partial \hat{x}^2} - \frac{\kappa\sigma T_{\infty}^4 \hat{T}^4 \ell}{\rho_{\infty}T_{\infty}u_R c_p} + \frac{p_R\ell}{c_p t_R \rho_{\infty}T_{\infty}u_R}\frac{\partial \hat{p}}{\partial \hat{t}} + \frac{\dot{Q}}{\ell^2 T_{\infty}\rho_{\infty}c_p u_R} (3.20)$$

The Π groups are once again extracted from the dimensionless equation.

$$\hat{\rho}\left(\frac{1}{\Pi_1}\frac{d\hat{T}}{d\hat{t}} + \hat{u}\frac{d\hat{T}}{d\hat{x}}\right) = \frac{1}{\Pi_5}\frac{1}{\mathrm{Re}}\frac{\partial^2\hat{T}}{\partial\hat{x}^2} - \Pi_6\Pi_7\hat{T}^4 + \Pi_8\frac{\partial\hat{P}}{\partial\hat{t}} + \Pi_9$$
(3.21)

$$\Pi_5 = \Pr = \frac{\mu c_p}{k} = \frac{\upsilon}{\alpha}, \left(\frac{\text{viscous diffusion}}{\text{thermal diffusion}}\right)$$
(3.22)

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Let
$$t_{R} = \ell/u_{R}$$
, $u_{R} = (g\ell)^{1/2}$:

$$\Pi_{6} = \left[\frac{\sigma T_{\infty}^{3}}{\rho_{\infty}c_{p}\sqrt{g\ell}}\right], \left(\frac{\text{radiant energy}}{\text{thermal flow energy}}\right)$$
(3.23)

$$\Pi_{7} = [\kappa \ell], \left(\frac{\text{physical length}}{\text{radiant absorption depth}}\right)$$
(3.24)

$$\Pi_8 = \frac{g\ell}{c_p T_{\infty}}, \left(\frac{\text{potential energy}}{\text{enthalpy}}\right)$$
(3.25)

$$\Pi_{9} = Q^{*} = \frac{\dot{Q}}{\rho_{\infty}c_{p}T_{\infty}\sqrt{g}\ell^{5/2}}, \left(\frac{\text{firepower}}{\text{enthalpy rate}}\right)$$
(3.26)

Here Π_5 , the Prandtl number, is included through the scaling of the convective heat transfer coefficient. Groups Π_6 and Π_7 pertain to the radiative heat losses in the fire and the enclosure. Π_8 defines an important linear relationship between the temperature and the dimensionless length scale. Π_9 is called the Zukoski number [8]. It describes the dimensionless heat release rate. This group is used in all fuel scaling to match the full scale Q* to the model Q*. Since the acceleration of gravity does not change with scale and $g \sim \ell^0$, the heat release rate is scaled as $\ell^{5/2}$ [2].

Conservation of Species

Fuel behavior is defined using conservation of species. In particular, the diffusivity and the fuel flow are defined.

$$\rho\left(\frac{\partial Y_i}{\partial t} + u\frac{\partial Y_i}{\partial x}\right) = \rho D \frac{\partial^2 Y_i}{\partial x^2} + \dot{m}_i^{"} y_i$$
(3.27)

When the governing equation is made dimensionless, mass flux is selected as the mass flux of the fuel.

$$\frac{\rho}{t_R}\frac{\partial Y_i}{\partial \hat{t}} + \frac{\hat{u}u_R}{\ell}\frac{\partial Y_i}{\partial \hat{x}} = \frac{\rho_{\infty}D}{\ell^2}\frac{\partial^2 Y_i}{\partial \hat{x}^2} + \frac{\dot{m}_{fuel,\sup}y_i}{\ell^3}$$
(3.28)

After each term is divided by (u_R/ℓ) , the dimensionless groups are extracted.

$$\frac{\rho_{\infty}\ell}{\rho_{\infty}u_{R}t_{R}}\frac{\partial Y_{i}}{\partial \hat{t}} + \hat{u}\frac{\partial Y_{i}}{\partial \hat{x}} = \frac{\mu\rho_{\infty}D\ell}{\mu\ell^{2}u_{R}\rho_{\infty}}\frac{\partial^{2}Y_{i}}{\partial \hat{x}^{2}} + \frac{\ell}{u_{R}}\frac{\dot{m}_{juel,sup}y_{i}}{\ell^{3}}$$
(3.29)

$$\frac{1}{\Pi_1} \left[\frac{\partial Y_i}{\partial t} + u \frac{\partial Y_i}{\partial x} \right] = \frac{1}{\Pi_2} \frac{1}{\Pi_{10}} \frac{\partial^2 Y_i}{\partial x^2} + \Pi_{11} y_i$$
(3.30)

$$\Pi_{10} = \frac{\nu}{D} = \frac{\mu}{\rho_{\infty}D}, \left(\frac{\text{viscous}}{\text{diffusion}}\right)$$
(3.31)

$$\Pi_{11} = \dot{m}_{F, \text{sup}}^* = \frac{\dot{m}_{F, \text{sup}}}{\rho_{\infty} \sqrt{g} \ell^{5/2}}, \left(\frac{\text{fuel flow}}{\text{advection}}\right)$$
(3.32)

Note that y_i is a yield given as mass of species to mass of fuel gases released or supplied. These groups will play a significant role for fuels with unsteady burning characteristics. The fuel

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characteristics of steady fires with a known heat release rate are included in the scaling of the heat release rate found in Π_9 .

Boundary Conditions

The thermal boundary conditions are used to determine the heat transfer scaling and the selections of the construction materials. The following Π groups apply to the boundary conditions of the enclosure. The thickness, δ , and thermal properties of the enclosure walls may be modified so the heat exchange rate between the hot gas layer and the boundaries scales directly with the convective heat transfer in the enclosure. This analysis begins with a heat transfer equation describing conduction and convection. The heat conduction equation governs the boundary construction material.

$$\rho_{w}c_{w}\frac{\partial T}{\partial t} = k_{w}\frac{\partial^{2}T}{\partial x_{w}^{2}}$$
(3.33)

The equation is now made dimensionless.

$$\frac{T_{\infty}\rho c}{t_R}\frac{\partial \hat{T}}{\partial \hat{t}} = \frac{T_{\infty}k}{\delta_w^2}\frac{\partial^2 \hat{T}}{\partial \hat{x}_w^2}$$
(3.34)

where

$$\hat{x}_w = \frac{x_w}{\delta_w} \tag{3.35}$$

The coefficient on the right hand side can be combined with the left hand side in order to extract the dimensionless group.

$$\frac{\rho_{w}c_{w}\delta_{w}^{2}}{k_{w}t_{R}}\frac{\partial\hat{T}}{\partial\hat{t}} = \frac{\partial^{2}\hat{T}}{\partial\hat{x}_{w}^{2}}$$

$$\Pi_{12} = \frac{\rho_{w}c_{w}\delta_{w}^{2}}{k_{w}t_{R}}, \left(\frac{\text{thickness}}{\text{thermal length}}\right)$$
(3.36)
(3.37)

The heat transfer to the construction material is based on the thermal length and reference time. This is a representative analysis used to obtain the dimensionless groups. The next equation balances conduction with convection, radiation from the gas, and radiation from the surface. At the solid surface (x=0):

$$-k_{w}\frac{\partial T}{\partial x_{w}} = h_{c}(T - T_{w}) + \varepsilon_{g}\sigma T^{4} - \varepsilon\sigma T_{w}^{4}$$
(3.38)

This equation is now made dimensionless.

$$-\frac{k_{w}}{\delta_{w}}\frac{\partial T}{\partial \hat{x}_{w}} = h_{c}(\hat{T} - \hat{T}_{w}) + T_{\infty}^{3}\varepsilon_{g}\sigma\hat{T}^{4} - T_{\infty}^{3}\varepsilon\sigma\hat{T}_{w}^{4}$$
(3.39)

The ratio of conductivity to thermal thickness is moved to the right hand side of the equation.

$$\frac{\partial \hat{T}}{\partial \hat{x}_{w}} = \frac{h_{c}\delta_{w}}{k_{w}}(\hat{T} - \hat{T}_{w}) + \frac{T_{w}^{3}\varepsilon_{g}\sigma\delta_{w}}{k}\hat{T}^{4} - \frac{T_{w}^{3}\varepsilon_{g}\sigma\delta_{w}}{k_{w}}\hat{T}_{w}^{4}$$
(3.40)

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$$\frac{\partial \hat{T}}{\partial \hat{x}_{w}} = \Pi_{13}(\hat{T} - \hat{T}_{w}) + \Pi_{15}\hat{T}^{4} - \Pi_{14}\hat{T}_{w}^{4}$$
(3.41)

Convection and radiation are described at the solid surface. The convective heat transfer coefficient changes throughout the experiment. Therefore, it is related to the Reynold's number and Prandtl number by the following [17]:

$$Nu = \frac{h_c \ell}{k} \sim \operatorname{Re}^{4/5} \operatorname{Pr}^{1/3} \quad \text{for turbulent flow}$$

$$\operatorname{Re}^{4/5} \sim \left(\frac{\sqrt{g\ell}\ell}{\upsilon}\right)$$

$$(3.42)$$

$$(3.43)$$

$$h_c \sim \left(\frac{\left(\ell^{3/2}\right)^{4/5}}{\ell}\right) \sim \ell^{1/5} \tag{3.44}$$

Using this relationship for the convective heat transfer coefficient, substitutions can be made in Π_{13} .

$$\Pi_{13} = \frac{h_c \delta_w}{k_w} \sim \left(\frac{g^{2/5} k_g}{\upsilon^{4/5}}\right) \frac{\ell^{1/5} \delta_w}{k_w}, \left(\frac{\text{convection}}{\text{conduction}}\right)$$
(3.45)
$$\Pi_{14} = \frac{\varepsilon \sigma T_{\infty}^3 \delta_w}{k_w}, \quad \left(\frac{\text{surface radiation}}{\text{conduction}}\right)$$
(3.46)

 Π_{13} relates the thermal properties of the boundaries, particularly conductivity. Since conductivity and density are scaled similarly, this can also be used to calculate the density in the scale model boundaries [8]. Π_{14} and Π_{15} address the radiation from the gas and boundary surfaces. As mentioned earlier, accurately scaling the emissivity poses a significant challenge as the fire changes throughout the experiment.

Note that the Π group pertaining to the gas radiation scaling is not included. $\Pi_{15} = \frac{\varepsilon_g \sigma T_{\infty}^3 \delta}{k}$ (3.47)

This group addresses the emissivity of the gas, ε_g . This value can be represented by the following relationship:

$$\varepsilon_g \sim 1 - e^{-\kappa \ell} \tag{3.48}$$

 Π_7 describes $\kappa \ell$, therefore Π_{15} does not need to be included.

The Dimensionless Groups

~

This section describes the dimensionless groups derived from the governing conservation equations in Section 3.5. The following Π groups rely on the general assumptions and initial conditions stated above with regard to time, velocity, and pressure. These groups form the basis of this scaling method.

$$\Pi_1 = \frac{u_R t_R}{\ell} \tag{3.49}$$

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$$\Pi_2 = \operatorname{Re} = \frac{\rho_{\infty} u_R \ell}{\mu}, \left(\frac{\operatorname{inertia}}{\operatorname{viscous}}\right)$$
(3.50)

$$\Pi_3 = Euler = \frac{p_R}{\rho_\infty u_R^2}$$
(3.51)

$$\Pi_4 = Fr = \frac{u_R^2}{g\ell} \tag{3.52}$$

Some groups in Π_5 through Π_{14} include assumptions for the reference time and velocity. These are

based on
$$\Pi_1=1$$
 (flow time) and $\Pi_4=1$ (gravity speed).
 $\Pi_5 = \Pr = \frac{\mu c_p}{k}, \left(\frac{\text{viscous diffusion}}{\text{thermal diffusion}}\right)$
(3.53)

The following Π groups apply to the fire source behavior and interactions with the surrounding enclosure. Relationships to the geometric scale length are also provided. These would only be used if the dimensionless groups are included in the scaling methodology.

$$\Pi_{6} = \left\lfloor \frac{\sigma T_{\infty}^{3}}{\rho_{\infty} c_{p} \sqrt{g\ell}} \right\rfloor, \left(\frac{\text{radiant energy}}{\text{thermal flow energy}} \right) \quad T_{\infty} \sim \ell^{1/6}$$
(3.54)

$$\Pi_{7} = \kappa \ell, \quad \left(\frac{\text{physical length}}{\text{radiant absorption depth}}\right) \quad \kappa \sim \ell^{-1}$$
(3.55)

$$\Pi_8 = \frac{g\ell}{c_p T_{\infty}}, \quad \left(\frac{\text{potential energy}}{\text{enthalpy}}\right)$$
(3.56)

$$\Pi_{9} = Q^{*} = \frac{\dot{Q}}{\rho_{\infty}c_{p}T_{\infty}\sqrt{g}\ell^{5/2}}, \left(\frac{\text{firepower}}{\text{enthalpy rate}}\right) \dot{Q} \sim \ell^{5/2}$$
(3.57)

$$\Pi_{10} = \frac{\upsilon}{D} = \frac{\mu}{\rho_{\infty}D}, \left(\frac{\text{viscous}}{\text{diffusion}}\right)$$
(3.58)

$$\Pi_{11} = \dot{m}_{F, \text{sup}}^* = \frac{y_i \dot{m}_{F, \text{sup}}}{\rho_{\infty} \sqrt{g} \ell^{5/2}}, \left(\frac{\text{fuel flow}}{\text{advection}}\right)$$
(3.59)

$$\Pi_{12} = \frac{\delta_w^2}{(k/\rho c)_w (\ell/g)^{1/2}}, \left(\frac{\text{thickness}}{\text{thermal length}}\right) \quad \delta_w \sim \ell^{1/4}$$
(3.60)

The thickness of the thermal boundaries is related to the geometric length scale by assuming that the density and conductivity of the boundaries scale similarly. It also assumes the specific heat does not change between scales. This results in $\delta_w^2/\ell^{1/2}$, so

 $\delta_{w} {\sim} \ell^{1/4}.$ Proportionalities for Π_{13} also rely on these assumptions.

$$\Pi_{13} = \frac{h_c \delta_w}{k} \sim \left(\frac{g^{2/5} k_g}{\nu^{4/5}}\right) \frac{\ell^{1/5} \delta_w}{k_w}, \left(\frac{\text{convection}}{\text{conduction}}\right) \quad k \sim \ell^{9/20}, \rho \sim \ell^{9/20}$$
(3.61)
$$\Pi_{14} = \frac{\varepsilon \sigma T_{\infty}^3 \delta}{k}, \left(\frac{\text{surface radiation}}{\text{conduction}}\right)$$
(3.62)

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Strategy of Partial Scaling

Complete scaling would preserve all fourteen dimensionless groups simultaneously. This is not possible. For example, the preservation of the Froude number Π_4 , which describes buoyancy of the plume flow, and the Reynold's number Π_2 , which describes turbulence of the plume flow, is difficult to implement.

$$\Pi_4 = F_r = \frac{u_R^2}{g\ell} = 1 \tag{3.63}$$

When the Froude number is one, the reference velocity becomes equal to the square root of gravity times the length scale. Substituting the reference velocity into the Reynold's number yields:

$$\Pi_2 = \operatorname{Re} = \frac{\rho_{\infty} u_R \ell}{\mu} = \frac{\rho_{\infty} \sqrt{g \, \ell^{3/2}}}{\mu}$$
(3.64)

This relationship suggests that Re is proportional to $\ell^{3/2}$. In other words, by changing the length scale, ℓ , the turbulence in the bulk flow also changes. The partial differential equations used to derive the dimensionless groups assume that the Reynold's number is large. Π_2 is not included in the scaling theory since Re > Re_{turb}=10⁵ is optimal. The Froude number is used and the Reynold's number is assumed to be turbulent in both the full and scale models [8]. This is one example of the application of partial scaling. A scale model serves to preserve certain factors in a full scale prototype so a reasonable level of accuracy is achieved. Partial scaling uses the Π groups that describe dominant fire effects and omits Π groups that are negligible under typical conditions [2]. A deep understanding of fire phenomena and the particular fire scenario is important when selecting the appropriate dimensionless groups.

The full scale experiments describe static fires with known heat release rates. Static fires reach an almost immediate steady-state while dynamic fires have a growth period before reaching steady state. Convection and conduction play an important role due to the large vents in the enclosure and the numerous partitions at the ceiling. Radiation is present, but it is not explicitly scaled due to the difficulty of accurately representing the changing emissivity in the corridor. Instead, the radiation present in the full scale fires is assumed to be accounted for through the boundary and design fire scaling. Since early convection is a factor, time is scaled using a flow time approach. The Reynold's number is assumed to be turbulent in the scaled model since the full scale experiment exhibited turbulent flow behavior. This assumption allows the Froude number to describe the buoyancy of the flow in the scale model. As a result of these experimental characteristics, this research employs Π_1 , Π_4 , Π_9 , Π_{11} , Π_{12} , and Π_{13} . The dimensionless groups can be related to the dependent dimensionless parameters.

$$\left[\frac{u}{u_{R}}, \frac{T - T_{\infty}}{T_{\infty}}, \frac{p}{P_{R}}, Y_{i}\right] = f\left[\frac{x}{\ell}, \frac{t}{t_{R}}, \Pi_{1}, \Pi_{4}, \Pi_{9}, \Pi_{11}, \Pi_{12}, \Pi_{13}\right]$$
(3.65)

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$$\Pi_1 = \frac{u_R t_R}{\ell} \sim 1 \text{ specified so that } t_R = \frac{\ell}{u_R} = \sqrt{\frac{\ell}{g}}, \quad t \sim \ell^{1/2} \text{ (gravity scaling)}$$
(3.66)

$$\Pi_4 = Fr = \frac{u_R^2}{g\ell} = 1 \quad \text{specified so that} \quad u_R = \sqrt{g\ell}, \quad u \sim \ell^{1/2}$$
(3.67)

$$\Pi_{9} = Q^{*} = \frac{\dot{Q}}{\rho_{\infty}c_{p}T_{\infty}\sqrt{g}\ell^{5/2}}, \left(\frac{\text{firepower}}{\text{enthalpy rate}}\right) \dot{Q} \sim \ell^{5/2}$$
(3.68)

$$\Pi_{11} = \dot{m}_{F, \text{sup}}^* = \frac{y_i \dot{m}_{F, \text{sup}}}{\rho_{\infty} \sqrt{g} \ell^{5/2}}, \left(\frac{\text{fuel flow}}{\text{advection}}\right)$$
(3.69)

$$\Pi_{12} = \frac{\delta_w^2}{(k/\rho c)_w (\ell/g)^{1/2}}, \left(\frac{\text{thickness}}{\text{thermal length}}\right) \delta_w \sim \ell^{1/4}$$
(3.70)

$$\Pi_{13} = \frac{h_c \delta_w}{k} \sim \left(\frac{g^{2/5} k_g}{\nu^{4/5}}\right) \frac{\ell^{1/5} \delta_w}{k_w}, \left(\frac{\text{convection}}{\text{conduction}}\right) \quad k_w \sim \ell^{9/20}, \rho_w \sim \ell^{9/20}$$
(3.71)

Other scenarios for localized burning in a convection driven cases may use other groups presented here or other scaling methodologies. Fire exhibiting significant growth, spread, or flashover should account for radiation in the scale model. The selection of dimensionless groups is left up to the researcher's understanding of the fire scenario to be modeled. Further guidance is given in Section 2.3. Table 3.1 lists the calculated dimensionless groups for the full scale and scale model used in this research. This is done to visualize the representation of the full scale experiments. Note that for some Π groups, there is a large difference between scales since not all were included in the partial scaling. The groups that are equal to one are based on the assumptions made during the derivation. Π_9 and Π_{10} change with fire size. These values have been calculated and matched between scales, as seen in Section 4.2. Π_{12} , Π_{13} , Π_{14} and are calculated for the partitions at the ceiling. These calculations were also performed for the walls and floor in the enclosure. The fire investigator should see that not every group needs to match in order to build a representative scale model. Groups with large values that are located in the denominator in the derivations, such as Π_2 , and with small values that are located in the numerator, such as Π_{10} , can be ignored.

Dimensionless Group	Full Scale Value	¹ / ₈ Scale Value
$\Pi_1 = \frac{u_R t_R}{\ell}$	1	1
$\Pi_2 = \operatorname{Re} = \frac{\rho_{\infty} u_R \ell}{\mu}$	217144	9596
$\Pi_3 = Euler = \frac{p_R}{\rho_\infty u_R^2}$	1	1
$\Pi_4 = Fr = \frac{u_R^2}{g\ell}$	1	1

 Table 3.1: Calculated Dimensionless Groups

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$\Pi_5 = \Pr = \frac{\mu c_p}{k}$	0.721	0.721
$\Pi_{6} = \left[\frac{\sigma T_{\infty}^{3}}{\rho_{\infty}c_{p}\sqrt{g\ell}}\right]$	0.00043	0.0011
$\Pi_7 = \kappa \ell$	0.8	0.1
$\Pi_8 = \frac{g\ell}{c_p T_{\infty}}$	0.0326	0.2676
$\Pi_9 = Q^* = \frac{\dot{Q}}{\rho_{\infty}c_p T_{\infty} \sqrt{g} \ell^{5/2}}$	Varies	Varies
$\Pi_{10} = \frac{\upsilon}{D} = \frac{\mu}{\rho_{\infty}D}$	0.0000154	0.0000144
$\Pi_{11} = \dot{m}_{F, \text{sup}}^* = \frac{y_i \dot{m}_{F, \text{sup}}}{\rho_{\infty} \sqrt{g} \ell^{5/2}}$	Varies	Varies
$\Pi_{12} = \frac{\delta_w^2}{(k/\rho c)_w (\ell/g)^{1/2}}$	2154	1632
$\Pi_{13} = \left(\frac{g^{2/5}k_g}{v^{4/5}}\right) \frac{\ell^{1/5}\delta_w}{k_w}$	323	268
$\Pi_{14} = \frac{\varepsilon \sigma T_{\infty}^3 \delta}{k}$	105	122

Harmonization with Past Research

The examples of scale modeling discussed in Section 2 can be related to this derived scaling theory. It is important to keep in mind that various assumptions made by the researcher can change the dimensionless groups. There is no single way to scale a fire. The main difference between the scaling method used by Quintiere, McCaffery, and Kashiwagi and the method presented here is that time is scaled as a burn time instead of a flow time, meaning events occurred at the same time in the scale model and in the full scale [10]. The other Π groups remained the same. Heskestad [11] also used burn time scaling. Additionally, his dimensionless groups were modified by a factor of the dimensionless heat release rate. This is valuable since it allows data from various size scales to be compared. The data collapses due to the Q* parameter that is included in each Π group. Heskestad did not include boundary scaling or radiation scaling in his derivation. The research completed by Heskestad [12] and later Croce and Xin [13] also employs the use of burn time scaling. This burn time follows $t_R \sim b^{3/2}$ where b is the thickness of the sticks in the wood crib. Heskestad and Croce also decided to scale the convective heat transfer coefficient differently in their boundary condition scaling. The relationship between convection and enthalpy is

scaled as $h_c \sim \ell^{1/2}$. This is a direct result of a laminar flow behavior instead of a turbulent flow behavior in the experiments. The heat release rate and fuel behavior were scaled similarly to the methodology presented above. Finally, Perricone [15] conducted experiments using a scaling theory that is most similar to the general scaling theory. The most noteworthy similarity is the use of a scaled flow time instead of a burn time. Scaling the velocity, heat release rate, and fuel behavior are also the same as the Π groups above. Table 3.1 compares the experiments and dimensionless groups of the past research to the scenario present in this research.

While seemingly varied, the basis of scale modeling in each of these examples is the same basis formed in this research. The dominant dimensionless groups remain generally the same for different fire scenarios. While some research chose to omit boundary scaling or radiation scaling, the foundation of scale modeling can be seen through past research. Time and flow velocity assumptions are left to the researcher. The Froude number is used to characterize buoyancy while the Reynold's number is assumed to remain turbulent. When the heat release rate is known, the Zukoski number is matched between the full and small scales. The boundaries are scaled with respect to convection and conduction. Sometimes radiation is included, but it requires a more complex dimensionless group to represent the emissivity. With such a wide range of research based on these simplified scaling laws, a fire investigator would most likely find scale modeling very useful.

Researcher Design Fire/ Set-up Time Differing Π Groups Comments • Time Lags Not Included Gas Burner in a Room with Burn • Radiation Groups Ignored Adjacent Corridor All Groups Remain the Quintiere, Time • Good Results for Convective Processes McCaffery, Scaling Same Except Π_1 : $(T_{gas}, u_{gas}, \dot{q}_{conv})$ 1/7 Scale and Kashiwagi $t_{model} = t_{full scale}$ • T_{surface}: model< full scale 30 Minute t~ℓ⁰ [10] • \dot{q} : model< full scale 300 to 1500 kW Fires Duration • Flame Heights Not Scaled (Includes Q* Factor) Initial Convective Flow $\Pi_{time} = \frac{t_R}{O^{*^{1/3}}} = \frac{\Pi_1}{O^{*^{1/3}}}$ Attempt to Apply Scaling • No Boundary or Radiation Scaling to Fire Detection $\Pi_{velocity} = \frac{\frac{u}{\sqrt{g\ell}}}{Q^{*1/3}} = \frac{\Pi_4}{Q^{*1/3}}$ • Ceiling Contours Affect Detection Scaling • Including Q* Factor Allows Various Scales to Only Turbulent Flows are Burn Heskestad [11] Considered Time Collapse $\Pi_{tempersature} = \frac{\Delta T / T_{\infty}}{\Omega *^{2/3}}$ Scaling • Good Results: T_{gas}, u_{gas}, Y_i Pool Fires and Wood Cribs • Initial Smoke Front Arrival Time is Insensitive to Size for Power-Law Fires $\Pi_{fuel} = \frac{Y_i}{\left(\frac{y_i c_p T_{\infty}}{A_i}\right) Q^{*2/3}}$ Partitioned and Flat Ceilings Wood Cribs in Enclosures • Good Results: Tgas, Tsurface, YO2, YCO, YCO2 $P = (sD_c)^{1/2} \frac{A_o}{A}$ Burn Time • Agreement between Burning Rate Factor and Laminar Flow: $h_c \sim \ell^{1/2}$ Scaling Heskestad [12] Ventilation Factor $\Pi_{time} = t_R \sim b^{3/2}$ Based on Croce and Xin • Inaccurate Scaling of Wall Thickness Affected Two Sizes of Enclosures Crib: [13] $\Pi_{conv} = \frac{T_{\infty}k_{w}}{a\delta} = \frac{k_{w}}{h_{c}\delta} = \frac{1}{\Pi_{13}}$ Results Compared at 1/4, 1/2, Full $t_R \sim b^{3/2}$ • Radiation Groups Ignored Scale • Only Large Ventilation Factors were Examined

 Table 3.2: Past and Present Scaling Comparisons

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	Ventilation Factors Played a Key Role		$\Pi_{boundary} = \frac{k_{w}t_{R}}{\delta^{2}\rho_{w}c_{w}}$ $= \frac{t_{R}}{\Pi_{12}\left(\frac{\ell}{g}\right)^{1/2}}$	
Perricone [15]	Wood Cribs and Structural Loads Laminar Flow: h _c ~ $\ell^{1/5}$ ¹ / ₈ , ¹ / ₄ , ³ / ₈ Scale	Flow Time Scaling: $t \sim \ell^{1/2}$	$\Pi_{w,\delta} = \frac{\delta g^{1/4}}{\left(\frac{k}{\rho c}\right)_{w}^{1/2}} \ell^{1/4}$ $= \sqrt{\Pi_{12}}$ $\Pi_{w,c} = \frac{h_c \ell^2 T_{\infty}}{\rho_{\infty} c_p T_{\infty} \sqrt{g} \ell^{5/2}}$ $\Pi_{w,k} = \frac{\left(\frac{k\rho c}{t_R}\right)^{1/2} \ell^2 T_{\infty}}{\rho_{\infty} c_p T_{\infty} \sqrt{g} \ell^{5/2}}$ $\Pi_{13} \sim \frac{\Pi_{w,c} \Pi_{w,\delta}}{\Pi_{w,k}}$	 Transient Data Comparisons Ventilation of the Wood Crib is Important Good Results: T_{gas}, m[¨], q[¨]_{rad}, T_{Structural Elements} Radiation Considered Through Gas Emissivity Scaling (Π_e)

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Article III. Results for Scaling Starting or Small Fire

(Article III is a continuation of II as the two compose a MS thesis by Allison Carey; a break is made to allow consistency in the flow of the information in sections of the Final Report)

4. Design and Experimentation

Full Scale Experiments

To confirm the validity of the scaling method presented in Section 3, a series of full scale experiments was conducted at the Bureau of Alcohol, Tobacco, Firearms, and Explosives in Beltsville, MD. The experiments were conducted in a 4.5 meter high enclosure with joist-like partitions at the ceiling. Figure 4.1 is a representation of the facility.

The walls of the structure were constructed of 2x4 lumber, spaced 60.9 cm on center and sheathed with a layer of 1.6 cm thick gypsum wallboard. A series of 19 ceiling bays were created using oriented strand board (OSB) plywood partitions and 2x4 lumber. These partitions extended about 0.6 meters down from the ceiling. As seen in Figure 4.1, two smaller corridors extended off of the main area corridor of the enclosure. The ends of the corridor were open to the ambient air in the laboratory space. Temperature, flame height, smoke detector response, velocity, and smoke obscuration were measured throughout the test series. A load cell was used to measure the mass loss in the wood crib and polyurethane (PU) foam tests. Flame height was visualized with the aid of a metal stand marked off every 0.25 meters. This stand was located 1.78 meters away from the east wall and centered with the test specimen.



Figure 4.1: Plan View of ATF High-Bay Corridor Facility.

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This resource was prepared by the author(s) using Federal funds provided by the U.S. Department of Justice. Opinions or points of view expressed are those of the author(s) and do not necessarily reflect the official position or policies of the U.S. Department of Justice In general, instrumentation included thermocouples, hot wire anemometers, ionization detectors, photoelectric detectors, and optical density meters. A vertical thermocouple tree was located in every bay with thermocouples 5.1 cm, 15.2 cm, and 45.7 cm below the ceiling. Select bays were also instrumented with a horizontal thermocouple tree oriented parallel to the partitions. The tree was 30.5 cm below the ceiling with a thermocouple located every foot. The fuel in each test was positioned in "Location 2", as labeled in Figure 4.2. The smaller square represents the stand used to indicate flame height. Video cameras and still cameras were used to document the tests.



(a) Plan View of Fire Location.

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(b) Actual View of Fire Location. Figure 4.2: Fire Location Within the ATF Enclosure.

A natural gas burner, a heptane pool, pine wood cribs, and polyurethane foam (PU foam) blocks were used as the fuel sources. This research encompasses the scaled tests for the gas burner and heptane pool fires. Future work with this project will address scaling the wood cribs and the PU foam. The maximum heat release rate was calculated based on expected temperature rise in the enclosure. The height of the enclosure was 4.42 meters. The ambient temperature read by the thermocouples in the enclosure was 27°C, however this varied slightly throughout the enclosure. In preliminary tests, the natural gas burner was set at a heat release rate of 150 kW. The measurements in the enclosure did not vary greatly with respect to location with this size fire. It was decided that smaller heat release rates using the box burner would cause more of a variance in temperature, optical density, and velocity relative to position. This is ideal for physical modeling in order to test the accuracy of the scaling. A few larger gas burner fires were also included in the interest of gathering a wide range of data. In summary, data was collected for burners with a maximum heat release rate of 25 kW, 50 kW, 75 kW, 150 kW, 250 kW, and 300 kW over a 15 minute time period for each test. Most of these trials used a ramp function to prescribe the flow rate of the natural gas. One trial with no growth ramp was conducted for each fire size. The scale model is compared to these semi-immediate steady-state trials.

Heptane was burned in metal pans for the liquid fuel portion of the test series. 30.5 cm, 45.7 cm, and 61 cm round pans were each filled with heptane. These are referred to as small, medium, and large pool fires for this research. The amount of heptane was calculated using the burning rate based on the diameter of the pan. Water was poured into the pans before testing to create a level surface for the fuel to promote even burning. The pans were placed on a piece of gypsum wallboard to protect the concrete floor. All dimensions and liquid levels were measured

prior to burning the heptane. The heptane was left to burn until self extinction, which occurred three to five minutes after ignition.

The wood cribs were designed using a relationship presented by McCaffery [8], which described the relationship between the temperature rise in the enclosure, the height of the enclosure, and the heat release rate of the fire [18]. This was used to predict the heat release rate based on an expected temperature rise. The maximum heat release rate was determined to be between 380 kW and 730 kW. These maximum anticipated fire sizes and the enclosure properties were used to design the wood cribs. The heat release rate is defined as:

$$Q = \dot{m}\Delta h_c$$

(4.1)

where the heat of combustion for pine wood is about 12 kJ/g [16]. The mass loss rate is calculated based on the characteristics of the wood crib:

$$\dot{m} = Cn(4lD_c^{1/2}) \tag{4.2}$$

Where C is a material constant, about $1 \text{ mg/ cm}^{1.5}$ s for pine, n is the number of sticks in the crib, *l* is the length of one stick, and D_c is the diameter of one stick. The availability of wood sticks is limited. The length and diameter are predetermined and the resulting number of sticks is calculated using the mass loss rate.

$$\dot{m} = n \ (1)(4)(76.2cm)(2cm)^{1/2}$$

 $\dot{m} = (0.431n)g/s$
(4.3)

Using the heat of combustion and the optimal heat release rate of 400 kW, the number of sticks is determined to be 77 for the wood crib.

$$\dot{Q} = (5.172n)kW = 400kW$$
 (4.4)

The crib was organized into 11 layers with 7 sticks per layer. The spacing of the sticks was calculated based on the total length on one stick.

$$l = (\# sticks / layer)D_c + [(\# sticks / layer) - 1]s$$
(4.5)

Where l is the length, D_c is the diameter, and s is the spacing between sticks.

76.2cm = (7)(1.9cm) + [(7)-1]s

$$s = 10.5 cm$$

An optimal porosity of the crib is needed for proper burning characteristics [14]. The porosity for this crib design is 0.065, which is in the optimal range.

$$P = (sD_c)^{1/2} \frac{A_o}{A} = 0.065$$
(4.6)

The wood crib sticks were 1.9 cm square. Ultimately, the wood crib was designed to have a heat release rate of \sim 400 kW. A total of 77 sticks were used in the crib, with 11 layers of 7 sticks. The sticks were spaced 10.5 cm apart. Each stick was 76.2 cm long. The cribs were assembled using 2.54 cm metal staples. Figure 4.3 shows a wood crib before testing.



Figure 4.3: A Wood Crib Before Testing.

Three trials were conducted in the test series. A pan filled with heptane was placed under the wood crib to promote even ignition. The wood cribs were stored in an air conditioned room with a 17.4°C ambient temperature and 68% relative humidity. Each crib burned for about five minutes.

The PU foam was cut to be 76.2 centimeters on a side and 12.7 centimeters high. Cardstock was glued to the sides of the foam to prevent the fire from spreading faster at the edges. A 12.7 cm by 12.7 cm grid was drawn on the top face of the foam in order to track flame spread. A 2.54 cm wide arc groove was cut out in one corner of the foam, as seen in Figure 4.4. Fifty milliliters of heptane were poured into this groove. This was intended to ignite the foam and spread the fire uniformly from the corner.



Figure 4.4: Arc Groove in the Polyurethane Foam.

The PU foam was placed on a piece of gypsum board, which rested on top of the load cell. A pilot light ignited the heptane after it was poured into the foam groove. The test time began when the heptane ignited and ended when the foam had almost completely burned out. A CO_2 extinguisher was used after the test ended. Three foam tests were run in this series, with individual test times of about three to five minutes each. Table 4.1 outlines the complete experimental series for all of the fuels.

FUEL	FREQUENCY	DESIGN FIRE	
Natural Gas Burner	1 Trial Each	25 kW, 50 kW, 75 kW, 150 kW, 250 kW, 300 kW	
	2 Trials	30.5 cm round (70 kW)	
Heptane Liquid Pool	4 Trials	45.7 cm round (150 kW)	
	2 Trials	61 cm round (400 kW)	
		7 sticks per layer with 11 layers	
Pine Wood Crib	3 Trials	1.9 cm square pine 76.2 cm sticks	
		400 kW	
PU Foam Blocks	3 Trials	76.2 cm x 76.2 cm x 12.7 cm high	
	5 111015	400 kW	

 Table 4.1: The ATF High-Bay Experimental Series.

Model Fuel Scaling

The fuels used in the full scale experiments were scaled to $\frac{1}{8}$ model size using the general scaling laws. Heat loss was ignored. The model was considered to be an inviscid, unsteady flow. All experimental times were scaled according to Π_1 , where the ratio of the time for model experiments to time for full scale experiments is proportional to $\ell^{1/2}$.

$$t_{\text{mod }el} = t_{full} \left(\frac{\ell_{\text{mod }el}}{\ell_{full}} \right)^{1/2}$$
(4.7)

 $t_{\text{mod }el} = 0.35355t_{full}$

The times for the completed experiments in this research effort are summarized in Table 4.2. Table 4.2: The Full Scale and Model Experiment Times.

FUE	FIRE SIZE	FULL SCALE	MODEL SCALE
TOLL	I IKL SIZL	TIME (sec)	TIME (sec)
Natural Gas Burner	All	900	~320
	Small	~260	90-95
Heptane Liquid Pool	Medium	~300	100-120
	Large	~200	70-75

Natural Gas Burner

The natural gas burner was scaled geometrically using a $\frac{1}{8}$ scale. The dimensions of the model burner were 6.4 cm x 6.4 cm x 3.8 cm high. The burner was machined out of an aluminum

block. A mesh screen sat inside the burner 1.9 cm above the bottom. Below this screen, the gas entered the burner through a copper pipe with holes drilled into the sides. This configuration allowed the gas to disperse throughout the entire bottom of the burner. Small stones sat above the screen in order to diffuse the gas to the top surface of the burner. The completed gas burner is depicted in Figure 4.5.



Figure 4.5: The Natural Gas Burner for the Scale Model.

The necessary mass flow of natural gas was calculated using Π_9 and the full scale heat release rates. The ratio of the model heat release rate to the full scale heat release rate is proportional to $\ell^{5/2}$, or $(\frac{1}{8})^{5/2}$. The mass flow rate of the fuel was calculated using the model heat release rate and the heat of combustion of natural gas, 54 kJ/g [16].

$$\dot{m} = \frac{\dot{Q}_{\text{mod }el}}{\Delta h_c}$$

(4.8)

Table 4.3 describes the model gas burner experiments and their relation to the full scale tests. **Table 4.3: The Natural Gas Burner Heat Release Rates and Flow Rates.**

FULL SCALE EXPI	ERIMENTS	MODEL SCALE EXPERIMENTS		
HRR (kW)	MFR (g/s)	HRR (kW)	MFR (g/s)	
25	0.46	0.14	0.0026	
50	0.93	0.28	0.0051	
75	1.39	0.41	0.0077	
150	2.78	0.83	0.0153	
250	4.63	1.38	0.0256	
300	5.56	1.66	0.0307	

Heptane Pool Fire

In order to scale the pool fires, the diameter of the pan was scaled according to the desired heat release rate. Therefore, unlike the gas burner, the pool fires could not be scaled geometrically. The full scale pool fires were designed using laboratory data from the University of Maryland [8]. This data described the mass loss rate and the heat release rate of heptane burning on ceramic boards as a function of the board diameter. The full scale experiments at ATF provided large diameter pool data to verify the correlations. In order to create a representative trend for pool fires in pans, small diameter experiments were conducted at the University of Maryland. With the data from ATF and the University of Maryland, an additional correlation was created for the mass loss rate and the heat release rate of heptane burning in a pan as a function of the pan diameter. These relationships are shown in Figure 4.6 compared to the existing correlation for heptane burning in a porous ceramic disk.



(a)

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(b)

Figure 4.6: (a) Mass Loss Rate and (b) Heat Release Rate for Heptane.

The results were slightly different than for heptane burning on a ceramic board. Since the fuel was the same in the full scale and the model experiments, the ratio of the model heat release rate to the full scale heat release rate is proportional to $\ell^{5/2}$, or $(\frac{1}{8})^{5/2}$. The three model size heat release rates were matched using the full scale data and full scale Q*. The appropriate diameter of the pans were found using Figure 4.6. Pans were constructed out of steel to meet the precise diameter requirements. The mass loss rate per unit area was then found from the relationship:

$$\dot{Q} = \dot{m}'' \Delta h_c \frac{\pi}{4} D^2 \tag{4.9}$$

Finally, the amount of fuel required for each test was calculated from the known time and mass loss rate.

 $m = \dot{m}\Delta t$

(4.10)

A summary of the scale model heptane experimental set-up and the corresponding full scale experimental set-up is provided in Table III.4.

FULL SCALE EXPERIMENTS		MODEL SCALE EXPERIMENTS		
Diameter (cm) Initial Mass (g)		Diameter (cm) Initial Mass (g)		
30.5 446		4.6	0.83	
45.7	1108	6.1	1.91	
61	2562	9.9	4.93	

Table III.4: The Heptane Pool Diameters and Initial Masses.

Fuel Scaling Concerns

The independence or dependence of the dimensionless heat release rate, Q^* , relies on whether or not the heat release rate of the object is known. This is relatively simple to calculate from gas burners and pool fires. However, it becomes more complicated with other materials, especially when flame spread is introduced. When the heat release rate is known, Q^* is considered an independent property and is scaled accordingly. When the heat release rate is unknown, Q^* is dependent upon the other scaled parameters in the model. It is up to the researcher to recognize these parameters and match them through the scaling methodology.

Compartment Design

The selected partial scaling methodology requires some compromise when selecting the materials and thickness of the boundaries in the scale model. The theory results in very precise quantities that need to be rounded or modified based on availability of materials. The following sections describe the dimensionless groups used to design the compartment, some analysis of these groups relative to the scale model, and the construction of the compartment.

Application of Scaling Theory

The heat transfer dynamics that occurred in the full scale experiments must be replicated in the scale model. The heat transfer through the boundaries is especially important since the material thermal properties change with scale according to Π_{12} and Π_{13} . The density and conductivity of the materials scale proportional to $\ell^{9/20}$.

$$\Pi_{13} = \frac{h_c \delta}{k} \sim \left(\frac{g^{2/5} k_g}{v^{4/5}}\right) \frac{\ell^{1/5} \delta}{k}$$
(4.11)

Gravity, conductivity of the air, and dynamic viscosity of the air are the same in the full and ¹/₈ scales, this becomes:

(4.13)

$$\Pi_{13} = \frac{h_c \delta}{k} \sim \frac{\ell^{1/5} \delta}{k} \tag{4.12}$$

 $k \sim \ell^{1/5} \delta$

$$\Pi_{12} = \frac{\delta^2}{(k/\rho c)(\ell/g)^{1/2}}$$
(4.14)

$$\delta \sim \ell^{1/4} \tag{4.15}$$

By combining these relationships, the conductivity can be scaled using:

$$k \sim \ell^{1/5} \ell^{1/4} \sim \ell^{9/20} \tag{4.16}$$

$$\frac{k_{\text{mod }el}}{k_{full}} = \left(\frac{\ell_{\text{mod }el}}{\ell_{full}}\right)^{9/20}$$
(4.17)

This is also used for the density. The specific heat of most materials is about 1 kJ/kg-K, therefore the specific heat does not change for the scale model. Table 4.5 provides the thermal properties of the full scale boundaries along with the scaled values [20].

 Table 4.5: The Thermal Properties of the Full and Model Scale Boundaries.

	FULL SCALE			MODEL SCALE		
MATERIAL	k (W/mK)	ρ (kg/m ³)	c (kJ/kgK)	k (W/mK)	ρ (kg/m ³)	c (kJ/kgK)
Concrete	1.7	2400	0.75	0.67	941.5	0.75
OSB Chipboard	0.15	640	1	0.058	251.1	1
Gypsum Board	0.17	600	1.09	0.067	235.4	1.09
2x4 Wood	0.15	530	2.5	0.057	207.9	2.5

It is difficult to find materials that closely match the scaled values. The gypsum and the particle board were relatively close in density and conductivity, Kaowool was used for both in the scale model [21]. Marinite A was selected as the floor material [22]. Table 4.6 lists the thermal properties for the selected boundary materials in the scale model. **Table 4.6: Thermal Properties of the Selected Material for the Model.**

er nes or the sereetea		me nie aen		
	ACTUAL VALUES			
MATERIAL	k	ρ	с	
	(W/mK)	(kg/m^3)	(kJ/kgK)	
Marinite A	0.28	1041	1.1	
Kaowool	0.06	250	1.08	

These values do not match the scaled values exactly; however the materials are still acceptable for building a scale model. The thickness of the boundaries scales according to $\ell^{1/4}$.

$$\frac{\delta_{\text{mod }el}}{\delta_{full}} = \left(\frac{\ell_{\text{mod }el}}{\ell_{full}}\right)^{1/4} \tag{4.18}$$

The 1.59 cm walls were scaled to 0.95 cm and the 1.27 cm partitions/ceiling were scaled to 0.74 cm using this relationship. The selected materials are not manufactured in these thicknesses. Therefore, the model walls were selected to be 0.953 cm thick and the model ceiling and partitions were selected to be 0.64 cm thick. The small differences in the boundary thicknesses could be a potential source of future error. The depth of the concrete floor in the full scale experiments was not measured since it was the floor in the lab space. Instead, it was assumed that this deep slab of concrete acted as a heat sink to the fires within the enclosure. The thermal penetration depth was calculated according to the thermal properties and the penetration time.

$$\delta \sim \sqrt{\alpha} t \tag{4.19}$$

$$\alpha = \frac{k}{1 + 1}$$

 ρc (4.20)

For the model, an 8 mm penetration depth was calculated. The Marinite A slab for the floor of the scale model was 1.27 cm thick based on these calculations.

The actual values (thermal properties and thickness) for the boundary materials were compared to the theoretical values by calculating the dimensionless ratios $(\Pi_{12})_{model} / (\Pi_{12})_{full}$ and $(\Pi_{13})_{model} / (\Pi_{13})_{full}$. The closer these ratios are to one, the more accurate the selected material represents the scaling theory. The ratios for all boundary materials are listed in Table 4.7. Table 4.7: The Comparative Ratios for the Model Physical Boundaries.

	110000101 1100001 1 119 STORE 200	
BOUNDARY	$(\Pi_{12})_{model}$ / $(\Pi_{12})_{full}$	$(\Pi_{13})_{model}$ / $(\Pi_{13})_{full}$
Concrete	0.77	0.94
Gypsum Board	1.09	1.12
OSB Chipboard	0.69	0.83

From these values, the selected materials are an acceptable representation of the full scale boundaries. The discrepancies between the scaled values and the actual material properties may become a source of error when analyzing the results.

Reynold's Number Specifications

Since the buoyancy flow was preserved in the general scaling methodology, the Reynold's number was assumed to be turbulent at all scales. The Reynold's number was used to verify turbulence in the model.

$$\operatorname{Re} = \frac{\sqrt{gh^{3/2}}}{D} \tag{4.21}$$

Where h is the height of the enclosure. Since the kinematic viscosity of air varies with temperature, the Reynold's number was calculated at 20°C, 100°C, and 200°C, as seen in Table 4.8. Based on full scale results, the model is not expected to reach temperatures above 200°C. **Table 4.8: Verifying Turbulence for the Scale Model.**

TEMPERATURE	REYNOLD'S NUMBER
20°C	0.8 x 10 ⁵
100°C	0.5 x 10 ⁵
200°C	0.2 x 10 ⁵

Re> 10⁵ is considered turbulent. These values do not account for mixing effects when the plume impinges on the ceiling and partitions; therefore it was assumed that the flows in all experiments remained in the turbulent regime.

% Scale Compartment Construction

The model was built on top of a drywall and metal stud foundation attached to two rolling tables for mobility around the lab. The Marinite A slab was bolted onto this foundation. The model as a whole was designed to allow relatively easy modification to the structure in case of emergency. The enclosure was constructed separately from the foundation, meaning it could be lifted and

modified if necessary. An aluminum channel support system made of 80/20 formed the enclosure. All dimensions of the enclosure were scaled geometrically ($\ell = \frac{1}{8}$). Note from Appendix A that the width of each bay varied slightly and some partitions were shorter than others. This was taken into consideration and scaled accordingly. The partitions were angled at the sides. The missing portion formed an isosceles right triangle. The angle remained the same in the scale model in order to ensure that the same flow characteristics were achieved from bay to bay. The Kaowool panels were measured, cut, and secured to the inside of the 80/20 structure. Seams between panels were filled with liquid Kaowool cement, which had the same material properties as the solid board. The partitions were held at a 90° angle using small clips that extended into the enclosure through the top of the ceiling. The partitions were then secured with the Kaowool cement. Various angles of the model can be seen in Figure 4.7.





The dimensions of the finished scale model (main corridor) measured 2.2 meters in length by 0.97 meters in width by 0.56 meters in height. The location of the burner and metal pans were scaled geometrically.

Measurement and Instrumentation

Scaling the Thermocouples

In the full scale experiments, 28 AWG Type K glass insulated thermocouples were used. The thermocouples were scaled to reduce potential error from a large diameter wire relative to the small bays in the scale model. The scaling methodology considered the diameter of the wire, the flow time in the model, and the velocity of the flow around the thermocouple. Conduction and convection were taken into account to create and additional Π group for thermocouples [8].

$$mc_{p}\frac{dT}{dt} = hA_{s}(T_{g} - T)$$
(4.22)

$$\Pi_{TC} = \frac{hA_s}{mc_p} t_R \tag{4.23}$$

$$\Pi_{TC} \sim \left(\frac{\ell^{1/2}}{d}\right)^{3/4}$$
(4.24)

The thermocouples for the scale model were matched as closely as possible to the scaled diameter; therefore 40 AWG Type K glass insulated thermocouples were used. The thermocouples wire was not rigid enough to remain in position during the experiments. As a result, small brackets were formed to hold the thermocouples in the desired location, as seen in Figure 4.8. The brackets were coated with liquid Kaowool to ensure the metal did not affect the temperature results. The locations of the thermocouples on the brackets were scaled geometrically based on the full scale thermocouples distance from the ceiling. High temperature resistant RTV silicone was used to hold the thermocouples to the brackets.



Figure III.8: The Thermocouples on Brackets.

The full scale enclosure was instrumented with over 60 thermocouples. The ¹/₈ model used 30 thermocouples. They were chosen based on elevation and distance from the fire. Table 4.9 describes the bay location, orientation, and elevation for the 30 thermocouples used in the scale model. Note: These values list the full scale distances for future comparison purposes.

III-14

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Table I	II.9 :	Scale	Model	Thermocou	ples	Locations.
---------	---------------	-------	-------	-----------	------	------------

Bay Location	Tree	Distance From	Distance From
	Orientation	Ceiling (cm)	East Wall (m)
Bay 2	Vertical	5.1	0.91
Bay 2	Vertical	15.2	0.91
Bay 2	Vertical	45.7	0.91
Bay 4	Vertical	5.1	0.91
Bay 4	Vertical	15.2	0.91
Bay 4	Vertical	45.7	0.91
Bay 6	Vertical	5.1	0.91
Bay 8	Vertical	5.1	0.91
Bay 10	Vertical	5.1	0.91
Bay 10	Vertical	15.2	0.91
Bay 10	Vertical	45.7	0.91
Bay 13	Vertical	5.1	0.91
Bay 15	Vertical	5.1	0.91
Bay 16	Vertical	5.1	0.91
Bay 17	Vertical	5.1	0.91
Bay 19	Vertical	5.1	0.91
Bay 19	Vertical	15.2	0.91
Bay 19	Vertical	45.7	0.91
Bay 2	Horizontal	30.5	0.31
Bay 2	Horizontal	30.5	3.4
Bay 2	Horizontal	30.5	4.6
Bay 4	Horizontal	30.5	0.31
Bay 4	Horizontal	30.5	3.4
Bay 4	Horizontal	30.5	4.6
Bay 10	Horizontal	30.5	0.31
Bay 10	Horizontal	30.5	3.4
Bay 10	Horizontal	30.5	4.6
Bay 19	Horizontal	30.5	0.31
Bay 19	Horizontal	30.5	3.4
Bay 19	Horizontal	30.5	4.6

Future Instrumentation

The pine wood crib and PU foam block experiments will also include a load cell under the fuel. If possible, optical density meters and ionization/photoelectric detectors should be included. Scaling these devices would be beneficial to investigators for analyzing detector response and front arrival times. They would also assist in modeling toxicity levels due to the presence of smoke as measured by the optical density meters. The full scale ATF test series is an excellent scaling example since a wide range of instrumentation was used.

Data Acquisition

The thermocouples and gas flow meter were connected to two NetDAQ Fluke 2645A data acquisition systems. The systems were interconnected to allow simultaneous scans. Each system was equipped with a terminal block consisting of 20 analog channels to convert transducer signals and relay them to a laptop PC equipped with an Ethernet connection and the *Fluke NetDAQ Logger* software [23]. A single channel was used to collect data from the mass flow meter during the gas burner tests. Measurements were taken at one second intervals.

5. Results and Discussion

Scaled High Bay Test Series

Comparisons Between 1/8 and Full Scale Models

The goal of scale modeling is to reproduce full scale values or to create an accurate prediction of expected full scale values. The transient data is compared against the full scale time values, meaning the $\frac{1}{8}$ scale model time has been increased by a factor of $\ell^{1/2}$. To measure the success of the partial scaling laws applied here, the following data will be examined.

- The temperature results between experimental trials will be compared for the scale model. This will give an indication of the repeatability of the scaled test series.
- The ¹/₈ scale fires were designed using a target heat release rate from the full scale experiments. The fuel supply rate of the gas and the burning rate of the gas and the heptane will be compared between the full and scale model.
- The uncertainty of the ¹/₈ model results compared to the full scale temperature values will be assessed. A statistical t-test was also be applied to the data to display the relationship between the full and model data.
- The general temperature rise results are discussed. These results gave an indication of the overall performance and application of the scaling laws for these fires. The transient temperatures will be observed based on fire size, fuel type, and thermocouple location. In particular, the temperature will be compared between scales for locations close to the ceiling, far from the ceiling, and various axial distances from the fire.
- The flame height and general flame behavior, such as twisting or necking, are compared for the gas burner and the heptane pool fire.
- The temperature rise at steady state will be compared between the ¹/₈ and full scale models. Steady state was determined by the duration of the test and visual indications during the experiments. Note that both fuels reach steady state relatively fast.

Experimental Procedure

The heptane pool fires were conducted first in this test series. Water was poured into the pans in order to create an even surface for fuel burning. The 4.6 cm diameter and 6.1 cm diameter pans (modeling the 30.5 cm and the 45.7 cm pans) were each filled with 25 mL of water. The 9.9 cm pan (modeling the 61 cm pan) was filled with 50 mL of water. The heptane was measured using a 10 mL graduated cylinder and an eye dropper to ensure accuracy. The amount of fuel for each size was calculated in Section 4.2.2. The 4.6 cm pan required 1.22 mL of heptane, the 6.1 cm pan required 2.82 mL of heptane, and the 9.9 cm pan required 7.3 mL of heptane. For each trial, the pan was positioned in the center of the enclosure under Bay 16. The NetDAQ system was started

and allowed to run a few seconds in order to synchronize scans between the two Fluke systems. A video camera was also started at this time. The heptane was lit using a butane lighter and a stop watch was started. Various still camera shots were taken. The heptane was allowed to burn to extinction and the total burn time was recorded. The video camera and data acquisition system were stopped. The lab hood was turned on to exhaust any gases and assist in cooling down the enclosure. The hood was not used during the tests since the mechanical exhaust had a significant impact on the flame behavior. While the enclosure cooled back to ambient temperature (~20-22°C), the pan was removed, emptied, and refilled with the appropriate amount of water and heptane to ensure the same initial conditions. This process was repeated for each trial. Three trials were conducted for the small pool fires, five trials were conducted for the medium pool fires, and three trials were conducted for the large pool fires.

A mass flow meter was connected to the NetDAQ system for the gas burner tests. Commercial grade methane was used at 137.9 kPa. Small stones were poured into the burner after testing the gas flow. The burner was placed and secured in the center of the enclosure below Bay 16. Due to the precision of the mass flow meter, the 25 kW experiments were not modeled. The scaled flow for the 25 kW fires was not strong enough to sustain a flame. Some flashing did occur, but this was not consistent enough for a comparison with the full scale experiments. Tests were conducted for the 50 kW, 75 kW, 150 kW, 250 kW, and 300 kW fires. In each trial, the NetDAQ system was started and allowed to run a few seconds in order to synchronize scans between the two Fluke systems. A video camera was also started at this time. The flow meter was opened to the prescribed mass flow depending on the size of the fire, as noted in Section 4.2.1. This was done manually. The valve on the flow meter was very sensitive, resulting in some variation from the full scale experiments due to human error. The mass flow meter measured flow in L/min. Table 5.1 displays the mass flow for each fire size.

Full Scale Fire Size	¹ / ₈ Scale Fire Size	Mass Flow (L/min)
50 kW	0.28 kW	0.428
75 kW	0.41 kW	0.642
150 kW	0.83 kW	1.29
250 kW	1.38 kW	2.14
300 kW	1.66 kW	2.57

Table III.5.1: The Mass Flow for Each Gas Burner Fire.

The burner was lit using a butane lighter and a stop watch was started. Various still camera shots were taken. Each trial burned for 318 seconds. At this time, the valve on the mass flow meter was closed and the flame extinguished immediately. The data acquisition and video camera were stopped. The hood was turned on to exhaust gases and cool down the enclosure to the ambient temperature. Two trials were conducted for each gas burner fire size.

General Observations

The flame in both the gas burner and heptane pool fire tests acted similar to the full scale experiments. As mentioned earlier, the gas burner tests used methane instead of natural gas. Both fuels burn relatively clean, however the smaller scaled experiments (50-150 kW fires) did not produce flames as yellow/orange as the full scale. Additionally, the flame height in the scaled experiments was slightly lower than in the full scale for all tests (see Figure 5.1), whereas the overall shape of the flame was the same between the two scales. The flame height is analyzed further in Section 5.2.



Full Scale

¹/₈Scale

Figure 5.1: Visual Flame Height Comparison for 300 kW Gas Burner Tests.

The flame in the $\frac{1}{8}$ scale experiments burned evenly across the burner suggesting that the burner design allowed the gas to fully disperse. The burner produced turbulent flames in all tests. It was difficult to capture a turbulent flame on camera for the 50 kW and 75 kW model experiments since the flame tip was a very faint blue. Spinning and leaning of the flame was observed in the experiments due to the burner placement relative to the small corridors. These patterns were also experienced in the full scale test series.

The heptane pool fires were also successful in simulating the full scale fire scenarios. The bright orange heptane flame produced smoke that could be seen exiting the main corridor, as noted in Figure 5.2.



Figure 5.2: Smoke Exiting the Scaled Corridor.

While optical density measurements and smoke detection were not included in the model experiments, the smoke dispersion was visually similar. This is notable due to the complex ceiling geometry created by the partitions. The scaled flame height visually matched the heights from the

full scale experiments. The heptane burned steadily in all tests. In the large pool fires, necking of the flame due to air entrainment was observed. This also resulted in a wider turbulent flame farther downstream as noted in Figure 5.3. The flame tips in the ¹/₈ and full scale large pool fires lifted off of the main body of the flame and extended almost to the height of the partitions.



Full Scale

1/8Scale

Figure 5.3: Visual Flame Height, Necking, and Turbulence for the Heptane Pool Tests. The flames in both the model and full scale experiments leaned slightly. Once again, this was due to the location of the fire relative to the two smaller corridors, which provided additional ventilation to the main corridor space. The heptane was allowed to burn until extinction. As the heptane supply diminished, the flame got noticeably smaller and less soot was produced. In some trials, a small portion of the heptane surface kept burning (see Figure 5.4).



Figure 5.4: Flamelet Due to Unevenly Spread Heptane.

These small flamelets were a result of extra heptane pooled at that location. The flamelets only burned an addition 3-5 seconds past the burnout of the rest of the flame, so they had little impact on the overall temperature results.

Repeatability of Scale Model Experiments

Repeatability ensures the test procedure is reliable. To assess the repeatability of the gas burner and heptane pool fires, the temperature rise distribution between trials was compared. Figures 5.5 through 5.9 show trials for each fuel at various sizes and locations.



Figure 5.5: ¹/₈ Scale Small Pool Fire Trials Above Fire

Figure 5.5 displays similar temperature trends and values. Some variation in temperature is acceptable due to human error. Small differences in the water level in the pan or the amount of heptane can result in slight temperature variations. These temperatures were measured directly above the fire where slight changes in flame height or flame trubulence also have an effect on temperature.



Figure 5.6: 1/8 Scale Large Pool Fire Trials Above Fire



Figure 5.7: 1/8 Scale 50 kW Gas Burner Trials Above Fire

Figure 5.7 shows that the two trials conducted for the 50 kW gas burner measured similar temperatures. There is some discrepancy between the two trials within the first minute of the experiment. This is directly related to human error since the gas flow meter was very sensitive and the flow was manually adjusted to the prescribed value.



Figure 5.8: 1/8 Scale 250 kW Gas Burner Trials Above Fire



Figure 5.9: 1/8 Scale 250 kW Gas Burner Trials Far From Fire

Figure 5.9 plots the temperatures in Bay 2 for the 250 kW trials. Distance is a factor when comparing the full scale to the ¹/₈ scale temperature rise data. Bays far from the fire show a greater temperature rise difference from the full scale data. Between trials in the scale model, this is not the case. This graph demonstrates that the boundary materials have the same heat transfer effects between trials, as expected. Both curves follow the same temperature rise trends. Based on the above comparisons, the ¹/₈ scale model demonstrates repeatability between trials. Temperature rise comparisons between this data and the full scale data are an accuarate representation of this scale model's performance. The slight variations in temperature between trials will contribute to a few degrees of uncertainty. The comparisons between full and model scale will be influenced by the trials compared from each test series.

Repeatability of the Full Scale Experiments

Multiple trials were also conducted for the full scale heptane pool fires experiments. The temperature distributions throughout the enclosure from each trial were compared to assess the repeatability of the full scale experiments. Recall that the mass flow rate was prescribed to grow as a function of time for some trials. One trial for each fire size did not include this ramp function. Since this research focuses on static fires that reach steady-state quickly, only one full scale trial for each fire size is compared to the model experiments. As seen in Figures 5.10-5.12, the full scale trials for the heptane pools demonstrate acceptable repeatibility.

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Figure 5.10: Full Scale Small Pool Fire Trials Above Fire



Figure 5.11: Full Scale Medium Pool Fire Trials Above Fire



Figure 5.12: Full Scale Large Pool Fire Trials Above Fire

There is more variation between trials in these full scale experiments. This will impact the scale model comparisons. At some times, a 10°C difference is seen between trials with the same fire size. These are especially notable for the medium pool fires. Uncertainty in temperature rise comparions between the full scale and scale model may originate from simple temperature discrpancies between trials.

Fuel Supply Rate and Burning Rate Natural Gas Burner

The fuel supply rate and heat release rate of the natural gas burner were very closely matched between the $\frac{1}{8}$ and full scale experiments. To determine the prescribed flow of the $\frac{1}{8}$ scale gas burner, the dimensionless heat release rate Q* was matched to the full scale Q*. Some noise is present in the beginning of the model experiments due to the sensitivity of the flow meter and human error. Figure 5.13 shows the heat release rates for the full scale and the model burner at 50 kW, 150 kW, and 250 kW based on the mass flow data measured during each experiment. Figure 5.13 scales the dimensionless heat release rate values to the full scale fire sizes. The scale model did not have a 50 kW, 150 kW, or 250 kW fire. The proximity to these values is a result of the increasing the model fire size by a factor of $\ell^{5/2}$. It is still an accurate representation of the model dimensionless heat release rate compared to the full scale dimensionless heat release rate.



Figure 5.13: Heat Release Rate of Model and Full Scale Gas Burner.

The flow meter was accurate to 0.01 L/min and the flow for each trial was prescribed manually at the beginning of the test. It was especially difficult to adjust the flow meter for the low flows corresponding to the scaled 50 kW and 75 kW experiments.

It is important to verify the prescribed scaling methodology using a controlled fuel. A gas burner is the simplest fuel to scale since there is a constant mass flow and a defined mass loss rate that quickly reaches steady state. This means the actual heat release rate of the burner is close to the theoretical Q values [18, 24]. With other fuels, radiative heat feedback or fuel depletion can have an effect on the burning characteristics and heat release rate. Therefore, it is suggested that a fuel with a prescribed burning rate, such as a gas burner, is used to reinforce the prescribed scaling methodology.

Heptane Pool Fire

The fuel supply rate and dimensionless heat release rate matched reasonably well between the $\frac{1}{8}$ and full scale experiments. Since a load cell was not used in either test series, the mass loss rate of the heptane was estimated based on the initial amount of fuel and the elapsed time of the experiment. This was initially done using the full scale (FS) test data in order to calculate Q* and the amount of fuel needed for the scaled experiments. Table 5.2 provides a summary of the average values for the test duration and dimensionless heat release rate for each fire in the $\frac{1}{8}$ and full scale models.

Fire Size	¹ / ₈ Time (sec)	¹ / ₈ Q*	FS Time (sec)	FS Q*
Small	99	0.027	93	0.030
Medium	94	0.067	109	0.064
Large	82	0.194	73.5	0.227

Table 5.2: Average Elapsed Time and Q* Comparison.

The slight differences between the two test series are likely due to the lack of data on the mass loss rate of the fuel. Since the mass loss rate may fluctuate as a function of time, a load cell would allow a more accurate mass loss rate and dimensionless heat release rate to be calculated and applied in the scale model. The faster mass loss rates in the full scale experiments result in a higher heat release rate. Since the ¹/₈ scale tests burned slightly longer than the full scale tests did, the heat release rate was lower. This accounts for some of the temperature discrepancies between the model and full scale measurements, as discussed in Section 5.1.4.

General Temperature Results

Overall, the proposed scaling theory provided a good method to predict full scale temperatures using a $\frac{1}{8}$ scale model. The best temperature results were measured directly above the fire in Bay 16. Reasonable results were also obtained in the two bays on either side of Bay 16; Bay 13, Bay 15, Bay 17, and Bay 19. The results were assessed using a maximum temperature rise uncertainty, a statistical t-test, and plots of temperature rise as a function of time. For all trials, the temperature rise was used instead of the measured temperature. This was done in order to reduce environmental impacts on the comparison. The full scale experiments were conducted in an open lab in July, where the ambient temperature was $\sim 27^{\circ}$ C. The small scale experiments were conducted in a closed lab in January, where the ambient temperature was $\sim 20^{\circ}$ C.

Maximum Temperature Rise Uncertainty

The maximum temperature rise uncertainty was calculated similar to validation and verification efforts by the Nuclear Regulatory Commission. The method resulted in a percentage of uncertainty based on the peak temperature relative to the ambient temperature. [25]

$$\frac{\Delta M - \Delta E}{\Delta E} x_{100} = \frac{\left(T_{\max} - T_{\infty}\right)_{\text{mod }el} - \left(T_{\max} - T_{\infty}\right)_{full}}{\left(T_{\max} - T_{\infty}\right)_{full}} x_{100} = \%$$
(5.1)

Where E represents the expected (full scale) values and M represents the model values. This percentage represents the relative difference between the model predictions and the full scale measurements [25]. A maximum value approach was used since the uncertainty between scales could not be compared as a function of time. The full scale data was collected every second. The $\frac{1}{8}$ scale data was collected every second in the experiments, but was then scaled using the t $\sim \ell^{1/2}$ scaling law. This results in fewer data points for the scale model. A transient comparison requires the measurements to have comparable time steps.

Twenty percent was decided to be the highest level of acceptable uncertainty. This would account for any discrepancies between temperature data due to human error, experimental measurement uncertainty, or model sensitivity due to model input uncertainty (such as boundary differences) [25]. If the uncertainty is positive, the full scale temperature difference was higher than the model temperature difference. If the uncertainty is negative, the full scale temperature difference was lower than the model temperature difference. In Figure 5.14 and 5.15, the absolute value of the uncertainty was used to show the uncertainty trends as a function of axial distance from the fire. The fire is located at the origin. Negative distance values represent bay locations north of the fire (Bays 2-15) and positive distance values represent bay locations south of the fire (Bays 17-19).



Figure 5.14: Pool Maximum Temperature Rise Uncertainty vs. Distance from Fire

The maximum temperature rise uncertainty increases dramatically for locations far from the fire. In fact, only two locations on either side of the fire have an acceptable level of uncertainty. The same trend occurs for the gas burner fires.



Figure 5.15: Burner Maximum Temperature Rise Uncertainty vs. Distance from Fire

Therefore, based on the maximum temperature rise, only Bays 13, 15, 16, 17, and 19 are reasonably accurate at simulating the full scale fire temperatures. This is directly related to the boundary

materials. The discrepancies in the thermal properties of the actual materials compared to the theoretical values cause a variation in the predicted heat transfer properties of the enclosure. In other words, due to the conductivity and density of the Kaowool, the ability of the ¹/₈ scale temperatures to predict the full scale temperatures decreases with increasing distance from the fire.

Statistical T-test

A statistical t-test was performed to show the significance between the $\frac{1}{8}$ scale data and the full scale data. Each set of temperature data has its own statistical mean. The t-test measures the means of each test relative to the entire temperature distribution and assigns a p-value which compares the two sets of data. This p-value indicates how likely it is that the scaled results occurred by chance. A low p-value means that the model is a significant representation of the full scale data and vice versa. The t-tests applied for this research assumed unequal variances and independent data sets. A two-tailed approach was used with α =0.05. Therefore, a low p-value for these tests is below 0.05. The temperature values were not averaged between trials. Each trial was compared to the full scale data for the heptane pool fires and the gas burner fires. In every case, p<<0.05, meaning there is no significant difference between the model and full scale data. Therefore, the $\frac{1}{8}$ scale model is representative of the full scale data. [26]

Transient Temperature Results

While the maximum temperature rise uncertainty and the t-test provided a good understanding of the relationship between the $\frac{1}{8}$ and full scale data, it is important to look at the temperature rise as a function of time. When comparing transient temperature data, it is necessary to consider the time step associated with each set of data. The $\frac{1}{8}$ experiments were shorter than the full scale experiments. In order to compare the results as a function of time, the $\frac{1}{8}$ scale data was stretched by a factor of $\ell^{1/2}$ to correspond to the full scale time. This means there are significantly fewer data points in the $\frac{1}{8}$ scale data. A five point running average was applied to the model data and a ten point running average was applied to the full scale data in order to smooth the temperature rise curves.

Temperature Distribution "Near" the Ceiling

The differences between the full scale and $\frac{1}{8}$ scale model temperatures were affected by the location of the thermocouple in the enclosure. Temperatures were measured 5.1 cm, 15.24 cm, and 45.7 cm below the ceiling in the full scale experiments. Figures 5.16-5.18 compare the temperature rise 0.72 meters away from the fire (Bay 17) for the heptane pool fires. These measurements were recorded 5.1 cm below the ceiling.



Figure 5.16: Temperature Rise 5.1 cm Below Ceiling in Bay 17 (0.72m away from fire) for Small Pool Fire.



Figure 5.17: Temperature Rise 5.1 cm Below Ceiling in Bay 17 (0.72m away from fire) for Medium Pool Fire.



Figure 5.18: Temperature Rise 5.1 cm Below Ceiling in Bay 17 (0.72m away from fire) for Large Pool Fire.

Bay 17 is one bay removed from the fire. There is a slight delay in temperature rise due to the time lag associated with the convective flow. The thermocouples are close to the ceiling, meaning that some fluctuations will occur between models due to the turbulence at that elevation. The partitions and the ceiling cause mixing in the flows close to the boundaries. The 40 AWG thermocouples used in the scale model are sensitive to the temperature variations caused by this turbulence. The $\frac{1}{8}$ model data, represented by the dotted line, is within 15°C of the full scale data at all times. Similar trends also occurred for the gas burner, where the $\frac{1}{8}$ scale temperatures were within 15°C at Bay 17.

Temperature Distribution "Far" From the Ceiling

The ¹/₈ scale and full scale temperatures are closer for thermocouples farther from the ceiling. This is because the mixing caused by the boundaries obstructing the flow diminishes. The thermocouples located 45.7 cm below the ceiling show a steadier temperature distribution. These results did a better job at predicting the full scale temperature data. This is shown in Figures 5.19-5.21, which depict Bay 10 at 45.7 cm below the ceiling for the heptane pool fires. It is important to compare temperature results at various elevations, but there is limited data since vertical thermocouples trees were only placed in Bays 2, 4, 10 and 19 in the scale model. This means that the temperatures 45.7 cm from the ceiling were only measured in a few bays.



Figure 5.19: Temperature Rise 45.7 cm Below Ceiling in Bay 10 (4.32m away from fire) for Small Pool Fire.



Figure 5.20: Temperature Rise 45.7 cm Below Ceiling in Bay 10 (4.32m away from fire) for Medium Pool Fire.



Figure 5.21: Temperature Rise 45.7 cm Below Ceiling in Bay 10 (4.32m away from fire) for Large Pool Fire.

At 45.7 cm below the ceiling, the difference in temperature rise between the full scale and the $\frac{1}{8}$ scale is less than the temperature difference at 5.1 cm below the ceiling. The overall temperature rise growth trend is also a better match farther from the ceiling. This is a result of less turbulence and the development of a layer at a farther distance from the ceiling. The $\frac{1}{8}$ scale gas burner tests were also closer to the full scale temperatures at 45.7 cm than at 5.1 cm below the ceiling. Figure 5.18 shows a 15°C difference between scales for the large pool fire; the same difference for temperature close to the ceiling. This introduces the temperature differences based on location from the fire. The 5.1 cm from the ceiling data in Figure 5.18 is six bays removed from the fire, whereas the 45.7 cm from the ceiling data in Figure 5.18 is six bays removed from the fire. The predictive capabilities of the model are affected by distance due to the material boundaries. This phenomenon is discussed at length in upcoming sections. Obtaining the same temperature difference six bays away shows that more accurate results are achieved at 45.7 cm below the ceiling.

Temperature Distribution Within the Bay

Horizontal thermocouples trees were placed in Bays 2, 4, and 10 in the full scale and ¹/₈ scale experiments. Bay 2 is 10.08 meters from the fire, Bay 4 is 7.2 meters from the fire, and Bay 10 is 4.32 meters from the fire. The tree was located 30.48 centimeters below the ceiling in each bay. While the larger scale had a thermocouple every 0.31 meters, the small scale experiments only had thermocouples corresponding to the 0.31m, 3.4m, and 4.6m full scale locations. Recall that an isosceles triangle was cut out of the east side of the partitions while the west side remained straight. The 0.31m location was adjacent to the angled side of the partition. This allowed the hot gases to flow freely into the next bay. The 4.6m location was bounded by the Kaowool partitions. The hot gases either had to fill up in the previous bay and spill over the partition or travel through the angled side of the partition and down the length of the bay in order to reach the 4.6m location. Regardless of how the hot gases reached the thermocouple, there was a significant effect on the

heat transfer as a result of the prolonged exposure to the partition material. The temperature rise comparison was very accurate inside of the bays for the large gas burner tests. The smaller gas burner tests matched well at the 0.31m location, but poorly at the 4.6m location. This is a result of the enclosure boundary materials and the configuration of the partitions. Figures 5.22-5.27 show the temperature rise at the 0.31m and 4.6m locations for the 50 kW, 150 kW, and 300 kW fires.



Figure 5.22: Temperature Rise 0.31m from East Wall in Bay 10 for 50 kW Gas Burner.



Figure 5.23: Temperature Rise 4.6m from East Wall in Bay 10 for 50 kW Gas Burner.

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Figure 5.24: Temperature Rise 0.31m from East Wall in Bay 10 for 150 kW Gas Burner.



Figure 5.25: Temperature Rise 4.6m from East Wall in Bay 10 for 150 kW Gas Burner.



Figure 5.26: Temperature Rise 0.31m from East Wall in Bay 10 for 300 kW Gas Burner.



Figure 5.27: Temperature Rise 4.6m from East Wall in Bay 10 for 300 kW Gas Burner. From these graphs, it is apparent that good results were obtained at all thermocouples located 0.31m from the east wall; the temperature rise between scales is within 10°C of one another. Only the largest fire was accurate at the 4.6m thermocouple location. After the first minute of the 300 kW fire, the temperature rise in the ¹/₈ scale and the full scale are very close. This shows that effects due to the material boundaries diminish over time for larger fires. The horizontal thermocouple

trees in the heptane pool fire experiments were also affected by the heat losses due to the boundary materials. Similar results were achieved where the temperatures at 0.31m from the east wall were within 5°C of the full scale data but the thermocouples 4.6m from the east wall were not as accurate with a temperature difference doubling to ~10°C. Figures 5.28-5.31 shows the temperatures at the 0.31m and 4.6m locations for the small and medium heptane pool fires.



Figure 5.28: Temperature Rise 0.31m From East Wall in Bay 10 for Small Pool Fire.



Figure 5.29: Temperature Rise 4.6m From East Wall in Bay 10 for Small Pool Fire.



Figure 5.30: Temperature Rise 0.31m From East Wall in Bay 10 for Medium Pool Fire.



Figure 5.31: Temperature Rise 4.6m From East Wall in Bay 10 for Medium Pool Fire.

It is important to recognize that the bays instrumented with the horizontal thermocouple trees (Bays 2, 4, 10) did not scale as well as the bays closer to the fire. This has a definite impact on the success of the horizontal thermocouple tree scaling. Future research may want to place additional instrumentation in Bays 13, 15, 16, 17, and 19 since these bays were the most accurate using this scaling methodology.

Temperature Based on Fire Size and Axial Distance from the Fire

Smaller fires for the heptane pool and gas burner have only a few degrees of difference between the $\frac{1}{8}$ and full scale data. The larger fires have a much greater discrepancy between scales. For a small burner fire, at 50 kW, the difference in the $\frac{1}{8}$ and full scale data is less than 5°C at any given time at 45.7 cm below the ceiling. Figures 5.32-5.34 show the temperatures for the 50 kW fire. Bays 4, 10, and 19 are examined, which provide temperature comparisons both close to and far from the fire. Note that Bay 4 is 8.64 meters from the fire, Bay 10 is 4.32 meters from the fire, and Bay 19 is 2.16 meters from the fire.



Figure 5.32: Temperature Rise 45.7 cm Below Ceiling in Bay 4 (8.64m away from fire) for 50 kW Gas Burner.



Figure 5.33: Temperature Rise 45.7 cm Below Ceiling in Bay 10 (4.32m away from fire) for 50 kW Gas Burner.



Figure 5.34: Temperature Rise 45.7 cm Below Ceiling in Bay 19 (2.16m away from fire) for 50 kW Gas Burner.

The 250 kW fire had a larger temperature difference between the $\frac{1}{8}$ and full scale experiments. Figures 5.35-5.37 shows the temperature differences at these locations for the 250 kW burner fire. The difference between the full scale and $\frac{1}{8}$ scale temperature rise has increased to about 10-15°C.



Figure 5.35: Temperature Rise 45.7 cm Below Ceiling in Bay 4 (8.64m away from fire) for 250 kW Gas Burner.



Figure 5.36: Temperature Rise 45.7 cm Below Ceiling in Bay 10 (4.32m away from fire) for 250 kW Gas Burner.



Figure 5.37: Temperature Rise 45.7 cm Below Ceiling in Bay 19 (2.16m away from fire) for 250 kW Gas Burner.

While numerically greater, the temperature differences between scales for larger fires are still relatively close to the differences in temperature for small fires. When the difference is compared to the overall temperature rise, it is seen that the predictive capability of the scale model does not change with fire size.

The accuracy of the model does vary based on the distance from the fire. Some differences exist between the theoretical thermal properties of the boundaries and the actual thermal properties of the boundaries. This causes the heat transfer at the enclosure boundaries to be different between scales. Thermocouples located farther from the fire measure temperatures that have been greatly affected by the numerous partitions. Thermocouples close to the ceiling in bays far from the fire yielded the largest temperature difference between the full and model scale due to heat losses through the boundary materials. While not as accurate as thermocouple measurements closer to the fire, these results still produced values within 20°C of the full scale data. Figures 5.38-5.42 describe the temperature rise throughout the enclosure for the 250 kW fire. Note: The fire is located in Bay 16.



Figure 5.38: Temperature Rise 5.1 cm Below Ceiling in Bay 2 (10.08m away from fire) for 250 kW Gas Burner.



Figure 5.39: Temperature Rise 5.1 cm Below Ceiling in Bay 8 (5.76m away from fire) for 250 kW Gas Burner.



Figure 5.40: Temperature Rise 5.1 cm Below Ceiling in Bay 16 (above fire) for 250 kW Gas Burner.



Figure 5.41: Temperature Rise 5.1 cm Below Ceiling in Bay 17 (0.72m away from fire) for 250 kW Gas Burner.



Figure 5.42: Temperature Rise 5.1 cm Below Ceiling in Bay 19 (2.16m away from fire) for 250 kW Gas Burner.

The effect of the material boundaries is seen in these temperature differences. The heptane pool fires followed the same trend; the difference in boundary materials between scales affected the temperature as a function of distance from the fire. This is seen more clearly when the centerline temperatures at steady-state are compared. Figure 5.43 shows the steady-state temperature rise (measured 5.1 cm below the ceiling) in the full scale and the model scale for each instrumented bay in the large pool fire experiments. The fire is located at zero, with the negative distances spanning Bay 2 to Bay 15 and the positive distance spanning Bay 17 to Bay 19. From the differences in these two curves, it is clear that the accuracy of the scale model changes with distance. In other words, the interaction with numerous walls and partitions causes a difference between full and $\frac{1}{8}$ scale temperatures due to the scaled boundary materials. As noted in Section 4.3, the scaled density and conductivity of the full scale materials was difficult to match to existing materials. Π_{12} and Π_{13} were compromised slightly in order to construct the scale model. The hot gases in locations close to the fire have not been widely changed by the materials. However, far from the fire, the differences in the heat transfer properties of the material boundaries have affected the hot gas layer for the scale model.



Figure 5.43: Centerline Temperatures for Large Pool Fire at Steady State.

This is also observed for the gas burner, as seen in Figure 5.44. The temperature rise comparison shows the model is accurate close to the fire, but this accuracy decreases with distance from the fire due to material boundaires.



Figure 5.44: Centerline Temperatures for 250 kW Fire at Steady State.

Fuels

In general, the temperature results were more accurate in the gas burner experiments that in the heptane pool experiments. This is expected since the gas burner has a prescribed burning based on the mass flow of methane. The gas burner was influenced by human error since the mass flow meter used a manually operated value. The burning of heptane is impacted by the characteristics of the pan, the enclosure, and the water in the pan.

Natural Gas Burner

The gas burner was precisely designed to allow the total amount of prescribed flow to be burned evenly across the top surface. In the experiments, the burner was positioned and fixed in order to ensure proper fire locations in each test. It used a mass flow rate determined from the dimensionless heat release rate. This rate, and therefore the heat release rate of the fire, was accurately modeled for each fire size. As stated earlier, the 25 kW data was not used since a flame could not be sustained with such a low flow (0.214 L/min). The overall success of the gas burner scaling shows that this scaling methodology does work based on the assumptions made in Sections 3 and 4. The model generally resulted in close, but slightly lower temperatures than the full scale model. The difference in temperature rise between the full scale and the 1/8 scale models was a maximum of 10-15°C through all tests and fire sizes. This includes bays at the far end of the enclosure. Some of the slight disparity between scales could be a result of the actual fuel used. Natural gas typically consists of 70-90% methane, with the remainder being a mixture of ethane, propane, butane, carbon dioxide, and other trace gases [27]. Another factor that contributed to the difference in the ¹/₈ and full scale results is the thermal properties of the enclosure. The difference between the precise density, conductivity, and specific heat calculated by the scaling theory and the actual values of the material used in the scale model result in heat losses over distance. This

causes a greater temperature discrepancy in bays farther from the fire. For example, the thermocouples for the model in Bay 2 register temperatures 15° C lower than in the full scale. The flow of hot gases must traverse through 14 bays, causing significant heat losses from the slight differences in boundary materials between the full and $\frac{1}{8}$ scale models.

Flame Height Comparisons for the Natural Gas Burner

The flame height for the gas burner is modeled accurately for the 250 kW and 300 kW fires (larger fires), but the full scale flames are slightly taller than the $\frac{1}{8}$ scale flames for the 50 kW, 75 kW, and 150 kW fires (smaller fires). Figure 5.45 plots the dimensionless flame height with the dimensionless heat release rate, Q*, for the gas burner.



Figure 5.45: Dimensionless Flame Height vs. Q* for Gas Burner.

The flames of the smaller fires in the scale model were very faint and generally laminar (the plume remained turbulent). For a laminar flame, the scaling theory changes slightly. The flame height is proportional to \dot{Q} . In a turbulent flame, the flame height is proportional to $\dot{Q}^{2/5}$ [8]. This difference explains the difference in flame height between the model and full scale experiments. The pure methane also contributed to the lower flame height. Figure 5.46 shows the visual flame heights seen in the full scale and the $\frac{1}{8}$ scale model. Note that the model flame is difficult to see since it burns so cleanly.



Figure 5.46: Visual Flame Heights for 75 kW Burner in Full and ¹/₈ **Scale.** The actual flame height of the modeled gas burner may have been taller than reported here, but the faint blue flame was difficult to measure.

Steady State Temperature Comparisons for the Gas Burner

The transient temperatures for the gas burner experiments have been examined in Section 5.1.6.3. A comparison between the full scale and $\frac{1}{8}$ scale model at steady state is provided in Figure 5.47. The graph shows results for 50 kW, 150 kW, and 250 kW at four locations in the enclosure (Bays 2, 13, 16, and 17). The temperatures were averaged over a short period of time during steady state burning. This occurred at 500 seconds since it is during the steady state phase for the gas burner. A perfect scaling theory would result in identical temperatures between the full and $\frac{1}{8}$ scales. As seen in Figure 5.47, there is very good agreement between scales in all fires for thermcouples close to the fire. The $\frac{1}{8}$ scale temperature measurements from Bay 2 and Bay 13 are slightly off. The full scale temperatures are higher than the model temperatures at these locations. This is due to the differences in the boundary material thermal properties between the full scale and $\frac{1}{8}$ model. This can be directly related to the dimensionless group comparison in Section 4.3.1. The boudary Π groups for the partition material (OSB Chipboard) were both below one, meaning that the model boundary thermal properties are not as high as the full scale boundary thermal properties. This results in a difference in temperature measurements in the full and $\frac{1}{8}$ scale experiments.



Figure 5.47: Steady State Temperature Comparison for Gas Burner. *Heptane Pool Fire*

The ¹/₈ scale model heptane pools compared reasonably well to the full scale experiment temperature. The pans were constructed out of 0.32 cm thick steel. This may have had an effect on the temperature of the flame and the burning of the heptane due to the high conductivity of steel. The temperature rise in the scale model was slightly faster than the temperature rise in the full scale. This was consistent with each experiment. This was due to the sensitivity of the small gauge thermocouple wire that registered a change in temperature faster than the 28 AWG wire used in the full scale experiments. If a larger wire had been used in the ¹/₈ scale experiments, a similar temperature rise growth trend would have occurred between the two scales. The temperature also fluctuates more due to the change in burning rate as a function of time. Unlike the gas burner, the pool fire burning rate is prescribed by the amount of fuel, the thermal properties of the fuel, and the pan dynamics (material, presence of water, etc.). The results of the scale model are still considered a good representation of the full scale experiments.

The temperature discrepancy between models did not vary based on fire size. For the small and medium pool fires, the maximum difference between the small and full scale temperatures was less than 20°C (Bay 17). For the large pool fires, the maximum temperature difference was a factor of two larger, about 40°C (Bay 17). These differences are relative to the overall temperatures reached. Since the large pool fires measured higher overall temperatures, a greater difference between scales is expected. The relative temperature difference was similar for the small, medium, and large pool fires. This is an acceptable representation of the full scale data based on the error associated with the thermal properties of the boundary materials. The scale model generally predicted slightly lower temperatures than the full scale. There are some spikes in the $\frac{1}{8}$ scale data due to the fluctuating flame height and the fact that the model time was stretched to full scale by a factor of $\ell^{1/2}$.

The water level in the pan had a significant impact on the burn time, and therefore the mass loss rate, of the heptane fuel. Ideally, the water only provides the heptane with a level burning

surface in case of imperfections in the pan. However, burning liquid fuels on water changes the burning characteristics depending on the amount of water present. Unfortunately, there is no current reliable quantification of the effects of water in burning pools fires. Observations from this experimental series suggest that less water results in a faster burn time since there is less of a heat sink to the fire.

Flame Height for Heptane Pool Fire

The flame height for the pool fires is generally accurate for all three fire sizes. Figure 5.48 shows the dimensionless flame height against the dimensionless heat release rate, Q^* .



Figure 5.48: Dimensionless Flame Height vs. Q* for Heptane Pool Fire.

This plot displays all the trials from the full and $\frac{1}{8}$ scales. The Q* represented in this graph has been calculated based on experimental results. Note that the $\frac{1}{8}$ scale fires were designed by matching Q*. The amount of fuel and diameter of the pan were determined before the experiments. Since no load cell was used, the elapsed time and the amount of fuel in the pan were related to a mass loss rate of fuel. From this rough calculation, a heat release rate for the model was estimated. The slight changes in the Q* values are a result of variations of experiment length and possible human error while measuring the amount of fuel. The small and medium pool fire dimensionless heat release rates were very close to the expected values. The large pool fire had a difference of about 0.03 between the full scale Q* and the scale model Q*. While minor changes are apparent based on the Q* values, the flame height of the full scale experiments is accurately represented by the scaled model.

Steady State Temperature Comparisons for Heptane Pool Fire

The transient temperatures for the heptane pool fire experiments have been examined in Section 5.1.6.3. A comparison between the full scale and ½ scale model at steady state is provided in Figure 5.49. The graph shows results for small, medium, and large pool fires at four locations in the enclosure (Bays 2, 13, 16, and 17). The temperatures were averaged over a short period of time during steady state burning. This occurred at 150 seconds since it is during the steady state

phase for the heptane pool. A perfect scaling theory would result in identical temperatures between the full and ¼ scales. As seen in Figure 5.49, there is very good agreement between scales for most of the fire sizes and locations. Differences in the boundary material thermal properties between the full scale and ¼ model account for some discrpancies. This can be directly related to the dimensionless group comparison in Section 4.3.1.





In terms of fire investigation, not all heptane pool fires will be scaled this easily. It is important to note that the surface energy balance for larger turbulent pool fires are driven by radiative feedback [28]. This is very difficult to accurately model since feedback changes based on spatial orientation and temperature of the fire. It has been neglected here since smaller pool fires are convective driven. In cases with larger pool fires, radiation will be a key component in scale modeling and emissivity must be taken into account.

Consideration of Uncontrolled Independent Variables

The assumptions that are made when deriving the dimensionless groups and constructing the scale model do have an effect on the results presented here. The largest factor that was assumed negligent in the conservation equations was radiation. The two fuels used here reached steadystate quickly; flame spread and fire growth were not the main factors as they are with the wood crib and PU foam. Based on the fuels, the fires in the enclosure are convection driven fires. In the methodology presented here, radiation is intrinsically accounted for in the scaled heat release rate. This means that the thermocouples above the fire experience comparable radiation to the full scale.

The selected boundaries of the enclosure have an impact on the inside temperatures. In fact, the majority of the temperature discrepancies in this experimental series are related to the imperfect boundary materials. Density, specific heat, and conductivity were scaled independently, yet the availability of materials that meet such specific values limits the accuracy of boundary scaling. The scaled thickness of the boundaries also changes the temperature measured in the enclosure due to thermal penetration. In this research, the boundary materials had a significant

impact on temperature measurements far from the fire. This was magnified due to the numerous partitions that affected heat transfer within the enclosure. In every scale model, the limitations of finding the "perfect" boundary materials affect the results. The true art of scaling comes from minimizing the errors associated with such compromise.

The thermocouples were scaled to a 40 AWG wire in order to reduce error of having such a large wire in a ¹/₈ scale model. The wire diameter prescribed by the dimensionless group was not manufactured, so a slightly larger diameter wire was used. Thermocouples of all sizes have an associated error between the measured bead temperature and the actual gas temperature. This diminishes with smaller thermocouples, but still exists in 40 AWG wire. The thermocouples used in the ¹/₈ scale model measure a temperature very close to the actual gas temperature. This is one reason why the model temperature data fluctuates significantly more than the full scale temperature data. If similar flow time of the thermocouple, and therefore similar signal error, is desired in the model, the same size thermocouples are appropriate. Using 28 AWG thermocouples in this scale model may have resulted in a similar response time to temperatures. The sensitivity of the 40 AWG thermocouples caused some ¹/₈ scale results to have a steeper initial growth than the 28 AWG thermocouples in the full scale experiments.

6. Conclusions

This report is part of a larger research effort to provide fire investigators with the necessary tools to utilize scale modeling. The research examines four fuel sources; a gas burner, liquid pools, wood cribs, and polyurethane foam blocks. These fuels were selected to represent the various fuels that are found in fire investigations. This particular study examined the steady gas burner and liquid pool fires in a complex geometry.

Froude modeling was applied in a $\frac{1}{8}$ scale compartment. The design fires were convection driven; therefore radiation was neglected in this research. 40 AWG Type K thermocouples were used to record temperature as a function of time. The thermal response of the enclosure had a significant impact on the overall simulation results. The model was constructed using Kaowool and Marinite products. The conductivity, density, and specific heat of these materials differed slightly from the thermal properties calculated using the scaling theory. The temperatures recorded in the $\frac{1}{8}$ model were slightly lower than the full scale values due to this difference. The gas burner and the heptane pans were specially designed and constructed for this research. The burner was scaled geometrically and ensured an even dispersion of the gas. The heptane pans were made to match the Q* from the full scale experiments, which changed with the diameter of the pan.

Gas temperature scaling results for the natural gas burner and heptane pool fire were generally well scaled. Distance was a factor in scaling accuracy due to the material boundaries. The best results were directly above the fire ± 2 bays. The burner scaled better than the heptane pool due to the prescribed mass flow rate of the fuel. The transient temperature measurements showed similar trends for both fuels. Some discrepancy existed in the model because a more temperature sensitive thermocouple was used.

The heptane pool yielded excellent flame height results. The behavior, shape, color, and turbulence of the flame were also successfully scaled. The smaller gas burner fires produced laminar flames in the scale model, which resulted in a lower flame height. The larger gas burner experiments in the scale model produced flame heights that were in very good agreement with the full scale. Using pure methane instead of natural gas had some effect of the flame height. The steady state comparisons of temperature between the full and ¹/₈ scale models showed that the

scaling laws applied are very accurate close to the fire. In locations far from the fire, the full scale temperatures were higher than the scale model. This is due to the heat transfer differences in the boundary materials.

Future research will include the scaling of the pine wood crib and the PU foam blocks. Porosity, stick spacing, and stick thickness will play a major role in the scaling of the wood crib. The density of the foam, among other thermal properties, will most likely play a role while scaling the PU foam blocks since it impacts the flame spread velocity across the surface. It is important to note that these fuels produce dynamic fires where radiation and flame spread must be considered in the scaling methodology. Accurate wood crib scaling has been accomplished by Perricone [15], whose work would make an excellent starting point for future development on this project. The PU foam must be considered carefully with respect to flame spread. Dr. William Pitts at the National Institute of Standards and Technology has completed work with foam mattresses that would also benefit future research.

Understanding how to accurately model convection driven fires provides fire investigators with the tools to recreate many fire scenarios. It is especially helpful in fires leading up to flashover or where detection plays a major role in the investigation. The dimensionless groups presented in this thesis are an accurate method to model a full scale fire static fire. Future research will provide investigators with theory to create scale models of fires where flame spread and fire growth play key roles.

Appendix A

Dimensions and Geometry of Full Scale Enclosure (meters) Plan View: Outer Dimensions



Plan View: Elevation of Enclosure



Plan View: Partitions Throughout Enclosure



Side View: Partition Locations from Northeast Corner of Enclosure





Partition Geometry: Angle and Depth



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Instrumentation and Location in Full Scale Enclosure

Temperature: 28 AWG Type K Glass Insulated Thermocouples

- Vertical Thermocouple Trees: All Bays 5.1cm, 15.2 cm, and 45.7 cm below ceiling 2 m from East Wall Center of Bay
- Horizontal Thermocouple Trees: Bays 2, 4, 10

0.3 m below ceiling 15 thermocouples; 1 TC every 0.31 m Center of Bay

Velocity: Hot Wire Anemometer
Bays 2, 6, 13, 19
15.2 cm below Ceiling
1.8 m from East Wall
15.2 cm from South Side

Obscuration: Optical Density Meter
Bays 2, 6, 13, 19

Attached to Ceiling (East-West) 2.3 m from East Wall Centered in bay Detection: Ionization and Photoelectric Smoke Detectors

Bays 2, 6, 13, 19
 Three Detectors Attached to Ceiling (North-South)
 1.5 m from East Wall
 Order: Ionization, Photoelectric, Ionization
 Centered in bay



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Article IV. Results of Scaling Fire Growth Scale Modeling of Fire Growth in A Furnished Room

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Abstract. Scale modeling of fire requires the measurements and observations made at a smaller scale than the actual event be related by scaling laws. Due to the complex and multivariable nature of fire, it is not possible to get complete scaling. This study pushed the envelope on scaling, and successfully produced results, both qualitative and quantitative, in good agreement between full scale and a quarter scale model. The scenario was a bedroom fire initiated by a wastebasket fire igniting the edge of bedding on a bed and the side of an upholstered chair. The fire then spreads away from this point. The basic scaling methodology was to use the same material at the same thickness as possible, but overall dimensions were 1/4. Theory of the relevant dimensionless groups showed how similarity would not totally be maintained, but gave an indication of the differences. These differences are substantiated in the results. Instrumentation measuring temperatures, heat flux, gas species (O₂, CO₂ and CO), plus extensive video records inside and outside display the phenomena and results. These will be presented. The results also provide an anatomy of the flashover process leading to a ventilation-limited fire. Explaining these effects through video and measurements is very revealing and noteworthy for fire modelin

Full-scale experiments

On July 16, 2008, three experiments were conducted to evaluate the effects of limited ventilation in compartment fires. The experiments were conducted in rooms, which were furnished with typical bedroom furniture. The ignition source used during the three experiments was an open flame to a trashcan filled with crumpled newspaper. The compartments were instrumented to measure temperature, gas specie concentration, heat flux, smoke detection activation time, CO detection activation time, and air velocity. The test program was conducted in the large burn room of the ATF Fire Research Laboratory (FRL) in Beltsville, MD.

Test Setup

The dimensions of the three test compartments used during the test series is shown in Figure 1. The test compartments all had ceiling heights of 2.44 m (96 inch).



Figure 1. Full-scale test compartment plan view.

Test Compartment Construction Details

The test compartments were built on top of platforms which were constructed of 2 x 6 dimensional lumber spaced 0.41 m (16 inch) on center. The top of the platforms, which made up the floors of the test compartments, were sheathed with 0.02 m (0.75 inch) thick plywood. The walls of the test compartments were framed using 2 x 4 dimensional lumber spaced 0.41 m (16 inch) on center. The test compartment ceiling joist were constructed using 2 x 6 dimensional lumber, spaced 0.41 m (16 inch) on center.

Test Compartment Interior Finish Details

All interior walls and ceilings were sheathed with one layer of 0.01 m (0.5 inch) thick gypsum wall board. The joints of the walls and ceilings were sealed with one layer of gypsum wall board tape and two layers of gypsum wall board joint compound. The walls were painted with two coats of interior latex flat white paint (Glidden Professional Finishes, Ultra Build).

The plywood flooring in the compartments was covered with one layer of 0.01 m (0.4375 inch) thick, "6-Pound" carpet padding and one layer of 0.02 m (0.75 inch) thick carpet. The carpet was, plush carpet, style name GAME CHANGER, color name GREEN ISLE.

The door located on the East wall of all three test compartments was an interior hollow core wood door, which measured 2.03 m tall, 0.91 m wide and 0.04 m thick ($80 \times 35.75 \times 1.4375$ inch).

Furniture Details

The three test compartments were furnished with the following pieces of bedroom furniture;

- Queen size bed, with wooden headboards and footboards,
- Nightstand,
- Wide dresser with mirror,
- Tall dresser,
- Wing-back chair
- Plastic waste basket

Figure 1 shows the placement of the furniture in the test compartments. The x-y axis in Figure 2 shows the origin from which furniture locations were measured. The black dots at the corner of each piece of furniture show where the x-y coordinates for the furniture pieces were measured from the origin. The x-y coordinates for the furniture pieces are shown in Table 1. The weight and dimensions of the furniture pieces are detailed in Table 2.



Figure 2. Furniture placement inside the test compartments.

Item	Location			
	X		У	
	[m]	[inch]	[m]	[inch]
Tall Dresser	0.38	17	3.30	130
Queen Size	0.86	34	2.11	83
Bed				
Wide Dresser	1.07	42	0.45	17.75
Nightstand	3.21	126.25	2.55	100.5
Wing Back	2.79	110	3.79	149.5
Chair				
Waste basket	2.54	100	4.07	160.25

Table 1. Full-scale Furniture placement coordinates.

Item	Weight		Dimensions							
	[kg]	[pound]	Length		Width		Height		Thickness	
			[m]	[inch]	[m]	[inch]	[m]	[inch]	[m]	[inch]
Nightstand	14.9	33.0	0.56	22	0.42	16.5	0.67	26.5	N/A	N/A
Wide Dresser – Mirror	11.2	124.8	N/A	N/A	0.85	33.625	1.22	46.875	0.02	0.875
Unit										
Wide Dresser –	44.4	102.4	1.56	61.5	0.43	16.75	0.90	35.375	N/A	N/A
Dresser										
Tall Dresser	41.4	91.3	0.91	36	0.42	16.5	1.32	52.125	N/A	N/A
Bed – Headboard	12.1	26.6	1.62	63.875	N/A	N/A	1.23	48.5	0.04	1.75
Bed – Footboard	10.5	23.2	1.62	63.875	N/A	N/A	0.86	34	0.04	1.5
Bed – Mattress	24.5	54.0	2.00	79.0	1.52	60.0	N/A	N/A	0.18	7
Bed – Box Spring	20.2	44.6	2.00	79.0	1.52	60.0	N/A	N/A	0.23	9
Wing Back Chair	22.6	49.8	0.69	27	0.63	25	1.08	42.5	N/A	N/A

Table 2. Full-scale Furniture details.

Note: N/A = Not Applicable.

Mattress and Box Spring Details

The mattresses and Box springs used during the test series were both manufactured by Symbol Mattress. All mattresses and box springs used during the test series were manufactured in 2008 and meet the requirements of 16 CFR Part 1633 (federal flammability open flame). The outer material of the mattresses were constructed with a 25% Resinated Textile Fiber PD / 75% Urethane Foam material. The wooden foundations of the box springs were covered using a 25% Corrugated Fiber Board / 75% Urethane Foam material.

Bed Linens

Two layers of 100% polyurethane foam padding, which measured 1.422 m x 1.956 m x 0.038 m (56 x 77 x 1.5 inch) were placed directly on top of the mattress. The foam padding was manufactured by Mainstays and was 100% Visco Memory Foam. The foam padding was covered with one queen fitted sheet which was covered with one queen flat sheet. The fitted and flat sheets were both 130 thread count sheets and were made of a 55% cotton / 45% polyester material. A full/queen size quilt was placed over the flat sheet. The quilts were manufactured by Peking Handcraft, INC. and were made of a 60% cotton / 40% polyester material. Two pillows covered with pillow cases were placed at the head of the bed. The outside cover of the pillows was made of a 100% cotton material, and the pillow filling was made of a 100% polyester material. The pillow cases were made of a 55% cotton / 45% polyester material.

Instrumentation

The test compartments were instrumented to measure the gas specie concentration at four different locations. Each gas specie concentration location measured the oxygen concentration (O_2) , carbon-monoxide concentration (CO), and carbon-dioxide concentration (CO_2) . The test compartments were also instrumented to measure the temperature, incident heat flux, air velocity, smoke detector activation time, and CO detector activation time. The locations of the instrumentation measurements are shown in Figure 3. The x-y axis in Figure 3 shows the origin from which instrument locations were measured.



Figure 3. Full-scale Instrumentation details.

Experiment Details

Door Position

As shown in Figure 1, the doors were opened to their maximum open position during all three experiments. During Experiment 2, the door closed two times and was reopened each time during the experiment. To prevent the same thing from occurring during Experiment 3, screws were inserted into the test compartment floor directly next to the door.

Ignition Source

The ignition source used during all three experiments consisted of ten full pages of loosely crumpled newspaper placed in the waste basket. The newspaper was ignited using a butane lighter.

Flashover Definition

For the purposes of the full-scale test series, "the flashover transition period" was defined as sustained flames venting out of the door opening for ten (10) seconds.

Test Procedure

The crumpled newspaper was lit at the top of the waste basket using a butane lighter. After sustained ignition, all personnel exited the test compartment. During Experiment 1, the fire was allowed to grow up to the flashover transition period and was extinguished immediately upon flashover using a water hand line. During Experiment 2, and Experiment 3, the fire was allowed to burn two minutes past the flashover transition period and was suppressed using a water hand line.

Comparison full-scale experiment 1 and 3

Experiment Events

The following table lists selected events that occurred during the experiment that may have affected the measured results.

Table 3. Full-scale Experiment Events Test 1

Event Description	Time (seconds)
suppression	209

Table 4. Full-scale Experiment Events Test 3

Event Description	Time (seconds)
suppression	361

Thermocouples

Thermocouples are temperature measurement sensors that consist of two dissimilar metals joined at one end (a junction) that produces a small thermo-electrical voltage when the wire is heated. The change in voltage is interpreted as a change in temperature. [1] There are many configurations of thermocouples which affect the temperature range, ruggedness, and response time. The information required to identify these factors for the thermocouples used in this experiment are provided in the "Thermocouple Descriptions" table in this section of the report. Thermocouples were used in this experiment in accordance with the method defined in FRL laboratory instruction "LI001 Thermocouples".

The following table provides a description of the instrumentation used to collect the temperature measurements during the experiments. The "Description" column describes the location of the temperature measurement. When thermocouples are mounted in a vertical configuration the "Tree" column identifies the thermocouple tree by number. The "Z" location is the height of the thermocouple above the floor. The "Thermocouple Type" describes the characteristics of the thermocouple used.

Description	Tree	X (meters)	Y (meters)	Z (meters)	Thermocouple Type
TC-B8	1	1.83	1.73	2.44	Type K, Glass Ins., 24 AWG wire
TC-B7	1	1.83	1.73	2.13	Type K, Glass Ins., 24 AWG wire
TC-B6	1	1.83	1.73	1.83	Type K, Glass Ins., 24 AWG wire
TC-B5	1	1.83	1.73	1.52	Type K, Glass Ins., 24 AWG wire
TC-B4	1	1.83	1.73	1.22	Type K, Glass Ins., 24 AWG wire
ТС-ВЗ	1	1.83	1.73	0.91	Type K, Glass Ins., 24 AWG wire
TC-B2	1	1.83	1.73	0.61	Type K, Glass Ins., 24 AWG wire
TC-B1	1	1.83	1.73	0.30	Type K, Glass Ins., 24 AWG wire
ТС-В0	1	1.83	1.73	0.00	Type K, Glass Ins., 24 AWG wire
TC-C8	2	0.06	1.70	2.44	Type K, Glass Ins., 24 AWG wire
TC-C7	2	0.06	1.70	2.13	Type K, Glass Ins., 24 AWG wire
TC-C6	2	0.06	1.70	1.83	Type K, Glass Ins., 24 AWG wire
TC-C5	2	0.06	1.70	1.52	Type K, Glass Ins., 24 AWG wire
TC-C4	2	0.06	1.70	1.22	Type K, Glass Ins., 24 AWG wire
TC-C3	2	0.06	1.70	0.91	Type K, Glass Ins., 24 AWG wire
TC-C2	2	0.06	1.70	0.61	Type K, Glass Ins., 24 AWG wire
TC-C1	2	0.06	1.70	0.30	Type K, Glass Ins., 24 AWG wire
TC-C0	2	0.06	1.70	0.00	Type K, Glass Ins., 24 AWG wire

 Table 5. Full-scale Thermocouple Descriptions

The following tables provide a summary of the temperature results for full-scale test 1 and test 3. The "Initial Temperature" column provides the measured temperature at the beginning of the test. The maximum temperature recorded during the test is provided in the "Maximum" column. The remaining columns provide the calculated maximum average temperatures.

r	1	1	r	1	T	1
	Initial		30 second	1 minute	5 minute	10 minute
	Temperature	Maximum	maximum	maximum	maximum	maximum
Description	(°C)	(°C)	average (°C)	average (°C)	average (°C)	average (°C)
TC-B8	30	900	884	835	277	154
TC-B7	27	927	906	880	299	163
TC-B6	27	934	904	878	281	154
TC-B5	27	915	886	852	259	143
TC-B4	27	890	876	828	240	133
TC-B3	27	822	803	727	200	113
TC-B2	26	696	590	452	128	77
TC-B1	26	573	465	345	102	64
TC-B0	26	219	192	167	65	46
тс-с8	27	817	739	674	279	153

 Table 6. Temperature Value Result Summary Full-scale Test 1

	Initial		30 s	second	1 m	ninute	5	minute	10	minute
	Temperature	Maximum	maximum		maximum		maximum		maximum	۱
Description	(°C)	(°C)	average (°	C)	average (°C	C)	average (°C)	average (°C)
TC-C7	27	791	733		705		299		163	
TC-C6	27	751	738		719		266		146	
TC-C5	27	764	744		720		257		142	
TC-C4	27	783	735		710		246		136	
TC-C3	26	812	679		632		206		116	
TC-C2	26	643	549		477		140		83	
TC-C1	26	551	272		199		75		50	
TC-C0	26	676	353		269		108		67	

Table 7. Temperature Value Result Summary Full-scale Test 3

	Initial		30 secon	d 1 minute	5 minute	10 minute
Description	Temperature (°C)	Maximum (°C)	maximum average (°C)	maximum average (°C)	maximum average (°C)	maximum average (°C)
TC-B8	31	881	852	833	580	316
TC-B7	30	897	862	849	598	325
TC-B6	30	913	872	859	582	314
TC-B5	30	898	865	847	560	302
TC-B4	30	877	834	802	553	297
TC-B3	29	889	835	800	503	269
TC-B2	29	810	773	737	480	256
TC-B1	29	837	798	734	449	240
TC-B0	29	807	756	710	405	219
TC-C8	29	829	785	760	542	293
TC-C7	30	851	800	763	563	303
TC-C6	29	841	779	755	571	310
TC-C5	29	825	769	752	562	303
TC-C4	29	847	793	770	546	293
TC-C3	29	829	792	754	503	269
TC-C2	29	845	809	767	465	249
TC-C1	29	880	804	762	439	235
TC-C0	29	866	782	737	414	223

Temperature at the Center of the Room





Temperature at the Door



Figure 5. Temperature @ B Full-scale Test 3





Figure 7. Temperature @ C Full-scale Test 3

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Heat Flux Transducers

A heat flux transducer is a device that measures the rate of absorbed incident energy, and expresses it on a per unit area basis. The operating principle of the Schmidt-Boelter heat flux transducer(s) used in this experiment is based on one-dimensional heat conduction through a solid. Temperature sensors are placed on a thin, thermally conductive sensor element, and applying heat establishes a temperature gradient across the element. The heat flux is proportional to the temperature difference across the element according to Fourier's Law [2].

There are many configurations of heat flux transducers which affect range, size, mode and sensitivity. The information required to identify these factors for the heat flux transducer(s) used in this experiment are provided in the "Heat flux measurement descriptions" table in this section of the report.

Heat flux transducers were used in accordance with the method defined in FRL laboratory instruction "LI002 Heat Flux Transducer".

The following table provides a description of the instrumentation used to collect heat flux measurements during the experiments. The "Description" column describes the location of the heat flux measurement. "Height" is the distance from the floor to the centerline of the transducer. If the heat flux measurement has to be discontinued during a test the "Out of Service Time" and "Out of Service Reason" columns report the test time and reason why the heat flux measurement was removed.

Table 8	. Heat flux	measurement	descriptions
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				Out	of	Service
Description	Height (meters)	Flux mode	Out of Service Time (seconds)	Reason		
THF-B	1.22	Total				
THF-C	1.22	Total	351 (Test 3)	OverRa	nge	

The following table provides a summary of the heat flux results. The time at which the heat flux first changed by a pre-determined amount is provided in the "Time of Initial Change" column. The pre-determined amount of change in heat flux is provided in the "Initial Change Amount" column. The maximum heat flux recorded during the test is provided in the "Maximum" column. The maximum average columns are calculated from the area under the measured heat flux data over a specified time span.

 Table 9. Heat flux result summary Test 1

	Time of	Initial		30 second	1 minute	5 minute	10 minute
	Initial	Change		maximum	maximum	maximum	maximum
	Change	Amount	Maximum	average	average	average	average
Description	(sec)	(kW/m²)	(kW/m²)	(kW/m²)	(kW/m²)	(kW/m²)	(kW/m²)
THF-B	76	5	191.1	143.8	107.7	30.4	15.2
THF-C	121	5	77.9	67.0	64.1	17.9	9.0



The following chart(s) present a time dependent representation of the instantaneous heat flux measured during the experiment.

Figure 8. Heat Flux Full-scale Test 1

Figure 9. Heat Flux Full-scale Test 3

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Gas Analyzer-Paramagnetic-O2

A gas analyzer was used to measure the oxygen (O₂) concentration at point measurement locations. The oxygen analyzer operates according to the paramagnetic alternating pressure principal. The O₂ analyzer has an output signal with a repeatability and linear deviation of <1% of the respective span value. The analyzer was zeroed and calibrated prior to each test. Nitrogen was used as the zero gas, and dried ambient air, which is assumed to have an oxygen concentration of 20.95 %, was used as the span gas.

When gas concentration point measurements are collected in a compartment, the gas samples are returned to the same compartment after passing through the analyzers. The return location is chosen such that it will not effect the air movement near the fire.

The gas concentration point measurements were conducted in accordance with the method defined in FRL laboratory instruction "Gas Concentration Measurements".

The following table provides information about the oxygen sampling location and the operating parameters of the meter. The "Oxygen delay time" is the time for the oxygen analyzer output to achieve 90% of its final value when subjected to a step change at the measuring location

Table 10. Oxygen measurement descriptions - full-scale tests

Description	Location X (meters)	Location Y (meters)	Location Z (meters)	Delay Time Oxygen (s)
GS-B	3.48	3.48	2.22	22

The following table provides a summary of the oxygen measurement results.

Table 11. Oxygen Measurement Results Full-scale Test 1

Description	O2 Analyzer Full Scale Range (mole fraction)	Oxygen Peak Minimum (mole fraction)
GS-B	0.25	0.0003

Table 12. Oxygen Measurement Results Full-scale Test 3

Description	O2 Analyzer Full Scale Range (mole fraction)	Oxygen Peak Minimum (mole fraction)
GS-B	0.25	-0.0003



The following chart presents the oxygen concentration(s) measured during the test.

Figure 10. Oxygen Concentration Full-scale Test 1

Figure 11. Oxygen Concentration Full-scale Test 3

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Gas Analyzer-NDIR-CO/CO₂

A gas analyzer was used to measure both the carbon monoxide (CO) and carbon dioxide (CO₂) concentrations at point measurement locations. The CO/CO₂ gas analyzer is of the NDIR type and has separate output signals with resolutions of < 0.1% of the output signals span value for each gas concentration measurement. The analyzer was set to ranges of 0 - 5% CO and 0 - 25% CO₂ during the tests. The analyzer were zeroed and spanned prior to each test. Nitrogen was used as the zero gas and a pre-mixed calibration gas with known concentrations of CO and CO₂ were used as the span gas.

When gas concentration point measurements are collected in a compartment, the gas samples are returned to the same compartment after passing through the analyzers. The return location is chosen such that it will not affect the air movement near the fire.

The gas concentration point measurements were conducted in accordance with the method defined in FRL laboratory instruction "Gas Concentration Measurements".

The following table provides information about the carbon monoxide and carbon dioxide sampling location and the operating parameters of the meters. The "CO/CO2 delay time" is the time for the gas analyzer output to achieve 90% of its final value when subjected to a step change at the measuring location

Table 13. CO and CO2 measurement descriptions – full-scale tests

Description	Location X (meters)	Location Y (meters)	Location Z (meters)	Delay Time CO/CO2 (s)
GS-B	3.48	3.48	2.22	22

The following table provides a summary of the carbon monoxide gas measurement results.

Table 14. CO Measurement Results Full-scale Test 1

	CO	Analyzer	Full	Scale	СО	Span	Gas	Value	Gas	Concentr	ation	- peak	carbon
Description	Rang	ge (mole fra	ction)		(mol	e fracti	on)		mon	oxide (mo	le frac	tion)	
GS-B	0.05				0.01				0.05	32			

Table 15. CO Measurement Results Full-scale Test 3

	CO	Analyzer	Full	ScaleCO) Span	Gas	Value	Gas	Concer	ntration	- peak	carbon
Description	Rang	ge (mole fra	ction)	(m	ole fract	ion)		mond	oxide (m	nole fra	ction)	
GS-B	0.05			0.0	1			0.053	32			

The following table provides a summary of the carbon dioxide gas measurement results.

Table 16. CO2 Measurement Results Full-scale Test 1

	CO2 /	Analyzer	Full	Scale	CO2 \$	Span	Gas	Value	Gas Concentration -	peak	carbon
Description	Range	(mole frac	tion)		(mole t	fractio	n)		dioxide (mole fraction))	
GS-B	0.10				80.0				0.1795		

Table 17. CO2 Measurement Results Full-scale Test 3

	CO2	Analyzer	Full	Scale	CO2	Span	Gas	Value	Gas Concentra	ation - peal	carbon
Description	Range	e (mole frac	tion)		(mole	fractio	n)		dioxide (mole f	raction)	
GS-B	0.10				80.0				0.2071		



The following charts present the carbon monoxide concentration(s) measured during the test.

Figure 12. Carbon Monoxide Concentrations Full-scale Test 1

Figure 13. Carbon Monoxide Concentrations Full-scale Test 3



The following charts present the carbon dioxide concentration(s) measured during the test.

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Figure 14. Carbon Dioxide Concentrations Full-scale Test 1

Figure 15. Carbon Dioxide Concentrations Full-scale Test 3

Quarter-scale experiments

On May 26, 2011, three experiments were completed to examine quarter-scale fire at 1:4 scale. The rooms and furniture were similar to the original rooms but constructed at 1:4 scale. Wherever possible, all of the furnishings were constructed using the full-scale material thickness with a 1:4 scale overall dimension. A similar mattress, box spring set, linens, pillows and a wingback chair used in the original tests were purchased. The materials from these items were used to construct the items burned in the 1:4 scale tests. The remaining furniture was constructed from dimensional lumber. The construction materials used in the 1:4 scale tests, including the carpet, padding and paint, were also similar to the materials used in the original full scale tests. The ignition source used during the three 1:4 scale tests was an open flame to a plastic cup filled with 1/2 sheet of crumpled newspaper. The compartments were instrumented to measure temperature, gas specie concentration and heat flux. These tests were conducted in the medium burn room of the ATF FRL in Beltsville, MD.

Section 4.01 Test Setup

The dimensions of the three test compartments used during the test series are shown in Figure 16. The test compartments all had ceiling heights of 0.61 m (24.0 inch). The test compartments measured 1.066 m length, 0.91 m width (42×36 inches). The test compartments were constructed on top of 0.02 m (0.75 inch) thick plywood. The walls of the test compartments were framed using 2 x 4 dimensional lumber spaced 0.41 m (16 inch) on center. The test compartment ceiling joist were constructed using 2 x 4 dimensional lumber, spaced 0.41 m (16 inch) on center. The completed test compartments were then secured to wooden pallets. Test cells #1 and #2 were constructed with a door opening measuring 0.22 m wide, 0.5 m height (9 x 20 inch).

Test cell #3 was constructed with a door opening measuring 0.45 m wide, 0.5 m height (18x 20 inch). This cell was altered to compare to previous scale modeling theories.



Figure 16. Quarter-scale test compartment plan view.

Test Compartment Interior Finish Details

All interior walls and ceilings were sheathed with one layer of 0.01 m (0.5 inch) thick gypsum wall board. The joints of the walls and ceilings were sealed with latex caulking. All exposed screw heads were covered with wall board joint compound. The walls were painted with two coats of interior latex flat white paint (Glidden Professional Finishes, Ultra Build).

The plywood flooring in the compartments was covered with one layer of 0.01 m (0.5 inch) thick "6 -Pound" carpet padding and one layer of 0.02 m (0.75 inch) thick plush carpet, style name GAME CHANGER.

The doors of the Test 1 and Test 3 compartments were constructed of luan plywood to replicate an interior hollow core wood door, which measured 0.51 m tall, 0.23 m wide and 0.006 m thick (20 x 9 x 0.25 inch). In test 2 the width of the door opening was altered and the door removed as a comparison to scale model testing conducted in the past.

Furniture Details

The three test compartments were furnished with 1:4 scale models of the following pieces of bedroom furniture;

- Queen size bed, with wooden headboards and footboards,
- Nightstand,
- Wide dresser with mirror,
- Tall dresser,
- Wing-back chair
- Plastic waste basket 16 oz. Solo TM, plastic cup, filled with ¹/₂ sheet crumpled newspaper

Figure 17 shows the origin from which furniture locations were measured. The black dots at the corner of each piece of furniture show where the x-y coordinates for the furniture pieces were measured from the origin. The x-y coordinates for the furniture pieces are shown in Table 18. The furniture details are shown in Table 19.



Figure 17. Furniture placement inside the compartments

Table 18. Quarter-scale furniture placement coordinates

Item	Location	
	X	У

	[m]	[inch]	[m]	[inch]
Tall Dresser	0.11	4.25	0.83	32.5
Queen Size	0.22	8.5	0.53	20.75
Bed				
Wide Dresser	0.27	10.5	0.11	4.25
Nightstand	0.80	31.5	0.64	25.0
Wing Back	0.70	27.5	0.95	37.25
Chair				
Waste basket	0.64	25.0	1.02	40.0

Table 19. Quarter-scale furniture details

Item	Length		Width		Height	
	[m]	[inch]	[m]	[inch]	[m]	[inch]
Nightstand	0.14	5.5	0.10	4.0	0.17	6.5
Wide Dresser – Mirror	0.21	8.25	0.02	0.75	0.33	13.0
Unit						
Wide Dresser –	0.39	15.5	0.10	4.0	0.22	8.75
Dresser						
Tall Dresser	0.23	9.0	0.10	4.0	0.33	13.0
Bed – Headboard	0.41	16.0	0.02	0.75	0.30	12.0
Bed – Footboard	0.41	16.0	0.02	0.75	0.22	8.5
Bed – Mattress	0.50	19.5	0.38	15.0	0.07	2.75
Bed – Box Spring	0.50	19.5	0.38	15.0	0.07	2.25
Wing Back Chair	0.17	6.75	0.16	6.25	0.30	12.0

Section 4.02

Mattress and Box Spring Details

The mattresses and box springs used during both test series were manufactured by Symbol Mattress. The mattress and box springs used during the test series were manufactured in 2010 and meet the requirements of 16 CFR Part 1633 (federal flammability open flame). The outer materials of the mattresses were constructed with a 24% Blended Fiber Batting / 76% Urethane Foam material. The wooden foundations of the box springs were covered using a 25% Corrugated Fiber Board / 75% Urethane Foam material.

The mattresses used in the 1:4 scale testing were constructed from material taken from the full scale item. The mattress was cut open and the metal springs were removed. The remaining material was shaped into a mattress at 1:4 scale to the original and held together with hot glue and safety pins.

The box springs used in the 1:4 scale testing were constructed from materials removed from the full scale item. The materials were stapled onto a wooden frame constructed at 1:4 scale to the original.

Bed Linens

One layer of 100% polyurethane foam padding, which measured 0.38 m x 0.48 m x 0.04 m (15 x 19 x 1.5 inch) was placed directly on top of the mattress. The foam padding was manufactured by Mainstays and was 100% Visco Memory Foam. The foam padding was covered with one fitted sheet which was covered with one flat sheet. The fitted and flat sheets were both 200 thread count sheets and were made of a 60% cotton / 40% polyester material. A quilt was placed over the flat sheet. The quilts were manufactured by Better Homes and Gardens and were made of a 52% cotton / 48% polyester material. Two pillows covered with pillow cases were placed at the head of the bed. The outside cover of the pillows was made of a 100% cotton material, and the pillow filling was made of a 100% polyester material. The pillow cases were made of 52% cotton / 48% polyester material. All of the bed linens and pillows used in the 1:4 scale testing were similar to the materials used in the full scale testing.

The bed linens and padding used in the 1:4 scale testing were cut from the original materials. The pillows used in the 1:4 scale testing were cut from the original material and held together with hot glue.

Wing-Back Chair

The wing-back chair used during both test series was manufactured by Lane Furniture Industries. The chairs were manufactured in June, 2010. The wooden frame of the chair was covered with materials consisting of 49% Acrylic/ 30% Polyester/ 21% Cotton.

The chairs used in the 1:4 scale testing were constructed using material from the original item. Polyurethane padding and cloth fabric were stretched over a 0.006 m thick (0.25 inch) luan plywood frame and held in place with staples and hot glue.

Bed Frame, Wide Dresser, Tall Dresser and Nightstand

All of the furniture was all constructed of 0.02m (0.75 inch) thick dimensional pine lumber. All of the furniture was covered with two coats of Minwax gloss polyurethane finish.

Instrumentation

The 1:4 scale test compartments were instrumented to measure the gas specie concentration at one location. This gas specie concentration location measured the oxygen concentration (O_2), carbon-monoxide concentration (CO), and carbon-dioxide concentration (CO_2). The test compartments were also instrumented to measure temperatures and heat flux. The locations of the instrumentation measurements are shown in Figure 3. The x-y axis in Figure 3 shows the origin from which instrument locations were measured.



Figure 18. Quarter-scale instrumentation details.

Table 20. Quarter-scale instrumentation plac	cement details
--	----------------

Item	Location					
	У		Z			
	[m]	[inch]	[m]	[inch]		
THF-B	0.86	34.25	0.30	12		
THF-C	0.26	10.5	0.30	12		
GSB	0.86	34.25	0.55	22		
Item	Location					
	X		У			
	[m]	[inch]	[m]	[inch]		
TC-B	0.45	18	0.43	17		
TC-C	0.01	0.5	0.42	16.75		

Experiment Details

Door Position

As shown in Figure 1, the doors were opened to their maximum open position during tests one and three. The door was removed for test two.

Ignition Source

The ignition source used during all three 1:4 scale tests consisted of one half (1/2) page of loosely crumpled newspaper placed in a 16 oz. SoloTM, plastic cup to replicate the plastic wastebasket used in the original tests. The newspaper was ignited using a butane lighter through a hole cut in the drywall directly above the cup.

Test Procedure

The crumpled newspaper was lit at the top of the plastic cup using a butane lighter. After sustained ignition, the hole was plugged with a piece of drywall and sealed with latex caulking. During tests 1 and 3 the fire was allowed to grow to the flashover transition period and was extinguished immediately upon reaching flashover using two sprinkler heads mounted in the ceiling. During Test 2, the fire was allowed to burn two (2) minutes past the flashover transition period and was suppressed using two sprinkler heads mounted in the ceiling.

Flashover Definition

In all of the 1:4 scale tests "Flashover" was defined as sustained flames venting out of the door opening. In all of the original full scale tests the fire was extinguished ten (10) seconds after sustained flames were observed venting out of the doorway openings.

Comparison of quarter-scale to full-scale

Comparison full-scale experiment 1 and quarter-scale experiment 1

Experiment Events

The following table lists selected events that occurred during the experiment that may have affected the measured results.

Table 21. Experiment Events Full-scale Test 1

Event Description	Time (seconds)
suppression	209

Table 22. Experiment Events Quarter-scale Test 1

Event Description	Time (seconds)
suppression	<mark>???</mark>

Thermocouples

Temperature at the Center of the Room



Figure 19. Temperature at the center of the room FS1 v QS1



Figure 20. Temperature at the door FS1 v QS1

Heat Flux Transducers



Figure 21. Heat Flux FS1 v QS1

Gas Species: O₂, CO, CO₂



Figure 22. Gas Species Concentrations FS1 v QS1

Comparison full-scale experiment 1 and quarter-scale experiment 3

Experiment Events

The following table lists selected events that occurred during the experiment that may have affected the measured results.

Table 23. Experiment Events Full-scale Test 1

Event Description	Time (seconds)
suppression	209

Table 24. Experiment Events Quarter-scale Test 3

Event Description	Time (seconds)
suppression	<mark>???</mark>

Thermocouples





Figure 23. Temperature at the center of the room FS1 v QS3



Figure 24. Temperature at the door FS1 v QS3

Heat Flux Transducers



Figure 25. Heat Flux FS1 v QS3

Gas Species: O₂, CO, CO₂



Figure 26. Gas Species Concentrations FS1 v QS3

Comparison full-scale experiment 3 and quarter-scale experiment 2

Experiment Events

The following table lists selected events that occurred during the experiment that may have affected the measured results.

Table 25. Experiment Events Full-scale Test 3

Event Description	Time (seconds)
suppression	<mark>???</mark>

Table 26. Experiment Events Quarter-scale Test 2

Event Description	Time (seconds)
suppression	<mark>???</mark>

Thermocouples

Temperature at the Center of the Room



Figure 27. Temperature at the center of the room FS3 v QS2



Figure 28. Temperature at the door FS3 v QS2

Heat Flux Transducers



Figure 29. Heat Flux FS3 v QS2

Gas Species: O₂, CO, CO₂



Figure 30. Gas Species Concentrations FS3 v QS2
Video Snapshots



Figure 31. Early Ignition Snapshot – Fires are similar



Figure 32. Early Growth - Faster lateral growth in quarter scale



Figure 33. Fire Growth - continued faster growth in quarter scale



Figure 34. Fire Growth - Onset of rapid growth across chair in quarter scale



Figure 35. Fire Growth - Onset of rapid growth across chair in full scale



Figure 36. Fire Growth - Quarter scale chair fully involved



Figure 37. Fire Growth - Full scale chair fully involved



Figure 38. Fire Growth - Full scale is beginning to pass quarter scale fire growth



Figure 39. Fire Growth - Full scale surpassed growth of quarter scale. Flames in ceiling layer of full scale



Figure 40. Fire Growth - Full scale flames moved away from corner



Figure 41. Fire Growth - quarter scale flames moved away from corner. Full scale flames moved to vent opening.



Figure 42. Fire Growth - quarter scale flames moved to vent opening.

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Article V. Conclusions

A review of fire scale modeling has been presented. The literature has been reviewed, and the basis for modeling rules has been presented. Hopefully the reader will benefit from this background and methodology so that it can be further applied. The benefits to scale modeling are in design, investigation, and education of fire phenomena. WE have investigated two scenarios. One easy for scaling and the other difficult, if not some might say impossible. Yet the results show promise and value. Used with understanding scale modeling can be an important tool in the understanding of fire. It is simpler and less costly than full-scale testing.

Article VI. Dissemination of Findings

The conduct of this project has allowed collaboration with the ATF Fire Laboratory and the ATF fire investigators. Presentations have been made at national and international meetings of the IAAI and the SFPE. Some of the results are presented in MS thesis by Allison Carey available at www.fpe.umd.edu.